Page 1

=>'FILE REG

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STRUCTURE FILE UPDATES: 24 SEP 2002 HIGHEST RN 454646-45-8 DICTIONARY FILE UPDATES: 24 SEP 2002 HIGHEST RN 454646-45-8

TSCA INFORMATION NOW CURRENT THROUGH MAY 20, 2002

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Crossover limits have been increased. See HELP CROSSOVER for details.

Experimental and calculated property data are now available. See HELP PROPERTIES for more information. See STNote 27, Searching Properties in the CAS Registry File, for complete details: http://www.cas.org/ONLINE/STN/STNOTES/stnotes27.pdf

=> FILE HCAPLUS

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FILE COVERS 1907 - 25 Sep 2002 VOL 137 ISS 13 FILE LAST UPDATED: 24 Sep 2002 (20020924/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

CAS roles have been modified effective December 16, 2001. Please check your SDI profiles to see if they need to be revised. For information on CAS roles, enter HELP ROLES at an arrow prompt or use the CAS Roles thesaurus (/RL field) in this file.

=> D QUE 936053 SEA FILE=REGISTRY ABB=ON PMS/CI L13 1 SEA FILE=REGISTRY ABB=ON "CRYSTAL VIOLET"/CN L16 12028 SEA FILE=REGISTRY ABB=ON L13 AND AMINIUM L17 L191 SEA FILE=REGISTRY ABB=ON "METHYLENE BLUE"/CN 341 SEA FILE=REGISTRY ABB=ON L13 AND PHENOTHIAZIN? L21 L24 SCR 2040 50774 SEA FILE=REGISTRY SUB=L13 SSS FUL L24 L26

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

in Registry

all polymers with a Charge

LEE 09/828075 Page 2 L28 36817 SEA FILE=HCAPLUS ABB=ON L26 L32 452 SEA FILE=HCAPLUS ABB=ON L28(L)(COLOR? OR DYE?)(L)?POLYMER? L34 325 SEA FILE=HCAPLUS ABB=ON L21 L40 13049 SEA FILE=HCAPLUS ABB=ON L28(L)(PREP OR IMF OR SPN)/RL L41 143 SEA FILE=HCAPLUS ABB=ON L32 AND L40 L42 3 SEA FILE=HCAPLUS ABB=ON L41 AND LITHOG? L43 13927 SEA FILE=HCAPLUS ABB=ON L17 L44 3671 SEA FILE=HCAPLUS ABB=ON L40 AND L43 L45 71 SEA FILE=HCAPLUS ABB=ON L32 AND L44 2 SEA FILE=HCAPLUS ABB=ON L45 AND LITHOG? L46 3 SEA FILE=REGISTRY ABB=ON "VICTORIA BLUE"/CN L47 "ETHYL VIOLET"/CN L48 1 SEA FILE=REGISTRY ABB=ON L49 4 SEA FILE=REGISTRY ABB=ON MONAZOLINE?/CN L50 1 SEA FILE=REGISTRY ABB=ON "1-ETHYL-2-METHYLQUINOLINIUM IODIDE"/CN L51 1 SEA FILE=REGISTRY ABB=ON "1-ETHYL-4-METHYLQUINOLINIUM IODIDE"/CN L52 1 SEA FILE=REGISTRY ABB=ON "CETYLPYRIDINIUM BROMIDE"/CN L53 1 SEA FILE=REGISTRY ABB=ON "ETHYLVFOLOGEN DIBROMIDE"/CN L54 78 SEA FILE=REGISTRY ABB=ON SOLVENT ORANGE 90 SEA FILE=REGISTRY ABB=ON L47 OR L48 OR L49 OR L50 OR L51 OR L61 L52 OR L53 OR L54 739 SEA FILE=HCAPLUS ABB=ON L61(L)RCT/RL L63 L64 1 SEA FILE=HCAPLUS ABB=ON L40 AND L63 L65 4 SEA FILE=HCAPLUS ABB=ON L28 AND L63 L69 817 SEA FILE=HCAPLUS ABB=ON (L16 OR L19) (L) RCT/RL 6 SEA FILE=HCAPLUS ABB=ON L40 AND L69 L70 L71 13 SEA FILE=HCAPLUS ABB=ON L64 OR L65 OR L65 OR L70 OR L42 OR L46 L72 139 SEA FILE=HCAPLUS ABB=ON L34(L) (PREP OR IMF OR SPN)/RL L73 O SEA FILE=HCAPLUS ABB=ON L72 AND LITHOG? L74 2 SEA FILE=HCAPLUS ABB=ON L72 AND PRINT? L75 4 SEA FILE=HCAPLUS ABB=ON L72 AND IMAG? 1.76 19 SEA FILE=HCAPLUS ABB=ON L71 OR L73 OR L74 OR L75 2235 SEA FILE=HCAPLUS ABB=ON L40 AND (QUAT?(2A)HETEROCYCL? OR L77 ?QUINOL? OR ?BENZOTHIAZOL? OR ?PYRIDIN? OR POLYMETHINE? OR ?CYANINE?) L78 21 SEA FILE=HCAPLUS ABB=ON L77 AND LITHOG? L79 39 SEA FILE=HCAPLUS ABB=ON L76 OR L78 L80 18 SEA FILE=HCAPLUS ABB=ON L77 AND PRINT? (3A) (PLATE? OR FORM?) O SEA FILE=HCAPLUS ABB=ON L77 AND COLOR?(3A) TAG? L81 L82 O SEA FILE=HCAPLUS ABB=ON L77 AND COLOR?(W) TAG? O SEA FILE=HCAPLUS ABB=ON L77 AND COLOR? (W) MARK? L83 O SEA FILE=HCAPLUS ABB=ON L77 AND COLOR? (3A) PENDENT? L84 46 SEA FILE=HCAPLUS ABB=ON (L79 OR L80 OR L81 OR L82 OR L83 OR L85 L84)

=> D L85 ALL 1-46 HITSTR

- L85 ANSWER 1 OF 46 HCAPLUS COPYRIGHT 2002 ACS AN 2002:396501 HCAPLUS
- DN 136:409051
- TI Development-free lithographic printing plate with good printability
- IN Hoshi, Satoshi; Kawamura, Koichi; Yamazaki, Sumiaki
- PA Fuji Photo Film Co., Ltd., Japan
- SO Jpn. Kokai Tokkyo Koho, 21 pp. CODEN: JKXXAF
- DT Patent

```
LEE 09/828075
                    Page 3
T.A
     Japanese
IC
    ICM B41N001-14
     ICS G03F007-00; G03F007-004
CC
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 38
FAN.CNT 1
    PATENT NO.
                      KIND DATE
                                           APPLICATION NO. DATE
                     ----
                                           -----
                            20020528
РΤ
    JP 2002154280
                      A2
                                           JP 2000-352924
                                                            20001120
    The plate has a hydrophilic image/recording layer on a support, wherein
ΑB
     the layer is changed to be hydrophobic by heat and comprises hydrophilic
    binder resins contg. precursor polymer particles having hydrophilic graft
    chains on their surface. A photothermal conversion agent may be contained
     in the recording layer and/or /ts neighboring layer. Image is formed on
    the plate by heat-mode light /rradn.
lithog printing plate hydrophilic image
ST
    recording layer; hydrophili polymer precursor heating hydrophobic
    printing plate
ΙT
    Lithographic plates
        (development-free lithog. printing plate
        with good printability)
TΤ
    Polyoxyalkylenes, preparation
    RL: PNU (Preparation, unclassified); TEM (Technical or engineered material
    use); PREP (Preparation); USES (Uses)
        (polystyrene-, grafft, image-recording layer component; development-free
        lithog. printing plate with good
        printability)
    109584-39-6P, Ethylene oxide-styrene graft copolymer 121287-81-8P,
    Methyl methacrylate/1-vinyl-2-pyrrolidinone graft copolymer
    135142-55-1P, Acry∤amide-methyl methacrylate graft copolymer
    146277-01-2P, Polyethylene glycol monomethacrylate-styrene graft copolymer
    163751-15-3P, tert-Butyl methacrylate-styrene graft copolymer
     429340-57-8P, Acfylacetamide-methyl methacrylate graft copolymer
    429340-63-6P, Styrene-4-vinyl-N-butylpyridinium bromide
    graft copolymen
    RL: PNU (Preparation, unclassified); TEM (Technical or engineered material
    use); PREP (Preparation); USES (Uses)
        (image-redording layer component; development-free lithog.
       printing plate with good printability)
ΙT
    429340-63-62, Styrene-4-vinyl-N-butylpyridinium bromide
    graft copo∤ymer
    RL: PNU (Preparation, unclassified); TEM (Technical or engineered material
    use); PRE/P (Preparation); USES (Uses)
        (image-recording layer component; development-free lithog.
       printing plate with good printability)
RN
     429340/63-6 HCAPLUS
CN
    Pyrid/nium, 1-butyl-4-ethenyl-, bromide, polymer with ethenylbenzene,
    graft (9CI) (CA INDEX NAME)
    CM
     CKN 1745-93-3
         C11 H16 N . Br
```

● Br⁻

CM 2

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

IT

```
L85 ANSWER 2 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     2002:396500 HCAPLUS
AN
DN
     136:409050
     Development-free lithographic printing plate
     with good printability
IN
     Hoshi, Satoshi; Kawamura, Koichi; Yama, Zaki, Sumiaki
PΑ
     Fuji Photo Film Co., Ltd., Japan
     Jpn. Kokai Tokkyo Koho, 17 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
     ICM B41N001-14
IC
     ICS G03F007-00; G03F007-004; GØ3F007-032; G03F007-11
CC
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s):
FAN.CNT 1
     PATENT NO.
                      KIND DATE
                                           APPLICATION NO.
                                                            DATE
                      ----
     JP 2002154279
                      A2
                            2,0020528
                                           JP 2000-352908
PΙ
                                                          20001120
     The plate has a hydroph lic image-recording layer on a support, wherein
     the layer is made of polymer particles having hydrophilic graft chains on
     their surface and chafiged to be hydrophobic by heat. A photothermal
     conversion agent may be contained in the recording layer and/or its
     neighboring layer. / Image is formed on the plate by heat-mode light
     irradn.
     lithog printing plate hydrophilic image
     recording layer; hydrophilic polymer precursor heating hydrophobic
     printing plate
```

with good printability)
Polyoxyalkylenes, preparation

(development-free lithog. printing plate

Lithographic plates

LEE 09/828075 Page 5

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(polystyrene-, graft, image-recording layer; development-free

lithog. printing plate with good

printability)

IT 109584-39-6P, Ethylene oxide-styrene graft copolymer 121287-81-8P,
Methyl methacrylate-1-vinyl-2-pyrrolidinone graft copolymer
135142-55-1P, Acrylamide-methyl methacrylate graft copolymer
146277-01-2P, Polyethylene glycol monomethacrylate-styrene graft copolymer
163751-15-3P, tert-Butyl methacrylate-styrene graft copolymer
429340-57-8P 429340-63-6P, Styrene-1-Butyl-4vinylpyridinium bromide graft copolymer
RL: PNU (Preparation, unclassified); TEM (Technical or engineered material

use); PREP (Preparation); USES (Uses)

(image-recording layer; development-free lithog.

printing plate with good printability)
IT 429340-63-6P, Styrene-1-Butyl-4-vinylpyridinium bromide

graft copolymer

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(image-recording layer; development-free lithog.

printing plate with good printability)

RN 429340-63-6 HCAPLUS

CN Pyridinium, 1-butyl-4-ethenyl-, bromide, polymer with ethenylbenzene,
 graft (9CI) (CA INDEX NAME)

CM 1

CRN 1745-93-3 CMF C11 H16 N . Br

● Br-

CM 2

CRN 100-42-5 CMF C8 H8

H2C CH-Ph

L85 ANSWER 3 OF 46 HCAPLUS COPYRIGHT 2002 ACS AN 2002:253298 HCAPLUS

```
LEE 09/828075
                      Page 6
     136:286628
TΙ
     Image-recording material and lithographic printing
     plate using the same
IN
     Shibuya, Akinori; Kunita, Kazuto; Oshima, Yasuhito
PA
     Fuji Photo Film Co., Ltd., Japan
SO
     Jpn. Kokai Tokkyo Koho, 67 pp.
     CODEN: JKXXAF
DΤ
     Patent
LA
     Japanese
IC
     ICM G03F007-11
     ICS G03F007-00; G03F007-027; G03F007-028/ G03F007-032; G03F007-09;
          G03F007-095
CC
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 35, 38, 41
FAN.CNT 1
     PATENT NO.
                      KIND DATE
                                           APPLICATION NO. DATE
                            _____
     JP 2002099094 A2 20020405
PΙ
                                           JP 2000-288940 20000922
     The invention relates to an image/recording material suitable for a CTP
AB
     system. The image-recording material comprises on a support a
     photopolymerizable layer contg./a photopolymn initiator, a sensitizing
     dye, an addn. polymerizable ethylenic compd., and a polymer binder and an
     intermediate layer contg. a photopolymn. initiator. The photopolymn.
     initiator in the intermediate layer is able to interact with the
     hydrophilic Al support. The lithog. printing
     plate using above image-forming material is also claimed.
ST
     computer to plate lithog printing
     plate
IT
    Lithographic plates
        (image-recording material for lithog. printing
       plate)
ΤT
     Polyesters, uses
     Polyurethanes, uses
     RL: TEM (Technical/or engineered material use); USES (Uses)
        (image-recording material for lithog. printing
        plate)
                 251/83-63-5
     3712-60-5
TΤ
                             246540-24-9
     RL: TEM (Technacal or engineered material use); USES (Uses)
        (IR dye; image-recording material for lithog.
       printing plate)
     106-96-7DP / Propargylbromide, reaction product with
IT
     poly((dimethylamino)ethyl methacrylate) 1592-20-7DP,
    4-(Chloromethyl) styrene, reaction products with poly(4-vinylpyridine) 2867-47-2DP, 2-(Dimethylamino) ethyl methacrylate,
     reaction product with poly(chloromethylstyrene) 9080-67-5DP,
     Poly (chloromethylstyrene), reaction product with 2-(dimethylamino)ethyl
     methacrylate 25154-86-3DP, Poly((dimethylamino)ethyl methacrylate),
     propargylbromide 25232-41-1DP, Poly(4-vinylpyridine), reaction
     products with 4-(chloromethyl)styrene 30674-80-7DP, reaction products
     with -vinylbenzoic acid-(4-vinylbenzyl)triethyl ammonium chloride
                 53160-51-3P, Acrylic acid-allyl methacrylate copolymer
     copolymer
     220227-02-1DP, reaction products with 2-
     (methacryloyoxy) ethylisocyanate 371971-09-4P, 4-Vinylbenzoic
     acid-(4-vinylbenzyl)triethyl ammonium chloride copolymer, ester with
     glycidyl methacrylate
     RL: SPN (Synthetic preparation); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (image-recording material for lithog. printing
        plate)
```

Page 7

IT 7429-90-5, Aluminum, uses

RL: DEV (Device component use); USES (Uses)

(support for lithog. printing plate)

IT 220227-02-1DP, reaction products with 2-

(methacryloyoxy)ethylisocyanate **371971-09-4P**, 4-Vinylbenzoic acid-(4-vinylbenzyl)triethyl ammonium chloride copolymer, ester with

glycidyl methacrylate

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(image-recording material for lithog. printing plate)

RN 220227-02-1 HCAPLUS

CN Benzenemethanaminium, 4-ethenyl-N,N,N-triethyl-, chloride, polymer with 4-ethenylbenzoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 14350-43-7 CMF C15 H24 N . C1

● C1-

CM 2

CRN 1075-49-6 CMF C9 H8 O2

RN 371971-09-4 HCAPLUS

CN Benzenemethanaminium, 4-ethenyl-N,N,N-triethyl-, chloride, polymer with 4-ethenylbenzoic acid, 2-hydroxy-3-[(2-methyl-1-oxo-2-propenyl)oxy]propyl ester (9CI) (CA INDEX NAME)

CM 1

CRN 5919-74-4

Page 8

CMF C7 H12 O4

$$\begin{array}{c|cccc} \text{OH} & \text{O} & \text{CH}_2 \\ | & || & || \\ \text{HO-CH}_2\text{-CH-CH}_2\text{-O-C-C-Me} \end{array}$$

CM 2

CRN 220227-02-1

CMF (C15 H24 N . C9 H8 O2 . C1) \times

CCI PMS

CM 3

CRN 14350-43-7 CMF C15 H24 N . Cl

● C1-

CM 4

CRN 1075-49-6 CMF C9 H8 O2

L85 ANSWER 4 OF 46 HCAPLUS COPYRIGHT 2002 ACS

AN 2001:814265 HCAPLUS

DN 135:350606

TI Negative-working lithographic printing plates with excellent interlayer adhesion and image reproducibility

IN Shimada, Kazuto; Oshima, Yasuhito; Kunida, Kazuto

PA Fuji Photo Film Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 27 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-11

```
LEE 09/828075
                      Page 9
     ICS B41N001-14; G03F007-00
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
FAN.CNT 1
     PATENT NO.
                      KIND DATE
                                           APPLICATION NO.
     JP 2001312068 A2 2001<u>11</u>09
                                           JP 2Ø00-132478
РΤ
                                                             20000501
     The plates for heat-mode recording, have (a) hydrophilized supports, (b)
     middle layers contg. radically polymerizable compds., and (C) IR laser-recordable photosensitive layers contg. IR absorbers, polymn.
     initiators, and compds. having polymerizable unsatd. groups, in this
     order. The radically polymerizable compds. may also have cationic groups.
     presensitized lithog printing plate heat
     mode; cationic polymer middle lafer interlayer adhesion; IR laser exposure
     neg lithog printing
ΙT
     Optical materials
        (IR absorbers, photosensitive layer contg.; presensitized
        lithog. printing plates having cationic
        polymer-contg. middle layers for IR laser exposure)
TΤ
     IR materials
        (absorbers, photosensitive layer contg.; presensitized lithog
        . printing plates having cationic polymer-contg.
        middle layers for AR laser exposure)
ΙT
    Lithographic plates,
       (neg.-working presensitized; presensitized lithog. printing plates having cationic polymer-contg. middle
        layers for IR laser exposure)
     134127-48-3 1/3783-73-8 244606-76-6
IT
     RL: DEV (Device component use); USES (Uses)
        (IR absorper, photosensitive layer contg.; presensitized lithog
        . printing plates having cationic polymer-contg.
        middle Mayers for IR laser exposure)
     106-96-7DF, Propargyl bromide, reaction products with
IT
     polydimerhylaminoethyl methacrylate 109-73-9DP, n-Butylamine, reaction
     2867-4/-2DP, 2-(Dimethylamino)ethyl methacrylate, reaction products with
     polyckloromethylstyrene and butylamine 9080-67-5DP,
     Poly(chloromethylstyrene), reaction products with dimethylaminoethyl
     metMacrylate and butylamine 25154-86-3DP, Poly[(dimethylamino)ethyl
     me/thacrylate], reaction products with propargyl bromide
     2-(Methacryloyloxy)ethyl isocyanate, reaction products with
     triethylvinylbenzylammonium chloride-vinylbenzoic acid copolymer
     120832-05-5P, Poly(4-vinylpyridine) salt with
     (chloromethyl) styrene 220227-02-1DP, Triethyl (p-
     vinylbenzyl)ammonium chloride-p-vinylbenzoic acid copolymer, reaction
     products with methacryloyloxyethyl isocyanate 371966-25-5P,
     Allyl methacrylate-methacrylic acid-[2-(methacryloyloxy)ethyl]trimethylamm
     onium methanesulfonate copolymer 371966-27-7P,
     N, N-Dimethylacetamide; (4-vinylbenzyl) triallylammonium hexafluorophosphate
     copolymer 371971-09-4P, Triethyl(4-vinylbenzyl)ammonium
     chloride-4-vinylbenzoic acid copolymer ester with glycidyl methacrylate
     RL: DEV (Device component use); PNU (Preparation, unclassified); PREP
     (Preparation); USES (Uses)
        (middle layer; presensitized lithog. printing
        plates having cationic polymer-contg. middle layers for IR
        laser exposure)
TΤ
     4986-89-4, Pentaerythritol tetraacrylate 139385-71-0, Glycerin
     dimethacrylate-hexamethylene diisocyanate copolymer
     RL: DEV (Device component use); USES (Uses)
        (photosensitive layer; presensitized lithog. printing
```

Page 10

plates having cationic polymer-contg. middle layers for IR laser exposure)

IT 37321-70-3D, JIS 1050, anodized

RL: DEV (Device component use); USES (Uses) (support; presensitized lithog. printing

plates having cationic polymer-contg. middle layers for IR laser exposure)

IT 220227-02-1DP, Triethyl(p-vinylbenzyl)ammonium
 chloride-p-vinylbenzoic acid copolymer, reaction products with
 methacryloyloxyethyl isocyanate 371966-25-5P, Allyl
 methacrylate-methacrylic acid-[2-(methacryloyloxy)ethyl]trimethylammonium
 methanesulfonate copolymer 371966-27-7P, N,N Dimethylacetamide;(4-vinylbenzyl)triallylammonium hexafluorophosphate
 copolymer 371971-09-4P, Triethyl(4-vinylbenzyl)ammonium
 chloride-4-vinylbenzoic acid copolymer ester with glycidyl methacrylate
 RL: DEV (Device component use); PNU (Preparation, unclassified); PREP
 (Preparation); USES (Uses)

(middle layer; presensitized lithog. printing
plates having cationic polymer-contg. middle layers for IR
laser exposure)

RN 220227-02-1 HCAPLUS

CN Benzenemethanaminium, 4-ethenyl-N,N,N-triethyl-, chloride, polymer with 4-ethenylbenzoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 14350-43-7 CMF C15 H24 N . C1

$$Et_3+N-CH_2$$
 $CH=CH_2$

• c1-

CM 2

CRN 1075-49-6 CMF C9 H8 O2

HO₂C CH CH₂

RN 371966-25-5 HCAPLUS

CN Ethanaminium, N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]-, methyl sulfate, polymer with 2-methyl-2-propenoic acid and 2-propenyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

Page 11

CM 1

CRN 96-05-9 CMF C7 H10 O2

CM 2

CRN 79-41-4 CMF C4 H6 O2

$$\begin{array}{c} \text{CH}_2 \\ || \\ \text{Me-C-CO}_2 \text{H} \end{array}$$

CM 3

CRN 6891-44-7

CMF C9 H18 N O2 . C H3 O4 S

CM 4

CRN 33611-56-2 CMF C9 H18 N O2

$$\begin{array}{c} \text{O} \quad \text{CH}_2 \\ || \quad || \\ \text{Me}_3 \text{+N} - \text{CH}_2 - \text{CH}_2 - \text{O} - \text{C} - \text{C} - \text{Me} \end{array}$$

CM 5

CRN 21228-90-0 CMF C H3 O4 S

Me-0-503-

RN 371966-27-7 HCAPLUS

CN Benzenemethanaminium, 4-ethenyl-N,N,N-tri-2-propenyl-, hexafluorophosphate(1-), homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 157424-95-8 CMF C18 H24 N

Page 12

$$H_2C = CH - CH_2$$
 $H_2C = CH - CH_2 - N + CH_2$
 $H_2C = CH - CH_2$
 $CH = CH_2$

CM 2

CRN 16919-18-9 CMF F6 P CCI CCS

RN 371971-09-4 HCAPLUS

ON Benzenemethanaminium, 4-ethenyl-N,N,N-triethyl-, chloride, polymer with 4-ethenylbenzoic acid, 2-hydroxy-3-[(2-methyl-1-oxo-2-propenyl)oxy]propyl ester (9CI) (CA INDEX NAME)

CM 1

CRN 5919-74-4 CMF C7 H12 O4

CM 2

CRN 220227-02-1

CMF (C15 H24 N . C9 H8 O2 . C1) x

CCI PMS

CM 3

CRN 14350-43-7 CMF C15 H24 N . C1

● C1-

CM 4

CRN 1075-49-6 CMF C9 H8 O2

L85 ANSWER 5 OF 46 HCAPLUS COPYRIGHT 2002/ACS

AN 2001:814264 HCAPLUS

DN 135:350565

TI Positive-working photoresist composition suitable for fabrication of magnetoresistive heads, copper or iron-nickel alloy substrate having photosensitive film, and its pattern formation

IN Masuda, Yasuo; Katano, Akira; Doi/, Kosuke; Ohara, Hidekatsu

PA Tokyo Ohka Kogyo Co., Ltd., Japah

SO Jpn. Kokai Tokkyo Koho, 13 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-085

ICS C08L061-14; G03F007-0/23

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)
Section cross-reference(s): 38, 77

FAN.CNT 1

ΡI

PATENT NO. KIND DATE APPLICATION NO. DATE

JP 2001312066 A2 20011109 JP 2000-132405 20000501

OS MARPAT 135:350565

The photoresist compn. contains an alkali-sol. novolak resin whose phenolic hydroxyl groups are partially esterified with 1,2-naphthoquinonediazide sulfonyl group, and an arom. N-contg. heterocycle. A Cu or Fe-Ni alloy substrate having a photosensitive film of .gtoreq.3.0 .mu.m thickness made from the compn. is claimed. The photosensitive film of .gtoreq.6.0 .mu.m thickness is image-wise patterned with i-line (365 nm) and developed to give .ltoreq.0.8 .mu.m line-and-space pattern. Addn. of the hetercycle remarkably improves adhesion of the photoresist to the substrate, so that the resist compn. is suitable for fabrication of giant magnetoresistive heads and

magnetoresistive heads.

ST novolak photoresist compn additive heterocycle adhesion improvement; arom heterocycle addn novolak photoresist adhesion improvement; giant magnetoresistive head fabrication novolak photoresist; magnetic recording head fabrication novolak photoresist; copper substrate novolak pos photoresist; iron nickel alloy substrate novolak pos photoresist; lithog pos novolak photoresist adhesion improvement

IT Heterocyclic compounds

RL: MOA (Modifier or additive use); USES (Uses)
 (arom.; pos.-working photoresist compn., and Cu or Fe-Ni alloy
 substrate having photoresist film suitable for fabrication of magnetic
 recording heads)

IT Aromatic compounds

RL: MOA (Modifier or additive use); USES (Uses) (heterocyclic; pos.-working photoresist compn., and Cu or Fe-Ni alloy substrate having photoresist film suitable for fabrication of magnetic recording heads)

IT Phenolic resins, preparation

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(novolak; pos.-working photoresist compn., and Cu or Fe-Ni alloy substrate having photoresist film suitable for fabrication of magnetic recording heads)

IT Lithography

Positive photoresists

(pos.-working photoresist compn., and Cu or Fe-Ni alloy substrate having photoresist film suitable for fabrication of magnetic recording heads)

IT 103-74-2, 2-(2-Hydroxyethyl)pyridine 29565-76-2, 2-

Hydroxyethylpyridine

RL: MOA (Modifier or additive use); USES (Uses) (pos.-working photoresist compn., and Cu or Fe-Ni alloy substrate having photoresist film suitable for fabrication of magnetic recording heads)

IT 202148-85-4P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(pos.-working photoresist compn., and Cu or Fe-Ni alloy substrate having photoresist film suitable for fabrication of magnetic recording heads)

IT 7440-50-8, Copper, uses 11148-32-6

RL: TEM (Technical or engineered material use); USES (Uses)
(substrate; pos.-working photoresist compn., and Cu or Fe-Ni alloy
substrate having photoresist film suitable for fabrication of magnetic
recording heads)

IT 202148-85-4P

RL: PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(pos.-working photoresist compn., and Cu or Fe-Ni alloy substrate having photoresist film suitable for fabrication of magnetic recording heads)

RN 202148-85-4 HCAPLUS

CN Formaldehyde, polymer with 3-methylphenol, 4-methylphenol and 2,3,5-trimethylphenol, 6-diazo-5,6-dihydro-5-oxo-1-naphthalenesulfonate (9CI) (CA INDEX NAME)

CM 1

CRN 20546-03-6 CMF C10 H6 N2 O4 S

CM 2

CRN 123236-78-2

CMF (C9 H12 O . C7 H8 O . C7 H8 O . C H2 O) \times

CCI PMS

CM 3

CRN 697-82-5 CMF C9 H12 O

CM

CRN 108-39-4 CMF C7 H8 O

5 CM

CRN 106-44-5 CMF C7 H8 O

CM 6

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$

```
L85 ANSWER 6 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     2001:705943 HCAPLUS
ΑN
ΤI
     Electrochemical copolymerization of aniline and azure B
     Shan, Dan; Mu, Shao-Lin; Mao, Bing-Wei; /Li, Yong-Fang
ΑU
CS
     Department of Chemistry, School of Science, Yangzhou University, Yangzhou,
     225002, Peop. Rep. China
SO
     Chinese Journal of Polymer Science (2001), 19(5), 483-492
     CODEN: CJPSEG; ISSN: 0256-7679
PB
     Springer-Verlag
DT
     Journal
LA
     English
CC
     35-7 (Chemistry of Synthetic High/Polymers)
     Section cross-reference(s): 41, /2, 73
     The electrochem. copolymn. of afiline and N,N,N'-trimethylthionin (azure
AB
     B) in aq. solns. has been carried out using the potential sweep method. The optimum conditions for the coelectrodeposition are that the pH value
     and the temp. of the electrolytic soln. are controlled at 5.57 and
     30.degree.C, resp., and the scan potential range is set between -0.25 and
     1.10 V (vs. SCE). The copolymn. rate of aniline and azure B is about 3
     times larger than that of aniline in the absence of azure B. The
     copolymn. of aniline and azure B was verified from the results of visible
     spectra during electrolysis, FTIR spectra and the at. force microscopy
     (AFM) images of the polymers. The in situ visible spectrum for the electrolysis of the soln. contg. aniline and azure B is different from
     that of the resp. aniline and azure B. The FTIR spectrum of the copolymer
     is not a superposition of that of polyaniline and poly(azure B). The AFM
     image of the copolymer is different from those of polyaniline and
     poly(azure B) and is not_a mixt. of individual polymers. The cond. of the
     copolymer synthesize at pH 5.57 is four orders of magnitude higher than
     that of polyaniline/synthesized under the same conditions, but in the
     absence of azure B./ The electrochem. properties of the copolymer are
     mainly attributed to polyaniline, but the copolymer has a better
     electrochem. reversibility and a much faster charge transfer than those of
     polyaniline.
ST
     aniline azure B dye electropolymn
ΙT
     pН
        (effect on a/hiline-azure B copolymer electrochem. prepn.)
ΙT
     Polyanilines
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
         (electrochem. prepn. and spectra of aniline-azure B copolymer)
IT
     Polymerization
        (electrochem.; electrochem. prepn. and spectra of aniline-azure B
        copolymer)
     Cyclic voltammetry
ΙT
        (in amiline-azure B copolymer prepn.)
ΙT
     Binding energy
     Electric conductivity
     UV and visible spectra
```

(of aniline-azure B copolymer)

IT 374777-76-1P, Aniline-azure B copolymer

> RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(electrochem. prepn. and spectra of)

RE.CNT THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD RE

- (1) Desilvestro, J; J Chem Soc Chem Commun 1985, P346
- (2) Dong, Y; Electrochim Acta 1991, V36, P2015 HCAPLUS
 (3) Fusalba, F; J Phys Chem B 1999, V103, P9044 HCAPLUS
- (4) Genies, E; J Appl Electrochem 1998, V18, P751
- (5) Gningue, D; J Electrochem Soc 1988, V135, P1695 HCAPLUS
- (6) Gospodinova, N; Prog Polym Sci 1998, V23, P1443 HCAPLUS
- (7) Huang, W; J Chem Soc Faraday Trans 1 1986, V82, P2385 HCAPLUS
- (8) Inganas, O; Synth Met 1985, V11, P239
- (9) Kan, J; Synth Met 1997, V87, P205 HCAPLUS
- (10) Kang, E; Surface and Interface Analysis 1993, V20, P833 HCAPLUS
- (11) Kim, H; Synth Met 1988, V26, P49
- (12) Kobayashi, T; J Electroanal Chem 1984, V177, P281 HCAPLUS
- (13) Kuwabata, S; J Electrochem Soc 1988, V135, P1691 HCAPLUS
- (14) Laborde, H; J Appl Electrochem 1990, V20, P524 HCAPLUS
- (15) MacDiarmid, A; Synth Met 1991, V41-43, P621
- (16) Pei, Q; Synth Met 1991, V45, P35 HCAPLUS
- (17) Ruckenstein, E; Synth Met 1993, V53, P283 HCAPLUS
- (18) San, B; Synth Met 1998, V94, P221
- (19) Sanchez, M; J Appl Electrochem 1997, V27, P831
- (20) Shan, D; Chinese Journal of Polymer Science in press
- (21) Shan, D; Electroanalysis in press
- (22) Tang, J; Synth Met 1988, V24, P231 HCAPLUS
- (23) Wan, X; J Electroanal Chem 1999, V470, P23 HCAPLUS
- (24) Wang, B; Synth Met 1986, V13, P329 HCAPLUS
- (25) Wang, H; Sensors and Actuators B 1999, V56, P22
- (26) Zotti, G; Chem Mater 1998, V10, P480 HCAPLUS
- 374777-76-1P, Aniline-azure B copolymer

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(electrochem. prepn. and spectra of)

- RN 374777-76-1 HCAPLUS
- Phenothiazin-5-ium, 3-(dimethylamino)-7-(methylamino)-, chloride, polymer CN with benzenamine (9CI) (CA INDEX NAME)

CM 1

CRN 531-55-5

C15 H16 N3 S . Cl CMF

Cl-

CRN 62-53-3 CMF C6 H7 N

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L85 ANSWER 7 OF 46 HCAPLUS COPYRIGHT 2002 ACS
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AN 2001:589711 HCAPLUS

DN 135:173028

TI Thermosensitive ionomer compositions containing oxonol IR dyes, their image-forming members, and their image **formation** and **printing** processes

IN Dominh, Thap; Zheng, Shiying; Williams, Kevin W.

PA Eastman Kodak Co., USA

SO Jpn. Kokai Tokkyo Koho, 39 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM B41N001-14

ICS C08K005-315; C08L101-00;/C09K003-00; G03F007-004; C09B023-00

CC 74-7 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

CHM.	CNII		/		
	PATENT NO.	KIND	PATE	APPLICATION NO.	DATE
		,	/		
ΡI	JP 2001219667	A2 /	20010814	JP 2000-355491	20001122
	US 6423469	В1/	20020723	US 1999-444695	19991122
	DE 10053721 \	ΑΛ	20010830	DE 2000-10053721	20001030
	GB 2358710 \	A 1	20010801	GB 2000-27723	20001114
PRAI	US 1999-444695	$\vee_{\mathtt{A}}$	19991122		

OS MARPAT 135:173028

- The compns. contain (a) hydrophilic thermosensitive ionomers, (b) water or water-miscible org. solvents, and (c) oxonol-type IR dyes sol. in water or the solvents and having lambda.max longer than 700 nm. The image-forming members having high photothermal conversion efficiency involve supports having hydrophilic image forming layers of the compns. The printing process involves (i) image formation involving a step of prepg. the image-forming members and a step of imagewise exposure of the members to provide exposed and unexposed regions in the image-forming layers, the heat provided by the exposure making the exposed regions become more hydrophilic than the unexposed regions, and (ii) a step of contacting the imagewisely exposed image-forming members with lithog.

 printing inks and transferring the inks from the members to
- receptor materials imagewisely.

 ST thermosensitive ionomer oxonol IR dye printing; thiosulfate polymer thermosensitive printing oxonol IR dye; photothermal conversion thermosensitive ionomer oxonol dye

IT Dves

(IR-absorbing, oxonols; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members)

IT Ionene polymers

RL: DEV (Device component use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)

(hydrophilic, thermosensitive; thermosensitive ionomer compns. contg.

LEE 09/828075 Page 19 oxonol IR dyes for printing members) TΤ Thermographic copying (thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) Polyoxyphenylenes RL: DEV (Device component use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses) (thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) ΙT 129587-84-4P RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (intermediate for monomers; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) IT 149976-02-3P RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (intermediate in dye prepn.; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) 31134-43-7P, Vinylbenzyl bromide IT RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (intermediates in monomer prepn.; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) IT 66-27-3, Methyl methanesulfonate 1072-63-5, 1-Vinylimidazole RL: RCT (Reactant); RACT (Reactant or reagent) (monomer prepn. from; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) IT 7268-68-0P 30030-25-2DP, thiosulfate sodium salt 264255-37-0P 354578-44-2P 354584-57-9P RL: PNU (Preparation, unclassified); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent) (monomers; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) ΙT 39198-78-2 RL: RCT (Reactant); RACT (Reactant or reagent) (reactants in dye prepn.; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) ΙT 77-78-1, Dimethyl sulfate RL: RCT (Reactant); RACT (Reactant or reagent) (reactants in monomer prepn.; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) ΙT 17576-35-1 101685-29-4 RL: RCT (Reactant); RACT (Reactant or reagent) (starting material in dye prepn.; thermosensitive ionomer compns. contq. oxonol IR dyes for printing members) 14216-23-0, 2-(Methylthio)ethyl TΤ 5188-07-8, Sodium methanethiolate methacrylate RL: RCT (Reactant); RACT (Reactant or reagent) (starting materials for monomers; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) 30030-25-2 RL: RCT (Reactant); RACT (Reactant or reagent) (starting materials in monomer prepn.; thermosensitive ionomer compns. contg. oxonol IR dyes for printing members) 344911-67-7 344911-71-3 354577-84-7 354577-86-9 344908-19-6 354577-88-1 354577-90-5 354577-92-7 354577-94-9 354577-96-1 354578-04-4 354578-06-6 354577-98-3 354578-00-0 354578-02-2 354578-08-8 354578-10-2 354578-11-3 354578-13-5 354578-14-6 354578-23-7 354578-15-7 354578-17-9 354578-19-1 354578-21-5

Page 20

Page 21

● c1-

RN 263551-57-1 HCAPLUS

CN Sulfonium, [(ethenylphenyl)methyl]dimethyl-, methyl sulfate, homopolymer (9CI) (CA INDEX NAME)

.CM 1

CRN 31094-38-9 CMF C11 H15 S CCI IDS

 $D1-CH = CH_2$

 $\begin{array}{c} \text{Me} \\ | \\ \text{Me} - \text{S} \xrightarrow{+} \text{CH}_2 - \text{D1} \end{array}$

CM 2

CRN 21228-90-0 CMF C H3 O4 S

Me-0-SO3-

RN 264255-38-1 HCAPLUS CN 1H-Imidazolium, 1-ethenyl-3-methyl-, methanesulfonate, polymer with

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Page 22

N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

$$^{\rm H2C}_{||}$$
 O $_{||}$ || Me-C-C-NH-(CH2)3-NH2

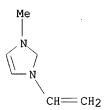
● HCl

CM 2

CRN 264255-37-0 CMF C6 H9 N2 . C H3 O3 S

CM 3

CRN 45534-45-0 CMF C6 H9 N2



*** FRAGMENT DIAGRAM IS INCOMPLETE ***

CM 4

CRN 16053-58-0 CMF C H3 O3 S

RN 304023-76-5 HCAPLUS

Sulfonium, dimethyl[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl]-, methyl sulfate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

Page 23

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

● HCl

CM 2

CRN 7268-68-0

CMF C8 H15 O2 S . C H3 O4 S

CM 3

CRN 44992-92-9 CMF C8 H15 O2 S

CM 4

CRN 21228-90-0 CMF C H3 O4 S

Me-0-S03-

RN 306767-46-4 HCAPLUS

CN Benzenemethanaminium, ar-ethenyl-N,N,N-trimethyl-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . Cl H

Page 24

HCl

CM 2

CRN 26616-35-3 CMF C12 H18 N . C1 CCI IDS



 $D1-CH=CH_2$

 Me_3+N-CH_2-D1

● C1-

RN 354578-39-5 HCAPLUS

CN 1-Propanaminium, N,N,N-trimethyl-3-[(2-methyl-1-oxo-2-propenyl)oxy]-,
chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide
monohydrochloride and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

$$^{\text{H}_2\text{C}}_{||}$$
 O $^{||}_{||}$ $||_{\text{Me}^-\text{C}^-\text{C}^-\text{NH}^-\text{(CH}_2)}\,3^-\text{NH}_2^-$

HCl

CM 2

Page 25

CRN 55918-38-2 CMF C10 H20 N O2 . C1

$$\begin{array}{c} \text{O} \quad \text{CH}_2 \\ || \quad || \\ \text{Me}_3 \text{+N- (CH}_2)_3 \text{-O-C-C-Me} \end{array}$$

● cl-

CM 3

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ & \parallel & \parallel \\ \text{Me-} & \text{C-} & \text{C-} & \text{OMe} \end{array}$$

RN 354578-39-5 HCAPLUS

CN 1-Propanaminium, N,N,N-trimethyl-3-[(2-methyl-1-oxo-2-propenyl)oxy]-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . Cl H

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ & || & || \\ ^{Me-C-C-NH-(CH_2)} \, _{3}^{-NH_2} \end{array}$$

● HCl

CM 2

CRN 55918-38-2 CMF C10 H20 N O2 . C1

Page 26

$$\begin{array}{c} \text{O} \quad \text{CH}_2 \\ || \quad || \\ \text{Me}_3 + \text{N} - \text{(CH}_2)_3 - \text{O} - \text{C} - \text{C} - \text{Me} \end{array}$$

● Cl-

· CM 3

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{c|c} ^{\text{H}_2\text{C}} & \text{O} \\ \parallel & \parallel \\ \text{Me-C-C-OMe} \end{array}$$

RN 354578-40-8 HCAPLUS '

CN 1-Propanaminium, N,N,N-trimethyl-3-[(2-methyl-1-oxo-2-propenyl)oxy]-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride and methyl 2-methyl-2-propenoate, acetate (9CI) (CA INDEX NAME)

CM 1

CRN 64-19-7 CMF C2 H4 O2

CM 2

CRN 354578-39-5 CMF (C10 H20 N O2 . C7 H14 N2 O . C5 H8 O2 . C1 H . C1)x CCI PMS

CM 3

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

● HCl

CM 4

CRN 55918-38-2 CMF C10 H20 N O2 . C1

• c1-

CM 5

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{c|c} ^{H_2C} & \text{O} & \cdot \\ \parallel & \parallel \\ \text{Me-} & \text{C-} & \text{C-} & \text{OMe} \end{array}$$

RN 354584-60-4 HCAPLUS

CN Phosphonium, [(ethenylphenyl)methyl]trimethyl-, bromide, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride, acetate (9CI) (CA INDEX NAME)

CM 1

CRN 64-19-7 CMF C2 H4 O2

CM 2

CRN 312966-33-9

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CMF (C12 H18 P . C7 H14 N2 O . Br . C1 H)x CCI PMS

CM 3

CRN 312965-31-4 CMF C12 H18 P . Br CCI IDS



 $D1-CH = CH_2$

Me3+P-CH2-D1

• Br-

CM 4

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

● HCl

L85 ANSWER 8 OF 46 HCAPLUS COPYRIGHT 2002 ACS

AN 2001:356528 HCAPLUS

DN 134:374064

TI Light-sensitive resin composition containing polyvinyl acetate dispersion medium and crosslinking dispersoid and method for production of rotary screen printing original plate using same

IN Kaneda, Sadayoshi; Yoshizawa, Kaizan; Kanetani, Yoshiyuki

PA Murakami Co., Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 19 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-038

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LEE 09/828075
                       Page 29
     ICS B05D001-26; B05D007-24; C08F002-44; Q08F002-50; C08G018-00;
          C08K005-29; C08L029-04; C08L079-00; C08L101-00; G03F007-004; G03F007-021; G03F007-12; G03F007-40
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
FAN.CNT 1
                                            APPLICATION NO. DATE
     PATENT NO.
                       KIND DATE
     JP 2001133976 A2 20010518 JP 1999-315468 19991105 The title resin compn. contains a water-sol. sapond. polyvinyl acetate
PΤ
     dispersion medium and a water hatdly sol. or insol. heat-crosslinking
     dispersoid, wherein the polyvin/yl acetate dispersion medium contains a
     styrylpyridinium or styryl quinolinium group and is
     .gtoreq.50 mol% sapond. The resin compn., which contains the polyvinyl
     acetate, provides the improved water- and org. solvent resistance and the
     high printing durability.
     light sensitive resin comp/ dispersion medium crosslinking dispersoid
     rotary; screen printing plate compn dispersion medium
     crosslinking dispersoid fotary
IT
     Light-sensitive materia 1/2s
        (light-sensitive resin compn. contq. polyvinyl acetate and method for
        prodn. of screen printing original plate using
        same)
ΙT
     Printing plates
        (screen; light-sensitive resin compn. contg. polyvinyl acetate and
        method for prody. of screen printing original plate
        using same)
     9003-20-7D, Poly vinyl acetate), partially sapond. RL: MSC (Miscellaneous)
        (dispersion medium in light-sensitive resin compn.)
     9003-20-7DP, Poly(vinyl acetate), partially sapond., reaction product with arylquinolium/9070-36-4P 78521-11-6DP,
IT
     (4-Formylsty y1)-N-methylquinolium, salt with dimethylsulfate,
     adduct with partially sapond. vinyl acetate polymer
     RL: SPN (Synthetic preparation); TEM (Technical or engineered
     material vse); PREP (Preparation); USES (Uses)
        (dispersion medium in light-sensitive resin compn.)
     77-78-1D/ Dimethylsulfate, salt with pyridinium deriv.
ΙT
     73264-13-8D, (4-Formylstyryl)-N-methylpyridinium, salt with
     dimeth#lsulfate
     RL: TEM (Technical or engineered material use); USES (Uses)
        (dispersion medium in light-sensitive resin compn.)
ΤT
     907Ø-36-4P
     RL/ SPN (Synthetic preparation); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (dispersion medium in light-sensitive resin compn.)
     9070-36-4 HCAPLUS
RN
     Benzenediazonium, 4-(phenylamino)-, sulfate (2:1), polymer with
     formaldehyde (9CI) (CA INDEX NAME)
     CM
          1
     CRN 50-00-0
     CMF C H2 O
```

 $H_2C = O$

Page 30

CM 2

CRN 150-33-4

CMF C12 H10 N3 . 1/2 O4 S

CM 3

CRN 16072-57-4 CMF C12 H10 N3

CM 4

CRN 14808-79-8 CMF O4 S

L85 ANSWER 9 OF 46 HCAPLUS COPYRIGHT 2002 ACS

AN 2001:231943 HCAPLUS

DN 135:11484

TI Growth of poly(methylene blue) film and proper fies of the film

AU Liu, Jin-cui; Mu, Shao-lin; Li, Yong-fang

CS Department of Chemistry, School of Sciences/Yangzhou University, Shouxi Lake Campus, Yangzhou, 225002, Peop. Rep. Phina

SO Wuli Huaxue Xuebao (2001), 17(3), 229-233 CODEN: WHXUEU; ISSN: 1000-6818

PB Beijing Daxue Chubanshe

DT Journal

LA Chinese

CC 72-2 (Electrochemistry)

Section cross-reference(s): 35, 36, 41

The in situ spectroelectrochem. expt. shows that an absorption peak at 580.5 nm attributable to poly(methylene blue) grows slowly at the beginning of the electrolysis of methylene blue, and then this peak grows more quickly. The visible spectra prove that there are two absorption peaks at 580.5 and 690.2 nm resp. for the mixt. soln. after electrolysis. The former one indicates that poly(methylene blue) polymd. on the ITO electrode is partly sol. the latter one is attributed to a water-sol. polymer since it completely dissolves in the aq. soln. The result from the rotating ring-disk electrode identifies that an intermediate is produced at the disk electrode, which is collected at the ring electrode. The ring current increases with decreasing the ring potential and with increasing the rotation rate. Thus, the intermediate is a species carrying a pos. charge and not too stable. The result from the XPS expt.

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LEE 09/828075
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Page 31

shows that the anions can be doped into poly (methylene blue) film and dedoped from it during oxidn. and redn. processes. The results from the impedance expt. indicate that the charge transfer impedance of poly (methylene blue) decreases with decreasing potential.

ST polymethylene blue in situ spectroelectrochem; rotating ring disk electrode polymethylene blue; impedance polymethylene blue

IT Polymerization Redox reaction

(electrochem.; growth and properties of poly(methylene blue) film)

IT Electrolysis

IT

Electron transfer

(growth and properties of poly(methylene blue) film)

IT 61-73-4, Methylene blue
RL: PEP (Physical, engineering or chemical process); PRP (Properties);
RCT (Reactant); PROC (Process); RACT (Reactant or reagent)

(growth and properties of poly(methylene blue) film)
150645-86-6P, Poly(methylene blue)
RL: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation)

RN 61-73-4 HCAPLUS

CN Phenothiazin-5-ium, 3,7-bis(dimethylamino)-, chloride (9CI) (CA INDEX NAME)

● Cl -

IT 150645-86-6P, Poly(methylene blue)

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(growth and properties of poly(methylene blue) film)

RN 150645-86-6 HCAPLUS

CN Phenothiazin-5-ium, 3,7-bis(dimethylamino)-, chloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 61-73-4

CMF C16 H18 N3 S . Cl

● C1 -

```
L85 ANSWER 10 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     2001:165716 HCAPLUS
ΑN
DN
     134:214934
     Thermal imaging composition as well as direct write lithographic
     printing plate containing cationic IR dye, and method of
     imaging and printing
IN
     Felming, James C.; Leon, Jeffrey W.; Stegman, David A.; Williams, Kevin W.
PA
     Eastman Kodak Co., USA
SO
     Ger. Offen., 26 pp.
     CODEN: GWXXBX
DТ
     Patent
LA
     German
     ICM G03F007-09
IC
     ICS B41C001-10; B41F007-02; B41M001-14; B41C001-00; B41C001-055
CC
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 38
FAN.CNT 1
                          KIND DATE
     PATENT NO.
                                                  APPLICATION NO. DATE
     DE 10042293
                          A1
                                 20010/308
                                                  DE 2000-10042293 20000829
                                 2002/0625
     US 6410202
                           В1
                                                  US 1999-387116
                                                                      19990831
     JP 2001162965
                                 200/10619
                                                  JP 2000-306855
                          A2
                                                                      20000831
PRAI US 1999-387116
                                 19990831
                          Α
os
     MARPAT 134:214934
     The invention relates to the neg.-working lithog.
     printing plate or cylinder in which the hydrophilic
      imaging layer is made up of heat-sensitive hydrophilic (ionomer) and
     IR-sensitive dye with several quaternary ammonium groups. Heat is generated by IR laser irradn. The heat-sensitive polymer is considered "switchable" in response to heat, and provides the lithog. image without wet processing. The IR dyes and the heat-sensitive hydrophilic
     ionomers were-synthesized.
     thermal imaging compn direct write lithog printing
     plate; cationic IR/dye heat sensitive hydrophilic ionomer
     printing plate
ΙT
     Imaging
         (IR; thermal maging compn. as well as direct write lithog. printing plate contg. cationic IR dye, and method of
         imaging and printing)
ΙT
      Ionomers
      Polyethers, pfeparation
      RL: SPN (Synthetic preparation); TEM (Technical or engineered material
      use); PREP (Preparation); USES (Uses)
         (prepn. of heat-sensitive hydrophilic ionomer for direct write
         lithog. printing plate)
ΙT
     Lithographic plates
```

```
(thermal imaging compn. as well as direct write lithog.
       printing plate contg. cationic IR dye, and method of
       imaging and printing)
IT
    329039-81-8P
                   329039-82-9P
                                   329039-85-2P
    RL: SPN (Synthetic preparation); TEM (Technical or engineered material
    use); PREP (Preparation); USES (Uses)
        (cationic IR-sensitive dye in direct write lithog.
       printing plate)
    312963-46-5P
    RL: SPN (Synthetic preparation); TEM (Technical or engineered
    material use); PREP (Preparation); USES (Uses)
        (heat-sensitive hydrophilic ionomer in direct write lithog.
       printing plate)
    1493-13-6, Trifluoromethanesulfonic acid
                                               1643-19-2,
    Tetrabutylammoniumbromide 3779-42-8 41532-84-7, 1,1,2-Trimethyl-1H-
                    63857-00-1
    benz[e]indole
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (prepn. of cationic IR-sensitive dye for direct write lithog.
       printing plate)
    329039-79-4P
IT
    RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
     (Reactant or reagent)
        (prepn. of cationic IR-sensitive dye for direct write lithog.
       printing plate)
    66-27-3, Methylmethanesulfonate 77-78-1, Dimethylsulfate
TΤ
                                                                 594-09-2,
    Trimethylphosphine 1072-63-5, 1-Vinylimidazole 5188-07-8, Sodium
                     7647-15-6, Sodium bromide, reactions
    methanethiolate
                                                            14216-23-0,
    2-(Methylthio)ethylmethacrylate 57458-41-0
    RL: RCT (Reactant); RACT (Reactant or reagent)
        (prepn. of heat-sensitive hydrophilic ionomer for direct write
       lithog. printing plate)
    7268-68-0P
                 25067-32-7P, Methyl methacrylate-2-vinylpyridine
    copolymer 26100-41-4P, Methyl methacrylate-4-vinylpyridine
    copolymer 264255-37-0P 304023-71-0P 304023-76-5P
    312965-31-4P 312966-33-9P
                                312966-34-0P
                                              312966-37-3P
    312966-38-4P
    RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (prepn. of heat-sensitive hydrophilic ionomer for direct write
       lithog. printing plate)
    24938-67-8DP, Poly(2,6-dimethyl-1,4-phenylene oxide), reaction products
    with N-bromosuccinimide and dimethylsulfide 24969-06-0DP,
    Epichlorohydrin homopolymer, reaction products with sodium thiosulfate
    25067-32-7DP, reaction products with Me tolylsulfonate, ion exchange from
    tolylsulfonate to formate 25134-01-4DP, Poly(2,6-dimethyl-1,4-phenylene
    oxide), reaction products with N-bromosuccinimide and dimethylsulfide
    26100-41-4DP, reaction products with Me tolylsulfonate or Bu bromide, ion
    exchange from tolylsulfonate or bromide to formate 51024-16-9DP,
    Vinylbenzyl chloride-styrene copolymer, reaction products with sodium
                 60098-22-8DP, Vinylbenzyl chloride-methyl methacrylate
    thiosulfate
    copolymer, reaction products with sodium thiosulfate
                                                           61710-61-0DP,
    Epichlorohydrin homopolymer, SRU, reaction products with sodium
    thiosulfate 115708-89-9P, Poly(p-xylidene-tetrahydrothiophenium
    chloride) 264255-38-1P, 1-Vinyl-3-methylimidazolium
    methanesulfonate-N-(3-aminopropyl)methacrylamide hydrochloride copolymer
    304023-71-0DP, ion exchange from chloride to acetate or fluoride
    304023-76-5DP, ion exchange from Me sulfate to chloride
    312966-33-9DP, ion exchange from bromide to acetate
    312966-36-2DP, ion exchange from Me sulfate to chloride
    312966-36-2P
```

IT

IT

RN

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 (prepn. of heat-sensitive hydrophilic ionomer for direct write lithog. printing plate)
25212-74-2P, Poly(thio-1,4-phenylene)
RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
 (reaction products with methanesulfonic acid and methyltriflate, ion exchange from triflate to chloride; prepn. of heat-sensitive hydrophilic ionomer for direct write lithog. printing plate)
312963-46-5P
RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)
(heat-sensitive hydrophilic ionomer in direct write lithog. printing plate)

312963-46-5 HCAPLUS

CN Benzenemethanaminium, 3(or 4)-ethenyl-N,N,N-trimethyl-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 74311-76-5 CMF C12 H18 N . C1 CCI IDS



D1-CH=CH2

 Me_3+N-CH_2-D1

• c1-

CM 2

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

Page 35

● HCl

304023-71-0P 304023-76-5P 312966-33-9P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP

(Preparation); RACT (Reactant or reagent)

(prepn. of heat-sensitive hydrophilic ionomer for direct write

lithog. printing plate)

304023-71-0 HCAPLUS RN

Ethanaminium, N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]-, CN chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM1

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

$$^{\rm H_2C}_{||\cdot||}$$
 O $^{\rm Me-C-C-NH-(CH_2)_{3}-NH_2}$

● HCl

CM 2

CRN 5039-78-1 CMF C9 H18 N O2 . C1

● cı-

CM 3

CRN 80-62-6 CMF C5 H8 O2

Page 36

RN 304023-76-5 HCAPLUS

CN Sulfonium, dimethyl[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl]-, methyl sulfate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

● HCl

CM 2

CRN 7268-68-0 CMF C8 H15 O2 S . C H3 O4 S

CM 3

CRN 44992-92-9 CMF C8 H15 O2 S

CM 4

CRN 21228-90-0 CMF C H3 O4 S

Me-0-503-

RN 312966-33-9 HCAPLUS

CN Phosphonium, [[3(or 4)-ethenylphenyl]methyl]trimethyl-, bromide, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

Page 37

CRN 312965-31-4 CMF C12 H18 P . Br CCI IDS



D1-CH-CH2

Me3+P-CH2-D1

● Br-

2 CM

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

● HCl

115708-89-9P, Poly(p-xylidene-tetrahydrothiophenium chloride) 264255-38-1P, 1-Vinyl-3-methylimidazolium methanesulfonate-N-(3-, ion exchange from chloride to acetate or fluoride 304023-76-5DP , ion exchange from Me sulfate to chloride 312966-33-9DP, ion exchange from bromide to acetate 312966-36-2DP, ion exchange from Me sulfate to chloride 312966-36-2P RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (prepn. of heat-sensitive hydrophilic ionomer for direct write lithog. printing plate) 115708-89-9 HCAPLUS RN CN

Poly[1,4-phenylene[1-(tetrahydrothiophenio)-1,2-ethanediyl] chloride] (9CI) (CA INDEX NAME)

Page 38

● C1-

RN 264255-38-1 HCAPLUS

CN 1H-Imidazolium, 1-ethenyl-3-methyl-, methanesulfonate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . Cl H

● HCl

CM 2

CRN 264255-37-0 CMF C6 H9 N2 . C H3 O3 S

CM 3

CRN 45534-45-0 CMF C6 H9 N2

Page 39

*** FRAGMENT DIAGRAM IS INCOMPLETE ***

CM 4

CRN 16053-58-0 CMF C H3 O3 S

RN 304023-71-0 HCAPLUS

CN Ethanaminium, N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . Cl H

● HCl

CM 2

CRN 5039-78-1 CMF C9 H18 N O2 . C1

● c1-

CM 3

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} \text{C-} \text{C-} \text{OMe} \end{array}$$

RN 304023-76-5 HCAPLUS
CN Sulfonium, dimethyl[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl]-, methyl sulfate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . Cl H

$$^{\rm H_2C}_{||}$$
 $^{\rm O}_{||}$ $^{\rm Me-C-C-NH-}_{\rm (CH_2)\,3-NH_2}$

● HCl

CM 2

CRN 7268-68-0 CMF C8 H15 O2 S . C H3 O4 S

CM 3

CRN 44992-92-9 CMF C8 H15 O2 S

Page 41

CM 4

CRN 21228-90-0 CMF C H3 O4 S

Me- 0- S03-

RN 312966-33-9 HCAPLUS

CN Phosphonium, [[3(or 4)-ethenylphenyl]methyl]trimethyl-, bromide, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 312965-31-4 CMF C12 H18 P . Br

CCI IDS



D1-CH=CH2

 Me_3+P-CH_2-D1

● Br-

CM 2

CRN 72607-53-5 CMF C7 H14 N2 O . Cl H

 $\begin{array}{c|c} ^{H_2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} & \text{C-} & \text{C-} & \text{NH-} & \text{(CH}_2)_3 - \text{NH}_2 \end{array}$

HC1

RN 312966-36-2 HCAPLUS CN Sulfonium, [[2(or 4)-ethenylphenyl]methyl]-dimethyl-, methyl sulfate,

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

Page 42

homopolymer (9CI) (CA INDEX NAME)

CRN 312966-35-1 CMF C11 H15 S CCI IDS



$$D1-CH=CH_2$$

$$\begin{array}{c} \text{Me} \\ | \\ \text{Me-S} \xrightarrow{+} \text{CH}_2 - \text{D1} \end{array}$$

CM 2

CRN 21228-90-0 CMF C H3 O4 S

Me-0-SO3-

RN 312966-36-2 HCAPLUS

CN Sulfonium, [[2(or 4)-ethenylphenyl]methyl]-dimethyl-, methyl sulfate, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 312966-35-1 CMF C11 H15 S CCI IDS



$$D1-CH=CH_2$$

CM 2

CRN 21228-90-0 CMF C H3 O4 S

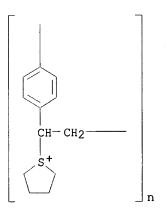
Me-0-503-

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L85 ANSWER 11 OF 46 HCAPLUS COPYRIGHT 2002 ACS
ΑN
     2001:129894 HCAPLUS
DN
     134:185985
TΙ
     Processless direct write printing plate having heat
     sensitive positively-charged polymers and methods of imaging and printing
     Leon, Jeffrey W.; Underwood, Gary M.; Fleming, James C.; Deboer, Charles
IN
PA
     Kodak Polychrome Graphics LLC, USA
SO
     U.S., 12 pp., Cont.-in-part of U. S. Ser./163,020, abandoned.
     CODEN: USXXAM
DT
     Patent
LA
     English
     ICM G03F007-004
IC
NCL
     430270100
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 35, 38
FAN.CNT 2
     PATENT NO.
                       KIND DATE
                                             APPLICATION NO. DATE
                                              _____
                             20010220
                                             US 1999-310038
PΙ
     US 6190831_
                        В1
                                                               19990511
                             20000421/
                                             JP 1999-276928
     JP 2000112123
                        A2
                                                               19990929
PRAI US 1998-163020
                       B2
                             1998092/9
                             199905/11
     US 1999-310038
                       Α
     An imaging member, such as a/neq.-working printing plate
     , can be prepd. using a hydrophilic imaging layer comprised of a
     heat-sensitive hydrophilic polymer having a pos. charged moiety, and
     optionally a photothermal conversion material. The heat-sensitive polymer
     has recurring units contg. an N-alkylated arom. heterocyclic group or an
     organoonium group that reacts to provide increased oleophilicity in areas
     exposed to energy that provides or generates heat. For example, heat can
     be supplied by laser irradn. in the IR region of the electromagnetic
     spectrum. Thus, the heat-sensitive polymer is considered "switchable" in response to heat, and provides an imaging means without wet processing. lithog printing plate heat sensitive polymer
ST
     prepn method; thermally switchable polymer digital imaging direct write
     printing plate
IT
     Polyoxyphenylenes
     Polythiophenylene/s
     RL: PRP (Propertdes); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); (ACT (Reactant or reagent)
(prepn. of hydrophilic heat-sensitive imaging polymers for direct write
        printing plate for imaging without wet processing)
     Carbon black, uses
TΤ
     RL: TEM (Teckinical or engineered material use); USES (Uses)
        processing using photothermal conversion material as)
ΙT
     Lithographic plates
        (processless direct write printing plate having
```

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heat sensitive pos.-charged polymers and methods of imaging and
        printing)
ΙT
     Polyesters, uses
     RL: TEM (Technical or engineered material use); USES (Uses)
        (support; prepn. of printing plates for imaging
        without wet processing using polyester support)
TΤ
     115970-62-2 163120-91-0
                                297173-98-9
     RL: TEM (Technical or engineered material use); USES (Uses)
        (IR dye; prepn. of printing plates for imaging
        without wet processing using IR dye as photothermal conversion
        material)
ΙT
     75-75-2DP, Methanesulfonic acid, reaction products with poly(phenylene
                333-27-7DP, Methyl triflate, reaction products with
     sulfide)
     poly(phenylene sulfide)
                              25212-74-2DP, Poly(thio-1,4-phenylene), reaction
     products with methanesulfonic acid and Me triflate 325975-91-5P
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); SPN
     (Synthetic preparation); TEM (Technical or engineered material use); PREP
     (Preparation); PROC (Process); USES (Uses)
        (prepn. of hydrophilic heat-sensitive imaging polymers for direct write
        printing plate for imaging without wet processing)
     115708-89-9P, Poly(p-xylidenetetrahydro-thiophenium chloride)
ΙT
                    325975-92-6P
                                  325975-93-7P
     RL: PEP (Physical, engineering or chemical process); SPN (Synthetic
     preparation); TEM (Technical or engineered material use); PREP
     (Preparation); PROC (Process); USES (Uses)
        (prepn. of hydrophilic heat-sensitive imaging polymers for direct write
        printing plate for imaging without wet processing)
ΙT
     264255-37-0P
     RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (prepn. of hydrophilic heat-sensitive imaging polymers for direct write
        printing plate for imaging without wet processing)
ΙT
     24938-67-8DP, Poly[oxy(2,6-dimethyl-1,4-phenylene)], brominated, reaction
     products with di-Me sulfide
     RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)
        (prepn. of hydrophilic heat-sensitive imaging polymers for direct write
        printing plate for imaging without wet processing)
TΤ
     66-27-3, Methyl methanesulfonate 1072-63-5, 1-Vinylimidazole
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (prepn. of hydrophilic heat-sensitive imaging polymers for direct write
        printing plate for imaging without wet processing)
     25067-32-7P, Methyl methacrylate-2-vinylpyridine copolymer
TΤ
     26100-41-4P, Methyl methacrylate-4-vinylpyridine copolymer
     264255-38-1P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (prepn. of hydrophilic heat-sensitive imaging polymers for direct write
        printing plate for imaging without wet processing)
     7429-90-5, Aluminum, uses
IT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (prepn. of printing plates for imaging without wet
        processing using aluminum support)
     325975-94-8
ΙT
     RL: TEM (Technical or engineered material use); USES (Uses)
        (prepn. of printing plates for imaging without wet
        processing using sol-gel)
ΙT
     222961-29-7, CAB-O-JET 200
     RL: TEM (Technical or engineered material use); USES (Uses)
        (prepn. of printing plates for imaging without wet
        processing using sol-gel and)
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Page 45

TΤ 25038-59-9, Poly(ethylene terephthalate), uses RL: TEM (Technical or engineered material use); USES (Uses) (support; prepn. of printing plates for imaging without wet processing using polyester support) THERE ARE 20 CITED REFERENCES AVAILABLE FOR THIS RECORD RE.CNT 20 (1) Ali; US 5460918 1995 HCAPLUS (2) Anon; JP 58097042 1983 HCAPLUS (3) Anon; WO 9209934 1990 HCAPLUS (4) Anon; EP 0652482 A1 1993 HCAPLUS (5) Anon; EP 0615162 1994 (6) Anon; EP 609930 1994 HCAPLUS (7) Anon; WO 9739894 1997 HCAPLUS (8) Esumi; US 4634659 1987 HCAPLUS (9) Etoh; US 4405705 1983 HCAPLUS (10) Grant; Hackh's Chemical Dictionary, Fourth Edition P515 (11) Lee; US 4548893 1985 HCAPLUS (12) Ma; US 5512418 1996 (13) Pacansky; US 4081572 1978 HCAPLUS (14) Peterson; US 3964389 1976 HCAPLUS (15) Rosen, S; Fundamental Princples of Polymeric Materials, Second Edition 1993, P15 (16) Schwartz; US 4693958 1987 HCAPLUS (17) Takahashi; US 5569573 1996 HCAPLUS (18) Totsuka; US 4920036 1990 HCAPLUS (19) Totsuka; US 5691103 1997 HCAPLUS (20) Uhlig; US 4034183 1977 HCAPLUS IT 115708-89-9P, Poly(p-xylidenetetrahydro-thiophenium chloride) 325975-90-4P RL: PEP (Physical, engineering or chemical process); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses) (prepn. of hydrophilic heat-sensitive imaging polymers for direct write



LEE 09/828075 Page 46 RN 325975-90-4 HCAPLUS CN 1H-Imidazolium, 1-ethenyl-3-methyl-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) INDEX NAME) CM 1 CRN 72607-53-5 CMF C7 H14 N2 O . C1 H H₂C $Me-C-C-NH-(CH_2)_3-NH_2$ ● HCl CM 2 CRN 13474-25-4 CMF C6 H9 N2 . C1 Мe $CH = CH_2$ ● Cl~ *** FRAGMENT DIAGRAM IS INCOMPLETE *** ΙT 264255-38-1P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (prepn. of hydrophilic heat-sensitive imaging polymers for direct write printing plate for imaging without wet processing) RN 264255-38-1 HCAPLUS CN 1H-Imidazolium, 1-ethenyl-3-methyl-, methanesulfonate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM

1

CRN 72607-53-5

CMF C7 H14 N2 O . Cl H

Page 47

$$\begin{array}{c|c} ^{H2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-} \text{C-} \text{C-} \text{NH-} \text{(CH}_2)_3 - \text{NH}_2 \end{array}$$

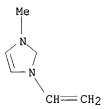
● HCl

CM 2

CRN 264255-37-0 CMF $C6\ H9\ N2$. C $H3\ O3\ S$

> CM 3

CRN 45534-45-0 CMF C6 H9 N2



*** FRAGMENT DIAGRAM IS INCOMPLETE ***

. CM 4

> CRN 16053-58-0 CMF C H3 O3 S

L85 ANSWER 12 OF 46 HCAPLUS COPYRIGHT 2002 ACS

2000:875699 HCAPLUS AN

134:49238 DN

ΤI Thermal imaging composition and member containing sulfonated IR dye and methods of imaging and printing Fleming, James C.; Leon, Jeffrey W.; Stegman, David A.; Williams, Kevin W.

IN

Eastman Kodak Company, USA PΑ

SO U.S., 22 pp. CODEN: USXXAM

Patent DT

LA English

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LEE 09/828075
                      Page 48
     ICM G03C001-73
IC
     ICS G03C001-76; G03C001-77
NCL
     430270100
     74-6 (Radiation Chemistry, Photochemistry, and/Photographic and Other
     Reprographic Processes)
FAN.CNT 1
                                            APPLICATION NO. DATE
                      KIND DATE
     PATENT NO.
PI US 6159657 A
DE 10042294 A1
JP 2001130159 A2
PRAI US 1999-387021 A
                      A
                                            us 1999-387021
                                                             19990831
                            20001212
                                            DE/2000-10042294 20000829
                            20010412
                                            J≱ 2000-262836
                            20010515
                                                             20000831
                           19990831
     An imaging member, such as a neg.-working printing plate
     or on-press cylinder, can be prepd. With a hydrophilic imaging layer
     comprised of a heat-sensitive hydrophilic polymer having ionic moieties
     and an IR radiation sensitive dye having multiple sulfo groups.
     heat-sensitive polymer and IR dye/can be formulated in water or
     water-miscible solvents to provide highly thermal sensitive imaging
     compns. In the imaging member, the polymer reacts to provide increased hydrophobicity in areas exposed to energy that provides or generates heat.
     For example, heat can be supplied by laser irradn. in the IR region of the
     electromagnetic-spectrum. The heat-sensitive polymer is considered
     "switchable" in response to heat, and provides a lithog. image
     without wet processing.
     lithog printing plates; thermal sensitive
ST
     polymer cyanine dye
     Cyanine dyes
IT
       Lithographic plates
     Thermal printing mater als
        (thermal imaging compn. and member contg. sulfonated IR dye)
     75-18-3, Dimethyl su/fide 75-75-2, Methanesulfonic acid
                                                                  77-78-1.
     Dimethylsulfate 1/0-18-9, N,N,N',N'-Tetramethylethylene diamine
     333-27-7, Methyl triflate 594-09-2, Trimethyl phosphine
                                                                   5188-07-8,
     Sodium methanethiolate 7647-15-6, Sodium bromide, reactions
     14216-23-0, 2-(M≠thylthio)ethylmethacrylate 52747-02-1
                                                                 57458-41-0
     RL: RCT (Reactart); RACT (Reactant or reagent)
         (thermal imaging compn. and member contg. sulfonated IR dve)
                  2/9636-96-2P 99044-67-4P 119261-36-8P 304023-76-5P
     7268-68-0P
TΨ
                   / 312963-52-3P 312965-31-4P 312966-33-9P
     312539-11-0P /
     312966-34-0P/312966-36-2P 312966-37-3P 312966-38-4P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
         (thermal imaging compn. and member contg. sulfonated IR dye)
     66-27-3P, Methyl methanesulfonate 1072-63-5P, 1-Vinylimidazole
     24938-67-8DP, Poly(2,6-dimethyl-1,4-phenylene oxide), brominated, reaction
     products with di-Me sulfide 25067-32-7P 25134-01-4DP,
     Poly(2,6-dimethyl-1,4-phenylene oxide), brominated, reaction products with
                     25212-74-2DP, Poly(phenylene sulfide), reaction products
     di-Me sulfide
     with methanesulfonic acid and Me triflate, chloride-exchanged
     26100-41-4P, Methyl methacrylate-4-vinylpyridine copolymer
     110866-77-8P
                    264255-37-0P 264255-38-1DP,
                           264255-39-2DP, ion exchange with formic acid
     chloride-exchanged
     264255-78-9DP, ion exchange with formic acid 264255-79-0DP, ion exchange
     with formic acid 304023-71-ODP, ion exchange from chloride to
     acetate or fluoride 304023-71-0P 304023-76-5DP, ion
     exchange from methylsulfate to chloride 312539-12-1P
     312966-33-9DP, ion exchange from bromide to acetate
     312966-40-8DP, ion exchange from Me sulfate to chloride
     RL: SPN (Synthetic preparation); PREP (Preparation)
         (thermal imaging compn. and member contg. sulfonated IR dye)
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Page 49
LEE 09/828075
ΙT
     100356-86-3P 113995-59-8P
                                  119261-38-0P
                                                  262283-81-8P
     262283-83-0P 312963-46-5P 312963-48-7P
                                             312963-49-8P
     312963-50-1P
                   312963-51-2P
     RL: SPN (Synthetic preparation); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (thermal imaging compn. and member contq. sulfonated IR dye)
RE.CNT
        19
             THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
(1) Anon; EP 0251282 1988 HCAPLUS
(2) Anon; WO 9209934 1992 HCAPLUS
(3) Anon; EP 0652483 A1 1995 HCAPLUS
(4) Anon; Research Disclosure, Item 19201 1980
(5) Esumi; US 4634659 1987 HCAPLUS
(6) Etoh; US 4405705 1983 HCAPLUS
(7) Gelbart; US 5713287 1998
(8) Laganis; US 4882265 1989 HCAPLUS
(9) Lee; US 4548893 1985 HCAPLUS
(10) Leenders; US 5378580 1995 HCAPLUS
(11) Lewis; US 5339737 1994
(12) Lewis; US 5353705 1994
(13) Lewis; US 5385092 1995
(14) Ma; US 5512418 1996
(15) Nowak; US 35512 1997
(16) Pacansky; US 4081572 1978 HCAPLUS
(17) Schwartz; US 4693958 1987 HCAPLUS
(18) Uhlig; US 4034183 1977 HCAPLUS
(19) West; US 5107068 1992 HCAPLUS
     304023-76-5P 312966-33-9P 312966-36-2P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); RACT (Reactant or reagent)
        (thermal imaging compn. and member contg. sulfonated IR dye)
RN
     304023-76-5 HCAPLUS
CN
     Sulfonium, dimethyl[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl]-, methyl
     sulfate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide
     monohydrochloride (9CI) (CA INDEX NAME)
     CM
     CRN 72607-53-5
     CMF C7 H14 N2 O . C1 H
 H<sub>2</sub>C O
Me-C-C-NH-(CH_2)_3-NH_2
```

HCl

CM 2

CRN 7268-68-0 CMF C8 H15 O2 S . C H3 O4 S

CM 3

Page 50

CRN 44992-92-9 CMF C8 H15 O2 S

CM 4

CRN 21228-90-0 CMF C H3 O4 S

Me-0-S03-

RN 312966-33-9 HCAPLUS

CN Phosphonium, [[3(or 4)-ethenylphenyl]methyl]trimethyl-, bromide, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 312965-31-4 CMF C12 H18 P . Br CCI IDS



D1-CH-CH2

 Me_3+P-CH_2-D1

• Br-

CM 2

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

Page 51

HCl

RN 312966-36-2 HCAPLUS

CN Sulfonium, [{2(or 4)-ethenylphenyl]methyl]-dimethyl-, methyl sulfate, homopolymer (9CI) (CA INDEX NAME)

CM 3

CRN 312966-35-1 CMF C11 H15 S CCI IDS



$$D1-CH=CH_2$$

CM 2

CRN 21228-90-0 CMF C H3 O4 S

Me-0-503-

IT 110866-77-8P 264255-38-1DP, chloride-exchanged 304023-71-0DP, ion exchange from chloride to acetate or fluoride 304023-71-0P 304023-76-5DP, ion exchange from methylsulfate to chloride 312966-33-9DP, ion exchange from bromide to acetate 312966-40-8DP, ion exchange from Me sulfate to chloride

RL: SPN (Synthetic preparation); PREP (Preparation) (thermal imaging compn. and member contg. sulfonated IR dye) 110866-77-8 HCAPLUS

CN Thiophenium, 1,1'-[1,4-phenylenebis(methylene)]bis[tetrahydro-, dichloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

Page 52

CRN 52547-07-6 CMF C16 H24 S2 . 2 C1

●2 C1-

RN 264255-38-1 HCAPLUS

CN 1H-Imidazolium, 1-ethenyl-3-methyl-, methanesulfonate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5

CMF C7 H14 N2 O . Cl H

$$\begin{array}{c|c} ^{H2C} & O \\ \mid \mid & \mid \mid \\ ^{Me-C-C-NH-(CH2)3-NH2} \end{array}$$

● HCl

CM 2

CRN 264255-37-0

 \mathtt{CMF} C6 H9 N2 . C H3 O3 S

CM 3

CRN 45534-45-0 CMF C6 H9 N2

*** FRAGMENT DIAGRAM IS INCOMPLETE ***

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

Page 53

CM 4

CRN 16053-58-0 CMF C H3 O3 S

RN 304023-71-0 HCAPLUS

Ethanaminium, N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM :

CRN 72607-53-5

CMF C7 H14 N2 O . Cl H

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ & || & || \\ \text{Me-C-C-NH-(CH}_2)} & 3^{-} \text{NH}_2 \end{array}$$

● HCl

CM 2

CRN 5039-78-1 CMF C9 H18 N O2 . Cl

● c1-

CM 3

CRN 80-62-6 CMF C5 H8 O2

Page 54

RN 304023-71-0 HCAPLUS

CN Ethanaminium, N,N,N-trimethyl-2-[(2-methyl-1-oxo-2-propenyl)oxy]-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride and methyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM :

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

$$\begin{array}{c|c} ^{H2C} & \text{O} \\ || & || \\ \text{Me-C-C-NH-(CH2)} \, \text{3-NH2} \end{array}$$

● HCl

CM 2

CRN 5039-78-1 CMF C9 H18 N O2 . C1

$$\begin{array}{c|c} & \text{O} & \text{CH}_2 \\ & || & || \\ \text{Me}_3 + \text{N} - \text{CH}_2 - \text{CH}_2 - \text{O} - \text{C} - \text{C} - \text{Me} \end{array}$$

● cl-

CM 3

CRN 80-62-6 CMF C5 H8 O2

RN 304023-76-5 HCAPLUS

CN Sulfonium, dimethyl[2-[(2-methyl-1-oxo-2-propenyl)oxy]ethyl]-, methyl sulfate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

Page 55

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

HCl

CM 2

CRN 7268-68-0

CMF C8 H15 O2 S . C H3 O4 S

CM 3

CRN 44992-92-9 CMF C8 H15 O2 S

CM 4

CRN 21228-90-0 CMF C H3 O4'S

Me-0-503-

RN 312966-33-9 HCAPLUS

CN Phosphonium, [[3(or 4)-ethenylphenyl]methyl]trimethyl-, bromide, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 312965-31-4 CMF C12 H18 P . Br CCI IDS

Page 56



 $D1-CH=CH_2$

 ${
m Me}_3^{+}{
m P}-{
m CH}_2-{
m D1}$

● Br-

CM 2

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

• HCl

RN 312966-40-8 HCAPLUS
CN Sulfonium, [2(or 4)-ethenylphenyl]dimethyl-, methyl sulfate, homopolymer
(9CI) (CA INDEX NAME)

CM :

CRN 312966-39-5 CMF C10 H13 S CCI IDS

Page 57



 $D1-CH = CH_2$

$$Me-S+Me$$

CM 2

CRN 21228-90-0 CMF C H3 O4 S

Me-0-S03-

IT 100356-86-3P 312963-46-5P 312963-48-7P

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(thermal imaging compn. and member contg. sulfonated IR dye)

RN 100356-86-3 HCAPLUS

CN Benzenemethanaminium, ar-ethenyl-N,N,N-trimethyl-, chloride, polymer with 2-methyl-2-propenoic acid (9CI) (CA INDEX NAME)

CM 1

CRN 26616-35-3 CMF C12 H18 N . C1 CCI IDS

 $D1-CH=CH_2$

 Me_3+N-CH_2-D1

Cl-

Page 58

CM 2

CRN 79-41-4 CMF C4 H6 O2

CH₂ || Me-C-CO₂H

RN 312963-46-5 HCAPLUS

CN Benzenemethanaminium, 3(or 4)-ethenyl-N,N,N-trimethyl-, chloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 74311-76-5 CMF C12 H18 N . C1 CCI IDS

D1-CH-CH2

 Me_3+N-CH_2-D1

● Cl- .

CM 2

CRN 72607-53-5 CMF C7 H14 N2 O . Cl H

HC1

RN 312963-48-7 HCAPLUS CN 1,2-Ethanediaminium, N-[(4-ethenylphenyl)methyl]-N,N,N',N',n'-pentamethyl-

Page 59

, dichloride, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM

CRN 312963-47-6 CMF C16 H28 N2 . 2 Cl

$$\begin{array}{c} \text{Me} \\ \text{Me}_3 + \text{N} - \text{CH}_2 - \text{CH}_2 - \text{N} + \text{CH}_2 \\ \text{Me} \end{array}$$

●2 C1-

CM 2

CRN 72607-53-5

CMF C7 H14 N2 O . C1 H

$$^{\rm H2C}_{||}$$
 O $_{||}$ $^{\rm H2}_{||}$ Me-C-C-NH-(CH2)3-NH2

● HCl

L85 ANSWER 13 OF 46 HCAPLUS COPYRIGHT 2002 ACS

2000:680396 HCAPLUS AN

DN 133:274344

Thermally reactive near infrared/absorption polymer coatings, method of ΤI preparing and methods of use

Nguyen, My T. IN

American Dye Source, Inp., Can. PA

SO U.S., 16 pp. CODEN: USXXAM

DT. Patent

LA English

ICM C08G073-00 IC

528422000 NCL

74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other CC Reprographic Processes)

r AIN.	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	US 6124425	А д 1	20000926	US 1999-275032 WO 2000-CA296	19990318 20000317

```
LEE 09/828075
                      Page 60
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, CA, CH, CN, CR,
             CU, CZ, DE, DK, DM, DZ, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU,
             ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU,
             LV, MA, MD, MG, MK, MN, MW, MX, NO, NZ, PL, PT, RO, RU, SD, SE,
             SG, SI, SK, SL, TJ, TM, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM
         RW: GH, GM, KE, LS, MW, SD, SL, SZ, TZ, UG, ZW, AT, BE, CH, CY, DE,
             DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, BF, BJ, CF,
             CG, CI, CM, GA, GN, GW, ML, MR, NE, SN, TD, TG
                       A1
                           20011212
                                           EP 2000-910470
            AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, LV, FI, RO
                     . B1
     US 6177182
                            20010123
                                           US 2000-561817 20000501
PRAI US 1999-275032
                       Α
                            19990318
     WO 2000-CA296
                       W
                            20000317
     Provided here are novel polymeric coating materials for direct digital
   imaging by laser. More specifically the novel coating materials are
     thermally reactive near IR absorption polymers designed for use with near
     IR laser imaging devices. This invention further extends to the prepn.
     and methods of use of the novel materials. The invention is particularly
     useful in the prepn. of lithog. printing
     plates for computer-to-plate and digital-offset-press
     technologies. The invention extends to photoresist applications, to rapid
     prototyping of printed circuit boards and to chem. sensor development.
ST
     thermally reactive IR absorption polymer coating
TΤ
     IR lasers
        (near-IR; prepn. of chem. sensor for measuring electrode cond. in
        direct digital laser imaging)
TT
     Coating materials
     Imaging
       Lithographic plates
     Photoresists
     Printed circuit boards
        (prepn. of chem. sensor for measuring electrode cond. in direct digital
        laser imaging)
ΙT
     Polyanilines
     Polyesters, uses
     RL: NUU (Other use, unclassified); USES (Uses)
        (prepn. of chem. sensor for measuring electrode cond. in direct digital
        laser imaging)
TΤ
     68584-99-6, P3000
     RL: NUU (Other use, unclassified); RCT (Reactant); RACT (Reactant or
     reagent); USES (Uses)
        (aq. developer; pos. thermal printing plate for
        direct digital laser imaging developed using)
IΤ
     872-50-4, 1-Methyl-2-pyrrolidinone, uses
                                                25233-30-1, Polyaniline
     RL: NUU (Other use, unclassified); USES (Uses)
        (prepn. of chem. sensor for measuring electrode cond. in direct digital
        laser imaging)
     26355-01-1, Methyl methacrylate-2-hydroxyethyl methacrylate copolymer
TΤ
     28015-39-6, Methyl methacrylate-N-(methoxymethyl)methacrylamide copolymer
     139301-16-9
     RL: NUU (Other use, unclassified); RCT (Reactant); RACT (Reactant or
     reagent); USES (Uses)
        (prepn. of neg. thermal printing plate for direct
        digital laser imaging using)
ΙT
     9016-83-5DP, SD 140A, ethers with cyanine dyes
                                                       110123-09-6DP,
     ethers with cyanine dyes 134127-48-3DP, ethers with
                               247248-90-4DP, ethers with hydroxy-contg.
     hydroxy-contg. polymers
```

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polymers 297174-00-6P 297174-03-9P
     297174-06-2P 297174-07-3P 297174-09-5P
     297174-11-9P 297174-13-1P 297174-15-3P
     297174-17-5P 297174-18-6P 297174-20-0P
     297752-34-2DP, ethers with cyanine dyes
     RL: IMF (Industrial manufacture); NUU (Other use, unclassified);
     PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)
        (synthesis of near-IR absorption polymer thermal coatings for direct
        digital imaging by laser)
RE.CNT
              THERE ARE 39 CITED REFERENCES AVAILABLE FOR THIS RECORD
RE
(1) Anon; GB 1489308 1977 HCAPLUS
(2) Anon; WO 0652483 Al 1994
(3) Anon; WO 9620429 1996 HCAPLUS
(4) Anon; EP 0770495 A1 1997 HCAPLUS
(5) Anon; EP 0770497 A1 1997 HCAPLUS
(6) Anon; EP 0773113 A1 1997 HCAPLUS
(7) Anon: WO 9739894 1997 HCAPLUS
(8) Anon; DE 0867278 A1 1998
(9) Anon; GB 2273366 1998
(10) Anon; EP 0514145 A1 2000 HCAPLUS
(11) Anon; EP 0770494 A2 2000 HCAPLUS
(12) Anon; EP 0770496 A1 2000 HCAPLUS
(13) Anon; EP 0773112 A1 2000 HCAPLUS
(14) Anon; EP 0774364 A1 2000 HCAPLUS
(15) Anon; EP 0800928 A1 2000 HCAPLUS
(16) Burns; US 5824768 1998 HCAPLUS
(17) Caddell; US 4054094 1977 HCAPLUS
(18) Eames; US 3962513 1976 HCAPLUS
(19) Elmasry; US 4666819 1987 HCAPLUS
(20) Elmasry; US 4680375 1987 HCAPLUS
(21) Fan; US 5262275 1993 HCAPLUS
(22) Gamson; US 4555475 1985
(23) Gravesteijn; US 4508811 1985 HCAPLUS
(24) Holmes; US 5362812 1994 HCAPLUS
(25) Holmes; US 5741620 1998 HCAPLUS
(26) Kashio; US 5665524 1997 HCAPLUS
(27) Leenders; US 5595854 1997 HCAPLUS
(28) Ma; US 5292556 1994
(29) Mitra; US 4477635 1984 HCAPLUS
(30) Nussstein; US 5360899 1994 HCAPLUS
(31) Ohno; US 5547819 1996 HCAPLUS
(32) Oransky; US 4245003 1981 HCAPLUS
(33) Pacansky; US 4081572 1978 HCAPLUS
(34) Persley; US 08922714 1997
(35) Peterson; US 3964389 1976 HCAPLUS
(36) Shaw; US 4046946 1977 HCAPLUS
(37) Takahashi; US 5569573 1996 HCAPLUS
(38) Vogel; US 5085972 1992 HCAPLUS
(39) Zahr; US 4501876 1985 HCAPLUS
     297174-00-6P 297174-03-9P 297174-06-2P
     297174-07-3P 297174-09-5P 297174-11-9P
     297174-13-1P 297174-15-3P 297174-17-5P
     297174-18-6P 297174-20-0P
     RL: IMF (Industrial manufacture); NUU (Other use, unclassified);
     PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)
        (synthesis of near-IR absorption polymer thermal coatings for direct
        digital imaging by laser)
     297174-00-6 HCAPLUS
RN
CN
     1H-Benz[e]indolium, 2-[2-[2-chloro-3-[(1,3-dihydro-1,1,3-trimethyl-2H-
```

Page 62

benz[e]indol-2-ylidene)ethylidene]-1-cyclohexen-1-yl]ethenyl]-1,1,3trimethyl-, chloride, compd. with 4-ethenylpyridine polymer with N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM

CRN 297173-98-9

CMF C40 H40 C1 N2 . C1

● cl-

CM 2

CRN 297173-99-0

(C7 H7 N . C6 H11 N O2)x CMF

CCI PMS

> CM 3

CRN 3644-12-0

CMF C6 H11 N O2

$$\begin{array}{c|c} ^{\text{H}_2\text{C}} & \text{O} \\ \parallel & \parallel \\ \text{Me-C-C-NH-CH}_2\text{-OMe} \end{array}$$

CM 4

CRN 100-43-6 CMF C7 H7 N

RN 297174-03-9 HCAPLUS

1H-Benz[e]indolium, 2-[2-[2-chloro-3-[[1,3-dihydro-3-(2-hydroxyethyl)-1,1-CN dimethyl-2H-benz[e]indol-2-ylidene]ethylidene]-1-cyclohexen-1-yl]ethenyl]-

Page 63

3-(2-hydroxyethyl)-1,1-dimethyl-, chloride, compd. with 4-ethenylpyridine polymer with N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 297174-02-8

CMF C42 H44 C1 N2 O2 . C1

• c1-

CM 2

CRN 297173-99-0

CMF (C7 H7 N . C6 H11 N O2) \times

CCI PMS

CM 3

CRN 3644-12-0 CMF C6 H11 N O2

$$\begin{array}{c|c} ^{H2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-C-C-NH-CH}_2\text{-OMe} \end{array}$$

CM 4

CRN 100-43-6 CMF C7 H7 N

RN 297174-06-2 HCAPLUS

CN 1H-Benz[e]indolium, 2-[2-[2-chloro-3-[[1,3-dihydro-3-(2-hydroxyethyl)-1,1-dimethyl-2H-benz[e]indol-2-ylidene]ethylidene]-1-cyclohexen-1-yl]ethenyl]-

Page 64

3-(2-hydroxyethyl)-1,1-dimethyl-, chloride, compd. with butyl 2-methyl-2-propenoate polymer with 4-ethenylpyridine and N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 297174-02-8

CMF C42 H44 C1 N2 O2 . C1

• c1-

CM2

CRN 297174-05-1

CMF (C8 H14 O2 . C7 H7 N . C6 H11 N O2)x

CCI PMS

> CM 3

CRN 3644-12-0

CMF C6 H11 N O2

CM 4

CRN 100-43-6 CMF C7 H7 N

5 . CM

Page 65

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

RN 297174-07-3 HCAPLUS

CN 1H-Benz[e]indolium, 2-[2-[2-chloro-3-[(1,3-dihydro-1,1,3-trimethyl-2H-benz[e]indol-2-ylidene)ethylidene]-1-cyclohexen-1-yl]ethenyl]-1,1,3-trimethyl-, chloride, compd. with 2-chloroethenol and 4-ethenylpyridine polymer with N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 297173-98-9

CMF C40 H40 C1 N2 . C1

• c1-

CM 2

CRN 107-07-3 CMF C2 H5 Cl O

 ${\rm C1-CH_2-CH_2-OH}$

CM 3

CRN 297173-99-0 CMF (C7 H7 N .

(C7 H7 N . C6 H11 N O2)x

CCI PMS

CM 4

Page 66

$$\begin{array}{ccc} ^{\rm H2C} & {\rm O} \\ \parallel & \parallel \\ {\rm Me-C-C-NH-CH_2-OMe} \end{array}$$

CM 5

CRN 100-43-6 CMF C7 H7 N

RN 297174-09-5 HCAPLUS

CN 1H-Benz[e]indolium, 2-[2-[2-chloro-3-[(1,3-dihydro-1,1-dimethyl-3-propyl-2H-benz[e]indol-2-ylidene)ethylidene]-1-cyclopenten-1-yl]ethenyl]-1,1-dimethyl-3-propyl-, chloride, compd. with 4-ethenylpyridine polymer with N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 297174-08-4 CMF C43 H46 C1 N2 . C1

• c1-

CM 2

CRN 297173-99-0

CMF (C7 H7 N . C6 H11 N O2) \times

CCI PMS

CM 3

Page 67

$$\begin{array}{c|c} ^{H_2C} & \text{O} \\ & || & || \\ \text{Me-C-C-NH-CH}_2\text{-OMe} \end{array}$$

CM 4

CRN 100-43-6 CMF C7 H7 N

RN 297174-11-9 HCAPLUS

CN Naphtho[2,1-d]thiazolium, 2-[2-[2-chloro-3-[(3-methylnaphtho[2,1-d]thiazol-2(3H)-ylidene)ethylidene]-1-cyclohexen-1-yl]ethenyl]-3-methyl-, chloride, compd. with 4-ethenylpyridine polymer with N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 297174-10-8 CMF C34 H28 C1 N2 S2 . C1

• c1-

CM 2

CRN 297173-99-0 CMF (C7 H7 N . C6 H11 N O2)x

CCI PMS

CM 3

Page 68

$$\begin{array}{c|c} ^{\text{H}_2C} & \text{O} \\ \parallel & \parallel \\ ^{\text{Me}-C-C-NH-CH}_2\text{-OMe} \end{array}$$

CM 4

CRN 100-43-6 CMF C7 H7 N

RN 297174-13-1 HCAPLUS

CN Naphtho[2,1-d]selenazolium, 2-[2-[2-chloro-3-[(3-methylnaphtho[2,1-d]selenazol-2(3H)-ylidene)ethylidene]-1-cyclohexen-1-yl]ethenyl]-3-methyl-, chloride, compd. with 4-ethenylpyridine polymer with N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 297174-12-0 CMF C34 H28 C1 N2 Se2 . C1

● Cl-

CM 2

CRN 297173-99-0

CMF (C7 H7 N . C6 H11 N O2)x

CCI PMS

CM 3

Page 69

$$\begin{array}{c|c} ^{H2C} & \text{O} \\ \parallel & \parallel \\ \text{Me-C-C-NH-CH}_2\text{-OMe} \end{array}$$

CM 4

CRN 100-43-6 CMF C7 H7 N

RN 297174-15-3 HCAPLUS

CN Naphth[2,1-d]oxazolium, 2-[2-[2-chloro-3-[(3-methylnaphth[2,1-d]oxazol-2(3H)-ylidene)ethylidene]-1-cyclohexen-1-yl]ethenyl]-3-methyl-, chloride, compd. with 4-ethenylpyridine polymer with N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 297174-14-2 CMF C34 H28 C1 N2 O2 . C1

• c1-

CM 2

CRN 297173-99-0

CMF (C7 H7 N . C6 H11 N O2) \times

CCI PMS

CM 3

Page 70

$$\begin{array}{c|c} ^{\rm H2C} & {\rm O} \\ \parallel & \parallel \\ ^{\rm Me-} \, {\rm C-C-NH-CH_2-OMe} \end{array}$$

CM 4

CRN 100-43-6 CMF C7 H7 N

RN 297174-17-5 HCAPLUS

Benz[cd]indolium, 2-[2-[2-chloro-3-[(1-methylbenz[cd]indol-2(1H)-ylidene)ethylidene]-1-cyclopenten-1-yl]ethenyl]-1-methyl-, chloride, compd. with 4-ethenylpyridine polymer with N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM 1

CRN 297174-16-4 CMF C33 H26 C1 N2 . C1

$$\begin{array}{c|c} C1 \\ \hline \\ N \\ \hline \\ Me \end{array}$$

• c1-

CM 2

CRN 297173-99-0

CMF (C7 H7 N . C6 H11 N O2)x

CCI PMS

CM 3

Page 71

CM 4

CRN 100-43-6 CMF C7 H7 N

RN 297174-18-6 HCAPLUS

CN 1H-Benz[e]indolium, 2-[2-[2-chloro-3-[(1,3-dihydro-1,1,3-trimethyl-2H-benz[e]indol-2-ylidene)ethylidene]-1-cyclohexen-1-yl]ethenyl]-1,1,3-trimethyl-, chloride, compd. with 2-chloroethanol and 4-ethenylpyridine polymer with 2-hydroxyethyl 2-methyl-2-propenoate (9CI) (CA INDEX NAME)

CM 1

CRN 297173-98-9 CMF C40 H40 C1 N2 . C1

• c1-

CM 2

CRN 107-07-3 CMF C2 H5 C1 O

 $C1-CH_2-CH_2-OH$

CM 3

CRN 36180-84-4

Page 72

CMF (C7 H7 N . C6 H10 O3) \times CCI PMS

CM 4

CRN 868-77-9 CMF C6 H10 O3

CM 5

CRN 100-43-6 CMF C7 H7 N

RN 297174-20-0 HCAPLUS

CN 1H-Benz[e]indolium, 2-[2-[2-chloro-3-[(1,3-dihydro-1,1-dimethyl-3-propyl-2H-benz[e]indol-2-ylidene)ethylidene]-1-cyclopenten-1-yl]ethenyl]-1,1-dimethyl-3-propyl-, chloride, compd. with butyl 2-methyl-2-propenoate polymer with 2-(dimethylamino)ethyl 2-methyl-2-propenoate and N-(methoxymethyl)-2-methyl-2-propenamide (9CI) (CA INDEX NAME)

CM :

CRN 297174-08-4 CMF C43 H46 C1 N2 . C1

● c1-

CM 2

CRN 297174-19-7

Page 73

CMF (C8 H15 N O2 . C8 H14 O2 . C6 H11 N O2)x CCI PMS

CM 3

CRN 3644-12-0 CMF C6 H11 N O2

CM 4

CRN 2867-47-2 CMF C8 H15 N O2

$$\begin{array}{c} \text{O} \quad \text{CH}_2 \\ \parallel \quad \parallel \\ \text{Me}_2 \text{N} - \text{CH}_2 - \text{CH}_2 - \text{O} - \text{C} - \text{C} - \text{Me} \end{array}$$

CM 5

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

L85 ANSWER 14 OF 46 HCAPLUS COPYRIGHT 2002 ACS

AN 2000:254687 HCAPLUS

DN 132:300960

TI Heat-sensitive **lithographic** imaging material and imaging method using it

IN Leon, Jeffrey W.; Underwood, Gary Marshall; Fleming, James C.; De Boer, Charles David

PA Eastman Kodak Co., USA

SO Jpn. Kokai Tokkyo Koho, 14 pp. CODEN: JKXXAF

DT Patent

LA Japanese

IC ICM G03F007-027

ICS C08L033-06; C08L039-04; C08L101-02; G03F007-00; G03F007-029

CC 74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes) Section cross-reference(s): 38

FAN.CNT 2

PATENT NO. KIND DATE APPLICATION NO. DATE
PI JP 2000112123 A2 20000421 JP 1999-276928 19990929

```
LEE 09/828075
                      Page 74
     US 6190831
                       В1
                            20010220
                                            US 19997310038
                                                             19990511
PRAI US 1998-163020
                            19980929
                       Α
     US 1999-310038
                            19990511
                       Α
     The imaging material has a hydrophilic imag/ng layer contq. vinyl polymers
     having pos.-charged N-alkyl-pendent arom. Meterocycle repeating units or
     nonvinyl polymers contg. org. onium_repearing units as hydrophilic
     heat-sensitive polymers. The method comprises imagewise irradn. of an
     energy to the above imaging material to make the exposed area more
     lipophilic than unexposed areas. The material gives neg.-working
     lithog. plates without development.
ST
     heat sensitive lithog plate quaternary ammonium polymer; onium
     pendent heat sensitive polymer lithog plate; neg lithog
     printing plate development free
IT
     Heat-sensitive materials
       Lithographic plates
        (heat-sensitive lithog. imaging/material contg.
        N-alkyl-pendent vinyl polymers/or onium unit-contg. nonvinyl polymer as
        heat-sensitive material)
TΤ
     Quaternary ammonium compounds, uses
     RL: DEV (Device component use);/USES (Uses)
        (polymers; heat-sensitive lithog. imaging material contg.
        N-alkyl-pendent vinyl polymers or onium unit-contg. nonvinyl polymer as
        heat-sensitive material)
    75-18-3DP, Dimethyl sulfide, compd. with brominated polyoxyphenylenes 24938-67-8DP, 2,6-Xylenol homopolymer, sru, brominated, compd. with di-Me
TΤ
     sulfide 25134-01-4DP, 2,6/Xylenol homopolymer, brominated, compd. with
                    25212-74-2pp, Poly(phenylene sulfide), chloride ion
     di-Me sulfide
     exchanged, reaction produ¢t with methanesulfonic acid 110866-77-8P
     264255-38-1DP, chloride ion exchanged 264255-39-2P, Methyl
     methacrylate-4-vinylpyridine copolymer-1-bromobutane compd.
     264255-78-9P, Methyl methacrylate-4-vinylpyridine copolymer
     methyl p-toluenesulfonate 264255-79-0P, Methyl methacrylate-2-
     vinylpyridine copolymer methyl p-toluenesulfonate
     RL: DEV (Device component use); IMF (Industrial manufacture);
     PREP (Preparation); USES (Uses)
        (heat-sensitive lithog. imaging material contg.
        N-alkyl-pendent √inyl polymers or onium unit-contg. nonvinyl polymer as
        heat-sensitive material)
ΙT
     264255-37-0P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation); RACT
     (Reactant or reagent)
        (prepn. and polymn. of; heat-sensitive lithog. imaging
        material confg. N-alkyl-pendent vinyl polymers or onium unit-contg.
        nonvinyl polymer as heat-sensitive material)
     1072-63-5, 1-Vinylimidazole
IT
     RL: RCT (Reacfant); RACT (Reactant or reagent)
        (reaction/with Me methanesulfonate; heat-sensitive lithog.
        imaging material contg. N-alkyl-pendent vinyl polymers or onium
        unit-confg. nonvinyl polymer as heat-sensitive material)
     66-27-3, Methyl methanesulfonate
ΙT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (reaction with vinylimidazole; heat-sensitive lithog. imaging
        material contg. N-alkyl-pendent vinyl polymers or onium unit-contg.
        nonvinyl polymer as heat-sensitive material)
ΙT
     110866-77-8P 264255-38-1DP, chloride ion exchanged
     RL: DEV (Device component use); IMF (Industrial manufacture);
     PREP (Preparation); USES (Uses)
        (heat-sensitive lithog. imaging material contg.
        N-alkyl-pendent vinyl polymers or onium unit-contg. nonvinyl polymer as
        heat-sensitive material)
```

LEE 09/828075 Page 75

RN 110866-77-8 HCAPLUS

CN Thiophenium, 1,1'-[1,4-phenylenebis(methylene)]bis[tetrahydro-,
dichloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 52547-07-6 CMF C16 H24 S2 . 2 C1

●2 C1-

RN 264255-38-1 HCAPLUS

CN 1H-Imidazolium, 1-ethenyl-3-methyl-, methanesulfonate, polymer with N-(3-aminopropyl)-2-methyl-2-propenamide monohydrochloride (9CI) (CA INDEX NAME)

CM 1

CRN 72607-53-5 CMF C7 H14 N2 O . C1 H

● HCl

CM 2

CRN 264255-37-0 CMF C6 H9 N2 . C H3 O3 S

CM 3

CRN 45534-45-0 CMF C6 H9 N2

$$\begin{array}{c}
\text{Me} \\
| \\
\text{N} \\
\text{CH} = \text{CH}_2
\end{array}$$

*** FRAGMENT DIAGRAM IS INCOMPLETE ***

CM 4

CRN 16053-58-0 CMF C H3 O3 S

L85 ANSWER 15 OF 46 HCAPLUS COPYRIGHT 2002 ACS

KATHLEEN FULLER ELÉ 1700/LAW LIBRARY 308-4290

```
1999:789817 HCAPLUS
AN
     132:28708
DN
     Photosensitive polymer aqueous emulsion for manufacture of screen
ΤI
     printing plate
     Tsuchida, Keiko; Morigaki, Toshio
ΤN
     Gooh Chemical Industry Co., Ltd., Japan
PΑ
     Jpn. Kokai Tokkyo Koho, 8 pp.
SO
     CODEN: JKXXAF
     Patent
DT
     Japanese
T.A
     ICM G03F007-004
     ICS G03F007-033; G03F007-12
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
     Section cross-reference(s): 45
FAN.CNT 1
                                                APPLICATION NO.
                                                                   DATE
                         KIND DATE
     PATENT NO.
                               19991214/
                                                JP 1998-170693 19980602
                         A2
      JP 11344801
PΤ
     The emulsion contains a 100:(20-4000) (wt. ratio) mixt. of water-sol. and water-insol. polymers and 0.2-10 wt.% (as solids) waxes. A screen printing plate obtained from the emulsion shows improved
     water resistance and water/repellency and gives clear images.
     photosensitive polymer ad emulsion screen printing plate
ST
      ; wax photosensitive emulsion screen printing water resistance
      Paraffin waxes, uses
TΨ
      RL: DEV (Device component use); MOA (Modifier or additive use); USES
         (Dyedit EY, Syetex K 2500; wax-contg. photosensitive polymer aq.
         emulsions for manuf. of screen printing plates)
      Printing plates
         (screen; wax-contg. photosensitive polymer aq. emulsions for manuf. of
```

```
LEE 09/828075 Page 77
screen printing plates)
IT Beeswax
```

Photoimaging materials (wax-contg. photosensitive polymer aq. emulsions for manuf. of screen printing plates)

IT 9003-20-7, Poly(vinyl acetate)

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(Nikasol H 03; wax-contg. photosensitive polymer aq. emulsions for manuf. of screen **printing plates**)

IT 9002-89-5DP, Gohsenol GH 17, reaction products with methyl(formylstyryl)
 pyridinium methosulfate 29989-17-1P 74401-04-0DP,
 N-Methyl-4-(p-formylstyryl)pyridinium methosulfate, reaction
 products with poly(vinyl alc.)

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(wax-contg. photosensitive polymer aq. emulsions for manuf. of screen
printing plates)

IT 9002-89-5, Gohsenol GH 17 143180-25-0, Poval 224

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(wax-contg. photosensitive polymer aq. emulsions for manuf. of screen printing plates)

IT 9002-88-4, Polyethylene 9003-07-0, Polypropylene
RL: DEV (Device component use); MOA (Modifier or additive use); USES
(Uses)

(waxes; wax-contg. photosensitive polymer aq. emulsions for manuf. of screen **printing plates**)

IT 29989-17-1P

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(wax-contg. photosensitive polymer aq. emulsions for manuf. of screen
printing plates)

RN 29989-17-1 HCAPLUS

CN Benzenediazonium, 4-(phenylamino)-, chloride, polymer with formaldehyde (9CI) (CA INDEX NAME)

CM 1

CRN 101-56-4 CMF C12 H10 N3 . C1

● C1-

CM 2

CRN 50-00-0

Page 78

CMF C H2 O

 $H_2C = 0$

```
L85 ANSWER 16 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1999:427006 HCAPLUS
AΝ
     131:122986
DN
     Presensitized lithographic plate with photosensitive layer
ΤI
     containing microgel
     Kojima, Noriyoshi
IN
     Konica Co., Japan
PΑ
     Jpn. Kokai Tokkyo Koho, 18 pp.
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
     ICM G03F007-00
ICS G03F007-004
TC
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 38
FAN.CNT 1
                                            APPLICATION NO. DATE
                      KIND DATE
     PATENT NO.
                                            --/----
                     ----
                                            Jr 1997-349419 19971218
                      A2 19990709
     JP 11184073
PΙ
     The title presensitized lithog. plate comprises an Al support
AB
     coated with a microgel-cont photos nsitive layer on 1 side and with a
     backcoat layer on the other side. The lithog. plate shows good scratch resistance when a stack of a large no. of the presensitized
     lithog. plates is transported or stored.
     presensitized lithog plate microgel photosensitive layer; back
ST
     coat layer presensitized lithog plate
     Polyvinyl butyrals
IT
     RL: DEV (Device component use); USES (Uses)
         (Denka Butyral 3000, backcoat layer; presensitized lithog.
        plate with photosensitatve layer contg. microgel and backcoat layer)
ΙT
     Polyesters, uses
     Polyurethanes, uses
     Silicates, uses
     RL: DEV (Device component use); USES (Uses)
         (backcoat layer; presensitized lithog. plate with
         photosensitive Payer contg. microgel and backcoat layer)
     Lithographic plates
TT
         (presensitized; presensitized lithog. plate with
         photosensitiye layer contg. microgel and backcoat layer)
     78-10-4D, Tetraethylsilicate, hydrolyzed 25068-38-6, Pheno Tohto YP-50
                                 116094-77-0, Kemit K-1089 117847-82-2, Kemit
      91727-18-3, D∉smolac 2100
      K-1294
      RL: DEV (Device component use); USES (Uses)
         (backcoat layer; presensitized lithog. plate with
         photosensitive layer contg. microgel and backcoat layer)
      85568-56-₺, Megafac F-177
 TT
      RL: DEV /Device component use); MOA (Modifier or additive use); USES
         (backcoat layer; presensitized lithog. plate with
         photosensitive layer contg. microgel and backcoat layer)
      74443-77-9, Divinylbenzene-styrene-benzyldimethyl-p-
 IT
      styrenylmethylammonium chloride copolymer
```

Page 79

RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(presensitized lithog. plate with photosensitive layer contg. microgel and backcoat layer)

3770-97-6DP, o-Naphthoquinonediazide-5-sulfonyl chloride, reaction products with acrylic copolymer 56343-12-5DP,
Styrene-trihexyl-p-styrenylmethylammonium chloride copolymer,
reaction products with dye 74443-77-9DP, reaction
products with azidonaphthalene sulfonic acid 93673-76-8DP, reaction
products with cationic polymer 104718-47-0DP, reaction products with
cationic polymer 231301-22-7DP, reaction products with cationic polymer
231301-23-8P, tert-Butyl p-vinyl benzoate-divinylbenzene-styrene copolymer
233277-83-3DP, 1,4-Butanediol diacrylate-ethyl acrylate-ethyl
methacrylate-2-hydroxyethyl methacrylate-methacrylic acid-methyl
methacrylate copolymer, reaction products with
naphthoquinonediazidesulfonyl chloride

RL: DEV (Device component use); MOA (Modifier or additive use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses)

(presensitized lithog. plate with photosensitive layer contg. microgel and backcoat layer)

74443-77-9, Divinylbenzene-styrene-benzyldimethyl-p-styrenylmethylammonium chloride copolymer
RL: DEV (Device component use); MOA (Modifier or additive use); USES (Uses)

(presensitized ${\bf lithog.}$ plate with photosensitive layer contg. microgel and backcoat layer)

RN 74443-77-9 HCAPLUS

CN Benzenemethanaminium, 4-ethenyl-N, N-dimethyl-N-(phenylmethyl)-, chloride, polymer with diethenylbenzene and ethenylbenzene (9CI) (CA INDEX NAME)

CM 1

TΤ

CRN 66099-76-1 CMF C18 H22 N . C1

$$\begin{array}{c} \text{Me} \\ \text{Ph-CH}_2 - \text{N} \xrightarrow{+} \text{CH}_2 \\ \text{Me} \end{array}$$

● Cl-

CM 2

CRN 1321-74-0 CMF C10 H10 CCI IDS

Page 80



3 CM

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

56343-12-5DP, Styrene-trihexyl-p-styrenylmethylammonium chloride copolymer, reaction products with dye 74443-77-9DP, reaction products with azidonaphthalene sulfonic acid

RL: DEV (Device component use); MOA (Modifier or additive use); PNU (Preparation, unclassified); PREP (Preparation); USES (Uses) (presensitized lithog. plate with photosensitive layer contg. microgel and backcoat layer)

56343-12-5 HCAPLUS RN

Benzenemethanaminium, 4-ethenyl-N,N,N-trihexyl-, chloride, polymer with CN ethenylbenzene (9CI) (CA INDEX NAME)

1 CM

CRN 56343-11-4 CMF C27 H48 N . Cl

• cl-

CM 2

CRN 100-42-5 CMF C8 H8

Page 81

 $H_2C == CH - Ph$

RN 74443-77-9 HCAPLUS

Benzenemethanaminium, 4-ethenyl-N, N-dimethyl-N-(phenylmethyl)-, chloride, polymer with diethenylbenzene and ethenylbenzene (9CI) (CA INDEX NAME)

CM 1

CRN 66099-76-1 CMF C18 H22 N . Cl

$$\begin{array}{c} \text{Me} \\ \mid \\ \mid \\ \text{Ph-CH}_2 - \text{N} \xrightarrow{+} \text{CH}_2 \\ \mid \\ \text{Me} \end{array}$$

● Cl-

CM 2 ·

CRN 1321-74-0 CMF C10 H10 CCI IDS



CM 3

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

L85 ANSWER 17 OF 46 HCAPLUS COPYRIGHT 2002 ACS

AN 1999:238783 HCAPLUS

DN 130:318624

TI Photosensitive material containing infrared absorber and agent for

X

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LEE 09/828075 Page 82
     multiplying acid or diazo compound for lithographic plate
     Kudou, Shinji
ΙN
     Konica Co., Japan
Jpn. Kokai Tokkyo Koho, 41 pp.
PΑ
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
     ICM G03F007-004
     ICS G03F007-004; B41N001-14; G03F007-00
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
FAN.CNT 1
                                             APPLICATION NO. DATE
     PATENT NO.
                      KIND DATE
                                             ----
     _____
                                             JP/1997-261914 19970926
     JP 11102066 A2 19990413
PΙ
     The title material comprises a support doated with a photosensitive layer
     contg. (a) a compd. (generating acid)under active ray irradn., (b) a compd. having an acid-decomposable portion, (c) an IR absorbent, and either (d) an acid-multiplying agent (or) (e) a diazo compd. The photosensitive layer
     may contain (a), (c), either (d) or/(e), and a compd. which becomes insol.
     in alkali in the presence of acid. The material using IR exposure system
     shows improved storage stability and photosensitivity.
     lithog plate photosensitive mate ial acid multiplying; IR
ST
     absorber diazo compd lithog plate; storage stability IR exposure
     lithog plate
     Cyanine dyes
ΙT
     Optical materials
     Optical materials
         (IR absorbers; photosen itive material contg. IR absorber and agent for
        multiplying acid or diazo compd. for lithog. plate with
         storage stability)
ΙT
     IR materials
     IR materials
         (absorbers; photose sitive material contg. IR absorber and agent for
         multiplying acid of diazo compd. for lithog. plate with
         storage stability)
ΙT
     Aminoplasts
     RL: TEM (Technical/or engineered material use); USES (Uses)
         (acid-insolubilizing agent; photosensitive material contg. IR absorber
         and agent for multiplying acid or diazo compd. for lithog.
         plate with storage stability)
     Azo compounds
IΤ
      RL: TEM (Technical or engineered material use); USES (Uses)
         (photosens/tive material contg. IR absorber and agent for multiplying
         acid or diazo compd. for lithog. plate with storage
         stability)
     Lithographic plates
 IT
         (presensitized; photosensitive material contg. IR absorber and agent
         for multiplying acid or diazo compd. for lithog. plate with
         storage stability)
      Phenolic resins, uses
 IT
      RL: TEM (Technical or engineered material use); USES (Uses)
         (resol, acid-insolubilizing agent; photosensitive material contg. IR
         absorber and agent for multiplying acid or diazo compd. for
         lithog. plate with storage stability)
                   115970-68-8
                                 173474-43-6
      RL: TEM (Technical or engineered material use); USES (Uses)
         (IR absorbers; photosensitive material contg. IR absorber and agent for
         multiplying acid or diazo compd. for lithog. plate with
         storage stability)
```

LEE 09/828075 Page 83

223391-81-9P 134335-38-9P RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (acid generator; photosensitive material contg. IR absorber and agent for multiplying acid or diazo compd. for lithog. plate with storage stability) 219736-12-6 80309-01-9 42573-57-9, TAZ 110 4257-81-2 IT RL: TEM (Technical or engineered material use); USES (Uses) (acid generator; photosensitive material contg. IR absorber and agent for multiplying acid or diazo compd. for lithog. plate with storage stability) 3089-11-0, Hexamethoxymethylmelamine 212693-31-7, CKP 918 ΙT RL: TEM (Technical or engineered material use); USES (Uses) (acid-insolubilizing agent; photosensitive material contg. IR absorber and agent for multiplying acid or diazo compd. for lithog. plate with storage stability) 25266-14-2P, Ethylene oxide-formaldehyde copolymer 115815-82-2P TΨ 215865-74-0P, Cyclohexanone-ethylene glycol copolymer RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (decomposable; photosensitive material contg. IR absorber and agent for multiplying acid or diazo compd. for lithog. plate with storage stability) 108-94-1, Cyclohexanone, reactions 122-99-6, RL: RCT (Reactant); RACT (Reactant or reagent) 122-99-6, Phenyl Cellosolve TT (photosensitive material contg. IR absorber and agent for multiplying acid or diazo compd. for lithog. plate contg. decomposable compd. from) 16941-11-0DP, Ammonium hexafluorophosphate, reaction product with diazonium resin 32762-05-3DP, 4-Diazodiphenylamine hydrogen sulfate-p-hydroxybenzoic acid-formaldehyde copolymer, reaction product with ammonium hexafluorophosphate 41432-19-3DP, 4-Diazodiphenylamine hydrogen sulfate-formaldehyde copolymer, reaction product with ammonium hexafluorophosphate RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (photosensitive material contg. IR absorber and agent for multiplying acid or diazo compd. for lithog. plate with storage stability) 169262-39-9 184289-71-2 138806-47-0 168281-30-9 75620-67-6 223433-60-1 223433-62-3 200441-10-7 202058-60-4 188590-03-6 223571-08-2 RL: TEM (Technical or engineered material use); USES (Uses) (photosensitive material contg. IR absorber and agent for multiplying acid or diazo compd. for lithog. plate with storage stability) 32762-05-3DP, 4-Diazodiphenylamine hydrogen sulfate-p-TT hydroxybenzoic acid-formaldehyde copolymer, reaction product with ammonium hexafluorophosphate 41432-19-3DP, 4-Diazodiphenylamine hydrogen sulfate-formaldehyde copolymer, reaction product with ammonium hexafluorophosphate RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (photosensitive material contg. IR absorber and agent for multiplying

stability) RN 32762-05-3 HCAPLUS

CN Benzenediazonium, 4-(phenylamino)-, sulfate (1:1), polymer with formaldehyde and 4-hydroxybenzoic acid (9CI) (CA INDEX NAME)

acid or diazo compd. for lithog. plate with storage

Page 84

CM1

CRN 99-96-7 CMF C7 H6 O3

CM

CRN 50-00-0 CMF C H2 O

$H_2C = O$

3 CM

CRN 4477-28-5

CMF C12 H10 N3 . H O4 S

> 4 CM

CRN 16072-57-4 CMF C12 H10 N3

5 CM

CRN 14996-02-2 CMF H O4 S

RN 41432-19-3 HCAPLUS

Benzenediazonium, 4-(phenylamino)-, sulfate (1:1), polymer with formaldehyde (9CI) (CA INDEX NAME)

CM

Page 85

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$

2 CM

CRN 4477-28-5

CMF C12 H10 N3 . H O4 S

CM 3

CRN 16072-57-4 CMF C12 H10 N3

CM

CRN 14996-02-2 CMF H O4 S

L85 ANSWER 18 OF 46 HCAPLUS COPYRIGHT 2002 ACS

1999:140948 HCAPLUS AN

130:215889 DN

Photosensitive lithographic printing plate having ΤI oil-desensitized and concaved surface edges

Aono, Koichiro IN

Fuji Photo Film Co., Ltd., Japan Jpn. Kokai Tokkyo Koho, 12 pp. PA

SO

CODEN: JKXXAF

DT Patent

LA Japanese IC

ICM G03F007-09

ICS G03F007-00

74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

KIND DATE APPLICATION NO. DATE PATENT NO.

```
Page 86
LEE 09/828075
                                           JP 1997-212201 19970806
                       A2
                            19990226
     JP 11052579
PΤ
     The photosensitive lithog. printing plate has a light-sensitive
AB
     layer on a hydrophilic aluminum support surface, wherein the surfaces of
     facing sides of the aluminum support are hydrophilisized and has concave
     shape and oil-desensitization treatment. The photosensitive
     lithog. printing plate generates little soiling on an image
     receiving sheet.
     photosensitive lithog printing plate aluminum support;
ST
     hydrophilisize concave oil desensitization soiling
     Lithographic plates
ΙT
        (photosensitive lithog. printing plate)
     89-25-8DP, 1-Phenyl-3-methyl-5-pyrazolone, readtion products with
ΙT
     hexafluorophosphate-exchanged p-diazodiphenylamine sulfate-formaldehyde
     copolymer 9070-36-4DP, p-Diazodiphenylamine splfate-
     paraformaldehyde copolymer, hexafluorophosphate-exchanged,
     reaction products with 1-phenyl-3-methyl-5-pyrazolone
                                                            19735-89-8DP,
     1-Phenyl-3-methyl-5-pyrazolone, reaction pfoduct with 4-Diazodiphenylamine
     sulfate-formaldehyde polymer
     RL: PNU (Preparation, unclassified); RCT/(Reactant); TEM (Technical or
     engineered material use); PREP (Preparation); RACT (Reactant or
     reagent); USES (Uses)
        (dye for light-sensitive layer of photosensitive
        lithog. printing plate)
                                              30525-89-4, Paraformaldehyde
     150-33-4, 4-Diazodiphenylamine sulfate
TΨ
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (dye for light-sensitive layer of photosensitive lithog.
        printing plate)
                                                                      9000-01-5,
                             7664-38-2/ Phosphoric acid, processes
     87-89-8, myo-Inositol
TΤ
     Gum arabic 50813-16-6, Sodium metaphosphate
     RL: PEP (Physical, engineering \phir chemical process); PROC (Process)
         (oil-desensitization treatme/ht for photosensitive lithog.
        printing plate)
     9070-36-4DP, p-Diazodiphenylamine sulfate-paraformaldehyde
ΙT
     copolymer, hexafluorophosphate-exchanged, reaction products with
      1-phenyl-3-methyl-5-pyrazolone
     RL: PNU (Preparation, unclassified); RCT (Reactant); TEM (Technical or
     engineered material use); PREP (Preparation); RACT (Reactant or
     reagent); USES (Uses)
         (dye for light-sensitive layer of photosensitive
        lithog. printing plate)
     9070-36-4 HCAPLUS
 RN
      Benzenediazonium, 4-(pplenylamino)-, sulfate (2:1), polymer with
      formaldehyde (9CI) (GA INDEX NAME)
      CM
           1
      CRN 50-00-0
      CMF C H2 O
 H_2C = 0
      CM
           2
      CRN 150-33-4
      CMF C12 H10 N3 . 1/2 O4 S
                3
           CM
```

LEE 09/828075 Page 87

> CRN 16072-57-4 CMF C12 H10 N3

CM

CRN 14808-79-8 CMF 04 S

L85 ANSWER 19 OF 46 HCAPLUS COPYRIGHT 2002 APS

1997:752715 HCAPLUS AN

128:55426 DN

Light-sensitive diazonium compound having both bisulfate and zincate parts for lithographic plate preparation

Deutsch, Albert S. IN

Precision Lithograining Corp., USA PΑ

U.S., 10 pp. SO CODEN: USXXAM

Patent DT

English LA

ICM G03F007-021 IC

NCL 430157000

74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

ΡŤ

APPLICATION NO. DATE PATENT NO. KIND DATÆ _____ US 1996-677438 19960702 1**\$**971118 A

A water-sol. diazonium dompd. made by the condensation of a diazoaryl amine and an aldehyde and pptd. to form a polymeric anionic species having both ZnCl42- and HSO4- moieties is disclosed. The compd. is blended with a suitable binder, a colorant, and other optional components to produce a light-sensitive coating compn. for the prepn. of a lithog. plate. When applied/to an aluminum substrate, the light-sensitive coating compn. forms a lithog. plate esp. stable for use under high humidity conditions. Such a light-sensitive coating compn. may be developed by eithe f water alone or by water contg. surfactants and/or a small amt. of an qrg. solvent.

photosensitive diazonium compd lithog plate; bisulfate zincate ST

diazonium compd /ithog plate

Photoimaging materials (contg. diazonium compds. having both bisulfate and zincate parts for

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

```
LEE 09/828075
                      Page 88
        lithog. plate prepn.)
     Diazonium compounds
     RL: TEM (Technical or engineered material use); USES (Uses)
        (having both bisulfate and zincate parts for photosensitive compns. for
        lithog. plate prepn.)
IT
     Lithographic plates
        (photosensitive compns. contg. diazonium compds. having both bisulfate
        and zincate parts for prepn. of)
     147-14-8, Copper phthalocyanine 24937-78-8, Ethylene-vinyl
TΨ
     acetate copolymer 37211-53-3, Triton CF-21
     RL: TEM (Technical or engineered material use); USES (Uses)
        (lithog. plate prepn. using photosensitive compns. contg.
        diazonium compds. and)
     41432-19-3DP, mixed salt with tetrachlorozincate
TΤ
     67290-46-4DP, mixed salt with bisulfate
     RL: SPN (Synthetic preparation); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (prepn. and use in photosensitive compns. for lithog. plate
        prepn.)
     50-00-0, Formaldehyde, reactions
                                        4477-28-5
ΙT
     RL: RCT (Reactant); TEM (Technical or engineered material use); RACT
     (Reactant or reagent); USES (Uses)
        (reaction in prepn. of diazonium compds. for use in photosensitive
        compns. for lithog. plate prepn.)
     41432-19-3DP, mixed salt with tetrachlorozincate
TΤ
     67290-46-4DP, mixed salt with bisulfate
     RL: SPN (Synthetic preparation); TEM (Technical or engineered
     material use); PREP (Preparation); USES (Uses)
        (prepn. and use in photosensitive compns. for lithog. plate
        prepn.)
     41432-19-3 HCAPLUS
RN
     Benzenediazonium, 4-(phenylamino)-, sulfate (1:1), polymer with
CN
     formaldehyde (9CI) (CA INDEX NAME)
     CM
      CRN 50-00-0
      CMF C H2 O
 H2C=0
           2
      CM
      CRN 4477-28-5
         C12 H10 N3 . H O4 S
      CMF
           CM
           CRN 16072-57-4
           CMF C12 H10 N3
```

Page 89

4 CM

14996-02-2 CRN CMF H O4 S

67290-46-4 HCAPLUS RN

Benzenediazonium, 4-(phenylamino)-, (T-4)-tetrachlorozincate(2-) (2:1), polymer with formaldehyde (9CI) (CA INDEX NAME)

CM

CRN 50-00-0 CMF C H2 O

 $H_2C = O$

2 CM

CRN 36505-86-9 CMF C12 H10 N3 . 1/2 C14 Zn

> 3 · CM

> > CRN 16072-57-4 CMF C12 H10 N3

CM

CRN 15201-05-5 CMF Cl4 Zn CCI CCS

```
L85 ANSWER 20 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1997:655780 HCAPLUS
AN
DN
     127:293898
     Structure and sorption properties of a polymer-colloid complex formed by
ΤI
     cetylpyridinium cations and polyacrylate anions
     Bobrov, A. B.; Skorikova, E. E.; Sul'Yanov, S. N.; Rogacheva, V. B.;
ΑIJ
     Zezin, A. B.; Kabanov, V. A.
     Shubnikov Inst. Crystallography, Moscow State Uniy., Moscow, 119899,
     Vysokomolekulyarnye Soedineniya, Seriya A i Ser/ya B (1997), 39(4),
SO
     CODEN: VSSBEE; ISSN: 1023-3091
     MAIK Nauka
PΒ
     Journal
DT
     Russian
LA
      36-5 (Physical Properties of Synthetic High Polymers)
      Section cross-reference(s): 38
     The structure and sorption properties of a stoichiometric polymer-colloid
AB
     complex, formed by polyacrylate anions and cetylpyridnium cations, were studied. The complex is essentially a cetylpyridnium salt of polyacrylic
     acid and comprises pos. charged surfactant lamellae linked by salt bonds to polyacrylate anions. The aliph, radicals of cetylpyridinium form a cryst. hexagonal packing inside the lamellae. The polymer-colloid complex
      has an amphiphilic character and/is capable of limited swelling in water
      and org. solvents of various polarities. The swelling in water and
      low-polarity org. substances is accompanied by strong modification of the
      complex structure. During the co-adsorption of satd. vapors of water and
      low-polarity org. substances,/the equil. degree of swelling was markedly
      greater as compared to that following from the additive model.
      lamellar cryst structure cetylpyridinium polyacrylate; sorption water
ST
      vapor cetylpyridinium polyacrylate; acetone sorption sodium polyacrylate
      cetylpyridinium complex; dioxane sorption sodium polyacrylate
      cetylpyridinium complex; chloroform sorption sodium polyacrylate
      cetylpyridinium complex; /toluene sorption sodium polyacrylate
      cetylpyridinium complex; heptane sorption sodium polyacrylate
      cetylpyridinium complex/ carbon tetrachloride sorption polyacrylate
      cetylpyridinium complex; diethyl ether sorption polyacrylate
      cetylpyridinium complex; sodium polyacrylate cetylpyridinium bromide
      complex structure
 IT
      Surfactants
          (amphiphilic; structure and sorption properties of a polymer-colloid
          complex formed by cetylpyridinium cations and polyacrylate anions)
 ΙT
      Polymer morphology/
          (lamellar; structure and sorption properties of a polymer-colloid
          complex formed by cetylpyridinium cations and polyacrylate anions)
 ΙT
      Solubility
       Sorption
       Swelling, physical
          (structure and sorption properties of a polymer-colloid complex formed
          by cetylpyridinium cations and polyacrylate anions)
```

LEE 09/828075 Page 91

Polyelectrolytes ΙT

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (structure and sorption properties of a polymer-colloid complex formed

by cetylpyridinium cations and polyacrylate anions)

56-23-5, Carbon tetrachloride, processes 60-29-7, Diethyl ether, IT 67-64-1, 2-Propanone, processes 67-66-3, Chloroform, processes 108-88-3, Toluene, processes 123-91-1, 1,4-Dioxane, 142-82-5, Heptane, processes 7732-18-5, Water, processes processes 7732-18-5, Water, processes processes RL: PEP (Physical, engineering or chemical process); PROC (Process) (sorbate; structure and sorption properties of a polymer-colloid complex formed by cetylpyridinium cations and polyacrylate anions)

197250-22-9P ΙT

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(structure and sorption properties of a polymer-colloid complex formed by cetylpyridinium cations and polyacrylate anions)

140-72-7, Cetylpyridinium bromide 9003-04-7, Polyacrylic acid, ΙT

sodium salt

RL: RCT (Reactant); RACT (Reactant or reagent) (structure and sorption properties of a polymer-colloid complex formed by cetylpyridinium cations and polyacrylate anions)

197250-22-9P TΤ

RL: PRP (Properties); SPN (Synthetic preparation); PREP

(Preparation)

(structure and sorption properties of a polymer-colloid complex formed by cetylpyridinium cations and polyacrylate anions)

RN

197250-22-9 HCAPLUS
Pyridinium, 1-hexadecyl-, bromide, compd. with 2-propenoic acid CN homopolymer sodium salt (9CI) (CA INDEX NAME)

1 CM

CRN 140-72-7 CMF C21 H38 N . Br



Br⁻

2 CM

CRN 9003-04-7 (C3 H4 O2)x . x NaCMF

CM

9003-01-4 CRN (C3 H4 O2)xCMF CCI PMS

Page 92 LEE 09/828075

CM

CRN 79-10-7 CMF C3 H4 O2

0 HO-C-CH=CH2

140-72-7, Cetylpyridinium bromide

RL: RCT (Reactant); RACT (Reactant or reagent)

(structure and sorption properties of a polymer-colloid complex formed by cetylpyridinium cations and polyacrylate anions)

140-72-7 HCAPLUS RN

Pyridinium, 1-hexadecyl-, bromide (8CI, 9CI) (CA INDEX NAME) CN



● Br-

L85 ANSWER 21 OF 46 HCAPLUS COPYRIGHT 2002 ACS

1997:509311 HCAPLUS AN

127:206938

DN Oxidatively polymerizable coloring materials and their nonreversible color ΤI change by polymerization

Yanagi, Masato; Ishiguro, Hideyuki; Sato, Keiichi; Kamimura, Toshifumi IN

Toyo Ink Mfg. Co., Ltd., Japan PΑ

Jpn. Kokai Tokkyo Koho, 8 pp. SO CODEN: JKXXAF

Patent DT

LA Japanese

IC

ICM C09B021-00 ICS B32B027-18; B41M005-132; C07D265-38; C09B019-00; C09B069-10; C09D011-16

41-5 (Dyes, Organic Pigments, Fluorescent Brighteners, and Photographic CC Sensitizers)

Section cross-reference(s): 35, 42, 74

FAN.	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
	JP 09194747 A2	A2 B2	19970729 20010625	JP 1996-7968	19960122

GΙ

The materials, useful for writing tools, pressure-sensitive copying paper, AB recording paper, or toys, comprise (A) layers contg. oxidatively polymerizable colorants I [R1-R7 = H, ha/co, C1-8 alkyl, C1-8 alkoxy, (un) substituted (hetero) aryl, OH, NH2, ON, CHO, CO2H, NO2, NO, or 2 groups may join to form arom. rings; X = S, O/Se, Te] on supports and (B) oxidizing agents kept out of contact with A. The color of the materials is changed by polymn. of A in contact/ with B. Thus, 100 parts Et acrylate-Me acrylate copolymer was mylxed with 12 parts 10% 2-methoxyphenothiazine in EtOAc, coated on a PET release film, and dried to form a sheet. Sep., a felt was/impregnated with a 100:2 mixt. of MeOH and CuCl2 to obtain a marking pen / which was used in writing on the sheet to show a green color initially, which then changed to blue, purple, and brown with passage of time.

oxidative polymn phenothiazine color change; writing tool phenothiazine ST color change; toy phenothiazing oxidative polymn color change; recording paper phenothiazine color change; pressure sensitive copying paper

phenothiazine

IT Quinones

RL: CAT (Catalyst use); USES/(Uses) (naphthoquinones, oxidiz#ng agents; oxidative polymn. of phenothiazines for nonreversible color khange)

Peroxides, uses TΤ

RL: CAT (Catalyst use); USES (Uses) (org., oxidizing agents; oxidative polymn. of phenothiazines for nonreversible color change)

TΤ Color

Oxidizing agents

Pens

(oxidative polymn. $q^{\prime}f$ phenothiazines for nonreversible color change)

Polymerization TΤ

Polymerization catalysts

(oxidative; oxidative polymn. of phenothiazines for nonreversible color change)

Halogens TΨ

Oxides (inorganic), uses

Peroxy acids

Peroxysulfates

Quinones

Salts, uses

RL: CAT (Catalyst use); USES (Uses)

(oxidizing agents; oxidative polymn. of phenothiazines for nonreversible color change)

Printing (nonimpact) IT

(paper; oxidative polymn. of phenothiazines for nonreversible color change)

Copying paper ΙT

(pressure-sensitive; oxidative polymn. of phenothiazines for nonreversible color change)

IT

Page 94

75788-67-9P 113254-03-8P **194475-58-6P** IT 194475-59-7P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(oxidative polymn. of phenothiazines for nonreversible color change)

92-30-8, 2-(Trifluoromethyl)phenothiazine 92-84-2, Phenothiazine IT 135-67-1, Phenoxazine 1771-18-2, 2-Methoxyphenothiazine

RL: TEM (Technical or engineered material use); USES (Uses)

(oxidative polymn. of phenothiazines for nonreversible color change) 67-68-5, Dimethyl sulfoxide, uses 84-58-2, 2,3-Dichloro-5,6-dicyano-1,4benzoquinone 94-36-0, Benzoyl peroxide, uses 614-45-9, tert-Butyl

peroxybenzoate 7447-39-4, Copper dichloride, uses

RL: CAT (Catalyst use); USES (Uses)

(oxidizing agents; oxidative polymn. of phenothiazines for nonreversible color change)

75788-67-9P 194475-58-6P 194475-59-7P ΙT

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(oxidative polymn. of phenothiazines for nonreversible color change)

75788-67-9 HCAPLUS RN

10H-Phenothiazine, homopolymer (9CI) (CA INDEX NAME) CN

1 CM

CRN 92-84-2 CMF C12 H9 N S

194475-58-6 HCAPLUS RN

10H-Phenothiazine, 2-methoxy-, homopolymer (9CI) (CA INDEX NAME) CN

CM1

CRN 1771-18-2 CMF C13 H11 N O S

194475-59-7 HCAPLUS RN

10H-Phenothiazine, 2-(trifluoromethyl)-, homopolymer (9CI) (CA INDEX CN NAME)

CM 1

CRN 92-30-8

CMF C13 H8 F3 N S

```
L85 ANSWER 22 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1996:660714 HCAPLUS
ΑN
     125:290781
DN
     Manufacture of printed circuit boards by electrophotography
TТ
     Inoe, Wakana; Hyodo, Kenji
ΙN
     Mitsubishi Paper Mills Ltd, Japan
PΑ
     Jpn. Kokai Tokkyo Koho, 9 pp.
SO
     CODEN: JKXXAF
DT .
     Patent
     Japanese
LA
     ICM G03G005-07
TC
     ICS G03G005-06; H05K001-09; H05K003-06; H05K003-/24
     76-3 (Electric Phenomena)
     Section cross-reference(s): 38, 74
FAN.CNT 1
                                            APPLICATION NO.
                                                             DATE
                      KIND DATE
     PATENT NO.
                                           _____
                     ----
                     A2 19960820
                                            JP 1995-18233
                                                             19950206
     JP 08211639
PI
     The manuf. of a printed circuit comprises formation of
AΒ
     (A) a photoconductor layer on an elec. conductor layer laminated on an
     elec. insulating board, (B) formation of a toner image on the
     photoconductor layer by electrophotog., (C) eluting the photoconductor
     layer area without the image, (D) etching the elec. conductor layer of the
     eluted photoconductor area, and (E) removing the remaining photoconductor
     layer and a toner image as require and the manuf. is characterized by the
     use of a photoconductive resin contg. /(a) 10-90% a polymerizable monomer
     having a carbazole group and/or a polymerizable monomer having a
     phthalocyanine group as a polymerizable component and (b) 10-40%
     polymerizable monomers having an antonic functional group for the
     photoconductor layer.
     carbazole polymer photoconductor printed circuit; phthalocyanine
ST
     polymer photoconductor printed ci/cuit; reversal development toner image
     printed circuit; electrodepositiøn photoconductor printed circuit
      Electrophotographic photoconductors and photoreceptors
 ΙT
         (manuf. of printed circuit boards by electrophotog.)
      Electrophotographic development
 ΙT
         (reversal; manuf. of printed circuit boards by electrophotog.)
      Electric circuits
 ΙT
         (printed, boards, manuf. \phif printed circuit boards by electrophotog.)
      182865-04-9P, Acrylic acid-butyl methacrylate-9-vinylcarbazole copolymer
 ΙT
      182865-05-0P, Butyl methacrylate-methacrylic acid-9-vinylcarbazole
                182865-06-1P, Acrylic acid-butyl methacrylate-4-nitro-9-
      copolymer
      vinylcarbazole copolymer 182865-07-2P, Cobalt
      phthalocyaninetetracarboxylic acid tetraester with ethylene
      glycol, acrylate, polymer/with 9-vinylcarbazole, methacrylic acid, butyl
      methacrylate, and butyl acrylate 182865-08-3P, Cobalt phthalocyaninetetracarboxylic acid tetraester with ethylene
      glycol, acrylate, polymer with methacrylic acid, butyl methacrylate, and
      butyl acrylate 182865-09-4P, Cobalt
      phthalocyaninetetracarboxylic acid tetraester with ethylene
      glycol, acrylate, polymer with acrylic acid, butyl methacrylate, and butyl
```

LEE 09/828075 Page 96

> 182865-10-7P, Butyl methacrylate-methacrylic acrylate acid-4-nitro-9-vinylcarbazole copolymer RL: PEP (Physical, engineering or chemical process); PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)

(manuf. of printed circuit boards by electrophotog.) 182865-07-2P, Cobalt phthalocyaninetetracarboxylic acid tetraester with ethylene glycol, acrylate, polymer with 9-vinylcarbazole, methacrylic acid, butyl methacrylate, and butyl acrylate 182865-08-3P, Cobalt phthalocyaninetetracarboxylic acid tetraester with ethylene glycol, acrylate, polymer with methacrylic acid, butyl methacrylate, and butyl acrylate 182865-09-4P, Cobalt phthalocyaninetetracarboxylic acid tetraester with ethylene glycol, acrylate, polymer with acrylic acid, butyl methacrylate, and butyl acrylate RL: PEP (Physical, engineering or chemical process); PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP

(Preparation); PROC (Process); USES (Uses) (manuf. of printed circuit boards by electrophotog.)

182865-07-2 HCAPLUS

Cobalt, [tetrakis[2-[(1-oxo-2-propenyl)oxy]ethyl] 29H,31H-phthalocyanine-C,C,C,C-tetracarboxylato(2-)-N29,N30,N31,N32]-, polymer with butyl 2-methyl-2-propenoate, butyl 2-propenoate, 9-ethenyl-9H-carbazole and 2-methyl-2-propenoic acid (9CI) (CA INDEX NAME)

CM

IT

CN

CRN 101240-49-7 CMF C56 H40 Co N8 O16 CCI CCS, IDS

CM 2

CRN 1484-13-5 CMF C14 H11 N

Page 97

CM 3

CRN 141-32-2 CMF C7 H12 O2

$$\begin{array}{c} & \text{O} \\ || \\ \text{n-BuO-C-CH----} \text{CH-----} \text{CH}_2 \end{array}$$

CM

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c} \text{O} \quad \text{CH}_2 \\ || \quad || \\ \text{n-BuO-C-C-Me} \end{array}$$

CM 5

CRN 79-41-4 CMF C4 H6 O2

RN 182865-08-3 HCAPLUS
CN Cobalt, [tetrakis[2-[(1-oxo-2-propenyl)oxy]ethyl] 29H,31H-phthalocyanineC,C,C,C-tetracarboxylato(2-)-N29,N30,N31,N32]-, polymer with butyl
2-methyl-2-propenoate, butyl 2-propenoate and 2-methyl-2-propenoic acid
(9CI) (CA INDEX NAME)

CM 1

CRN 101240-49-7 CMF C56 H40 Co N8 016 CCI CCS, IDS

CM

CRN 141-32-2 CMF C7 H12 O2

$$\overset{\text{O}}{\underset{\text{n-BuO-C-CH}}{\parallel}}\text{CH}_2$$

CM 3

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c|c} \text{O} & \text{CH}_2 \\ \parallel & \parallel \\ \text{n-BuO-C-C-Me} \end{array}$$

CM

CRN 79-41-4 CMF C4 H6 O2

Page 99

182865-09-4 HCAPLUS RN

Cobalt, [tetrakis[2-[(1-oxo-2-propenyl)oxy]ethyl] 29H,31H-phthalocyanine-CN C,C,C,C-tetracarboxylato(2-)-N29,N30,N31,N32]-, polymer with butyl 2-methyl-2-propenoate, butyl 2-propenoate and 2-propenoic acid (9CI) (CA INDEX NAME)

CM1

CRN 101240-49-7

CMF C56 H40 Co N8 O16

CCI CCS, IDS

2 CM

CRN 141-32-2 CMF C7 H12 O2

$${\overset{\circ}{\underset{n-\text{BuO}-C-CH}{\parallel}}} \text{CH}_2$$

CM 3

CRN 97-88-1 CMF C8 H14 O2

$$\begin{array}{c} \text{O} \quad \text{CH}_2 \\ || \quad || \\ \text{n-BuO-C-C-Me} \end{array}$$

LEE 09/828075 Page 100

CM 4

CRN 79-10-7

CMF C3 H4 O2

O || HO- C- CH == CH₂

```
L85 ANSWER 23 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1996:302911 HCAPLUS
AN
     125:45126
DN
     Photosensitive resin composition
TΙ
     Gybin, Alexander S.; Van Iseghem, Lawrence/C.
ΤN
     Chromaline Corp., USA
PA
     U.S., 11 pp., Division of U. S. Ser. No 28, 420.
SO
     CODEN: USXXAM
     Patent
DT
     English
T.A
     ICM G03C001-73
TC
     74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
NCL 430287000
     Reprographic Processes)
 FAN.CNT 2
                                             APPLICATION NO. DATE
                        KIND DATE
      PATENT NO.
                                             US 1994-195258
                                                               19940214
                              199604/09
                         Α
      US 5506089
 PΙ
                                                               19930309
                                             US 1993-28420
                              20000/201
      US 6020436
                         Α
                              1993/0309
 PRAI US 1993-28420
      For diagram(s), see print d CA Issue.
      A universal method to make photosensitive polymers from poly(vinyl alc.)
 GI
      and poly(vinyl pyridine) is disclosed which generally does not
 AR
      require final purifn. and can produce photosensitive polymers which are of
      similar photosensitivity whether prepd. with a poly(vinyl alc.) or
      poly(vinyl pyridine) backbone. These polymers comprise a heterocyclic, light-sensitive pendant group including a moiety having the
      formula I or II wherein Z1 denotes the atoms necessary to form a
      substituted or unsubstituted arom. heterocyclic ring; Z2 denotes the atoms
      necessary to form/a substituted or unsubstituted arom. or arom.
      heterocyclic ring; R is hydrogen or a substituted or unsubstituted alkyl
      group; Y is a residue from a grafting group that is capable of grafting
      the pendant group onto a polymeric backbone; n is 1 or 2.
      photoresist scheen printing styrylpyridinium polymer
 ST
          (photosen itive compns. contg. styrylpyridinium group-contg.
       Stencils
          polymers /for)
          (photo-, contg. styrylpyridinium group-contg. polymers for
       Resists
  ΙT
          screen printing)
       Printing plates
          (screen, photosensitive compns. contg. styrylpyridinium
  IT
          group-contg. polymers for)
       RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
  IT
        (Reactant or reagent)
```

Page 101

(9; prepn. and reaction in prepg. styrylpyridinium group-contg. polymers for photosensitive compns.) 718-24-1P 722-21-4P 4945-26-0P, 2-Styrylquinoline TT 178058-95-2P 178058-93-0P 178058-94-1P 178058-91-8P 178058-90-7P 178059-01-3P 178059-00-2P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (prepn. and reaction in prepg. styrylpyridinium group-contg. polymers for photosensitive compns.) 178058-96-3P 178058-97-4P 178058-99-6P ΙT 178059-02-4P RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (prepn. and use in photosensitive compns.) 110-52-1, 1,4-Dibromobutane 103-31-1, 4-Styrylpyridine ΙT 629-03-8, 1,6-Dibromohexane RL: RCT (Reactant); RACT (Reactant or reagent) (reaction in prepg. styrylpyridinium group-contg. polymers for photosensitive compns.) 178058-96-3P 178058-97-4P 178058-99-6P ΙT 178059-02-4P RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (prepn. and use in photosensitive compns.) 178058-96-3 HCAPLUS RN Pyridinium, 1-[2-(1,3-dioxan-2-yl)ethyl]-4-(2-phenylethenyl)-, bromide, CN polymer with ethenyl acetate (9CI) (CA INDEX NAME)

CRN 178058-92-9 CMF C19 H22 N O2 . Br

• Br-

CM 2

CM

Ph-CH-CH

1

CRN 108-05-4 CMF C4 H6 O2

AcO-CH-CH2

RN 178058-97-4 HCAPLUS CN Pyridinium, 1-[2-(1,3-dioxan-2-yl)ethyl]-4-(2-phenylethenyl)-, bromide, polymer with ethenol (9CI) (CA INDEX NAME)

- CM 1

Page 102

CRN 178058-92-9 CMF C19 H22 N O2 . Br

• Br-

CM 2

CRN 557-75-5 CMF C2 H4 O

 $_{\rm H2C}$ = $_{\rm CH}$ - $_{\rm OH}$

RN 178058-99-6 HCAPLUS
CN Quinolinium, 1-[2-(1,3-dioxan-2-yl)ethyl]-4-(2-phenylethenyl)-, bromide, polymer with ethenyl acetate (9CI) (CA INDEX NAME)

CM 1

CRN 178058-98-5 CMF C23 H24 N O2 . Br

• Br-

CM 2

Page 103

CRN 108-05-4 CMF C4 H6 O2

 $AcO-CH=CH_2$

178059-02-4 HCAPLUS

Pyridinium, 1-(4-bromobutyl)-4-(2-phenylethenyl)-, bromide, polymer with CN 4-ethenylpyridine (9CI) (CA INDEX NAME)

CM

CRN 178059-01-3 CMF C17 H19 Br N . Br

• Br

CM 2

CRN 100-43-6 CMF C7 H7 N

L85 ANSWER 24 OF 46 HCAPLUS COPYRIGHT 2002 ACS

1995:1005423 HCAPLUS AN

124:41190 DN

Spatial Electrochromism in Metallopolymeric Films of Ruthenium Polypyridyl ΤI Complexes

Leasure, Robert M.; Ou, Wei; Moss, John A.; Linton, Richard W.; Meyer, ΑU

Thomas J. Department of Chemistry, University of North Carolina, Chapel Hill, NC, CS NORTH CAROLINA, USA

Chemistry of Materials (1996), 8(1), 264-73 SO CODEN: CMATEX; ISSN: 0897-4756

American Chemical Society PΒ

Journal DT

English LA

74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other CC Reprographic Processes)

```
Page 104
LEE 09/828075
     Section cross-reference(s): 73
     Thin films of poly[Ru(vbpy)2(py)2]2+ (vbpy/is 4-vinyl-4'-methyl-2,2'-
AB
     bipyridine; py is pyridine) have been depøsited on
     electrodes by reductive electropolymn. Photolysis of the film in the presence of chloride ion leads to photochem. loss of the pyridine
     ligands and sequential formation of polyg[Ru(vbpy)2(py)Cl]+ and
     poly[Ru(vbpy)2Cl2], as detd. by cyclic voltammetry and FTIR spectroscopy.
     Contact lithog. was used to control the photosubstitution
     process spatially and form laterally/resolved, bicomponent films with
     image resoln. below 10 .mu.m. Smal\vec{I} spot XPS was used to confirm that the
     photolyzed and nonphotolyzed region's of an imaged film were chem.
     distinct. Dramatic changes occur/in the absorption spectra and redox
     potentials of the ruthenium complexes upon substitution of chloride for the pyridine ligands. This provides a basis for the fabrication
     of bicomponent, electrochromic f ilm assemblies on optically transparent
     electrodes of tin doped indium/oxide on glass. The spectroelectrochem.
     response of these films has been studied by slow scan cyclic voltammetry
     and potential step chronoamperometry.
     electrochromism ruthenium bipyridine complex polymer
     lithog; photolysis electropplymerized ruthenium bipyridine
     vinylmethylbipyridine lithøg; photolithog imaging
     ruthenium bipyridine viny/methylbipyridine polymer
     Ultraviolet and visible spectra
(photolithog. imaging of poly[bis(vinylmethylbipyridine)bis(
         bipyridine) ruthenium / (2+) film in presence of chloride for
         electrochromic devide fabrication)
      Photolysis
IT
         (photolysis and lithog. imaging of poly[bis(
         vinylmethylbipyridine) bis (bipyridine) ruthenium] (2+)
         film in presence pf chloride for electrochromic device fabrication)
 IT
         (photolysis of phly[bis(vinylmethylbipyridine)bis(
         bipyridine)ruthenium](2+) film in presence of chloride for
         electrochromic Hevice fabrication)
      Electrochromism
 IT
         (spatial; of products of photolysis and lithog. photoimaging
         of poly[bis(vinylmethylbipyridine)bis(bipyridine
         )ruthenium](2/+) film in presence of chloride)
      Optical imaging/devices
 IT
          (electrochromic, photolithog. imaging of poly[bis(
         vinylmethylbipyridine) bis (bipyridine) ruthenium] (2+)
         film in presence of chloride for electrochromic device fabrication)
      Lithography
 IT
          (photo-, photolysis and lithog. imaging of poly[bis(
         vinylmethylbipyridine) bis (bipyridine) ruthenium] (2+)
         film in presence of chloride for electrochromic device fabrication)
      Substitution reaction
 ΙT
          (photochem., in thin electropolymd. films of bis(
          vinylmethylbipyridine)bis(bipyridine)ruthenium(2+)
          film in presence of chloride for electrochromic device fabrication)
      Electric potential
          (redox, of poly[bis(vinylmethylbipyridine)bis(
         bipyridine) ruthenium] (2+) and its photoproducts produced in
          presence of chloride)
                    171899-54-0
       171899-53-9
      RL: PRP (Properties)
          (elec. potential of)
       171899-56-2P
 IT
       RL: PNU (Preparation, unclassified); PRP (Properties); PREP (Preparation)
          (electronic absorption spectra of)
```

```
Page 105
LEE 09/828075
     124815-46-9
     RL: PRP (Properties)
        (electronic absorption spectra of)
     104704-09-8, 4-Formyl-4'-methyl-2,2'-bipyridine
IT
     RL: RCT (Reactant); RACT (Reactant or reagent)
        (in prepn. of vinylmethylbipyridine)
                                171899-60-8
     171899-51-7
                  171899-52-8
ΙT
     RL: FMU (Formation, unclassified); PEP (Physical, engineering or chemical
     process); PRP (Properties); FORM (Formation, nonpreparative); PROC
     (Process)
        (photolysis and lithog. imaging of poly[bis(
        vinylmethylbipyridine)bis(bipyridine)ruthenium](2+)
        film in presence of chloride for electrochromic device fabrication)
     161272-31-7P 171899-58-4P
     RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT
     (Reactant); SPN (Synthetic preparation); PREP
     (Preparation); PROC (Process); RACT (Reactant or reagent)
         (photolysis and lithog. imaging of poly[bis(
        vinylmethylbipyridine)bis(bipyridine)ruthenium](2+)
        film in presence of chloride for electrochromic device fabrication)
     1112-67-0, Tetrabutylammonium chloride
TΤ
     RL: RCT (Reactant); RACT (Reactant or reagent)
         (photolysis and lithog. imaging of poly[bis(
        vinylmethylbipyridine) bis (bipyridine) ruthenium] (2+)
        film in presence of chloride for electrochromic device fabrication)
     80864-60-4P
ΙT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
      (Reactant or reagent)
         (reaction with pyridine)
     74173-48-1P, 4-Methyl-4'-vinyl-2,2'-bipyridine
ΙT
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
      (Reactant or reagent)
         (reaction with ruthenium chloride)
     171899-49-3P
 TΨ
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT
      (Reactant or reagent)
         (reductive electropolymn.)
      161272-31-7P 171899-58-4P
 ΙT
      RL: PEP (Physical, engineering or chemical process); PRP (Properties); RCT
      (Reactant); SPN (Synthetic preparation); PREP
      (Preparation); PROC (Process); RACT (Reactant or reagent)
         (photolysis and lithog. imaging of poly[bis(
         vinylmethylbipyridine)bis(bipyridine)ruthenium](2+)
         film in presence of chloride for electrochromic device fabrication)
      161272-31-7 HCAPLUS
 RN
      Ruthenium(2+), bis(4-ethenyl-4'-methyl-2,2'-bipyridine-
 CN
      .kappa.N1, .kappa.N1')bis(pyridine)-, bis[hexafluorophosphate(1-)],
      homopolymer (9CI) (CA INDEX NAME)
      CM
      CRN 153824-72-7
      CMF C36 H34 N6 Ru
      CCI CCS
```

Page 106

CM 2

CRN 16919-18-9

CMF F6 P

CCI CCS

171899-58-4 HCAPLUS RN

Ruthenium(2+), bis(4-ethenyl-4'-methyl-2,2'-bipyridine-N,N')bis(pyridine)-, dichloride, homopolymer (9CI) (CA INDEX NAME) CN

1 CM

CRN 171899-57-3

CMF C36 H34 N6 Ru . 2 Cl

CCI CCS

●2 C1⁻

L85 ANSWER 25 OF 46 HCAPLUS COPYRIGHT 2002

ΑN 1995:863255 HCAPLUS

123:325613 DN

Photolithographically defined electropolymerized films. Fabrication of an electrochemically switchable diffraction grating comprised of poly-(bpy) 2Ru(vpy) 22+

ΑU

Hauser, Brian T.; Bergstedt, Troy S.; Schanze, Kirk S. Dep. Chem., Univ. Florida, Gainesville, FL, 32611-7200, USA CS

Journal of the Chemical Society, Chemical Communications (1995), (19), SO 1945-6 CODEN: JCCCAT; ISSN: 0022-4936

Royal Society of Chemistry PB

DT Journal

LA English

74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other CC Reprographic Processes),

A combination of photo/ithog. and electropolymn. is utilized to fabricate AB spatially patterned films of poly-(bpy) 2Ru(vpy) 22+; the method has been applied to fabricate/poly-(bpy)2Ru(vpy)22+ based diffraction gratings.

photolithog electropolymn ruthenium pyridine polymer photoresist ST

Diffraction gratings IT (photolithog. and electropolymn. to fabricate spatially patterned films of ruthenium-pyridine polymer)

IT Polymerization

(electrochem/, photolithog. and electropolymn. to fabricate spatially patterned films of ruthenium-pyridine polymer)

ΙT Lithography (photo-, photolithog. and electropolymn. to fabricate spatially patterned films of ruthenium-pyridine polymer)

88670-65-9P/ RL: PEP (Physical, engineering or chemical process); PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses) (photolithog. and electropolymn. to fabricate spatially patterned films

Page 108

of ruthenium-pyridine polymer)

3109-63-5, Tetrabutylammonium hexafluorophosphate 79813-96-0 ΙT

RL: RCT (Reactant); RACT (Reactant or reagent)

(photolithog. and electropolymn. to fabricate spatially patterned films of ruthenium-pyridine polymer)

146088-00-8, Poly(methylphenyl 76188-55-1, Poly(methylphenyl silane) ΙT silane)

RL: TEM (Technical or engineered material use); USES (Uses) (photolithog. and electropolymn. to fabricate spatially patterned films of ruthenium-pyridine polymer)

88670-65-9P IT

RL: PEP (Physical, engineering or chemical process); PNU (Preparation, unclassified); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)

(photolithog. and electropolymn. to fabricate spatially patterned films of ruthenium-pyridine polymer)

88670-65-9 HCAPLUS RN

Ruthenium(2+), bis(2,2'-bipyridine-.kappa.N1,.kappa.N1')bis(4-CN ethenylpyridine)-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 75687-40-0 CMF C34 H30 N6 Ru CCI CCS

$$_{\rm H_2C}$$
 CH

L85 ANSWER 26 OF 46 HCAPLUS COPYRIGHT 2002 ACS

1995:839110 HCAPLUS AN

123:301601 DN

Preparation of 1,2-naphthoquinonediazide-4-sulfonic acid ester compound ΤI for presensitized lithographic plates

Sasaki, Mitsuru; Oota, Katsuko; Matsuo, Fumyuki; Matsubara, Shinichi; IN Tono, Katsuhiko

Mitsubishi Kagaku Kk, Japan; Konishiroku Photo Ind PΑ

Jpn. Kokai Tokkyo Koho, 5 pp. SO CODEN: JKXXAF

Patent DT

LA Japanese

ICM C08G008-28 IC

Page 110

2 CM

38333-84-5 CRN

(C6 H6 O3 . C3 H6 O)x CMF

CCI PMS

> 3 CM

CRN 87-66-1 CMF C6 H6 O3

OH OH

CM

67-64-1 CRN CMF СЗ Н6 О

H3C-C-CH3

L85 ANSWER 27 OF 46 HCAPLUS COPYRIGHT 2002/ACS

1995:725237 HCAPLUS AN

123:164528

Reduction of thiazine dyes by bovine culmonary arterial endothelial cells ΤI in culture

Bongard, Robert D.; Merker, Marilyn P.; Shundo, Ryushi; Okamoto, ΑU Yoshiyuki; Roerig, David L.; Linehan, John H.; Dawson, Christopher A.

Dep. Physiol., Med. Coll. Wisconsin, Milwaukee, WI, 53226, USA CS American Journal of Physiology (1995), 269(1, Pt. 1), L78-L84 SO CODEN: AJPHAP; ISSN: 0002-9513

American Physiological Society PΒ

DTJournal

English LA

9-11 (Biochemical Methods) CC Section cross-reference(s)/ 13

The uptake of methylene blue (MB), and toluidine blue O (TBO) by bovine pulmonary arterial endothelial cells grown on microcarrier beads was detected as a decrease in the concn. of dye in the medium after these thiazine dyes were added to the medium surrounding the cells. Because the reduced forms of these dyes are much more lipophilic than the oxidized forms, the authors considered the possibility that redn. of the dyes at the cell surface might have preceded the uptake by the cells. Therefore, the authors studied the ability of the cells to reduce a toluidine blue O-polyacrylamide polymer (TBOP), which was too large to enter the cells in either the oxidized or reduced form. The TBO moieties of the polymer were

reduced by the cells, indicating that the dyes did not have to enter the cells to be reduced and that redn. can occur at, or near, the cell surface. The rate of TBOP redn. was about the same as the rate of uptake of the monomeric dyes, indicating that the cell surface redn. mechanism had a sufficient capacity to account for the monomer uptake by the cells. The authors also found that ferricyanide ion, which also did not permeate the cells, was reduced by the cells and that external ferricyanide inhibited the monomeric MB uptake. Thus the results with ferricyanide were also consistent with the concept that the monomeric thiazine dyes are reduced at the cell surface before the more lipophilic reduced forms are taken up by the endothelial cells.

pulmonary artery endothelial cell dye redn; thiazine dye redn endothelium ST cell culture

Animal tissue culture TΤ

Cell membrane

Reduction

(redn. of thiazine dyes by bovine pulmonary arterial endothelial cells in culture)

TT Dyes

(thiazine; redn. of thiazine dyes by bovine pulmonary arterial endothelial cells in culture)

Biological transport ΙT

(absorption, redn. of thiazine dyes by bovine pulmonary arterial endothelial cells in culture)

IT Artery

(pulmonary, endothelium, redn. of thiazine dyes by bovine pulmonary arterial endothelial cells in culture)

92-31-9, Toluidine blue O 13408-62-3, 61-73-4, Methylene blue ፐጥ

Ferricyanide

RL: RCT (Reactant); RACT (Reactant or reagent) (redn. of thiazine dyes by bovine pulmonary arterial endothelial cells in culture)

167498-53-5P TΤ

RL: RCT (Reactant); SPN (Synthetic preparation); PREP

(Preparation); RACT (Reactant or reagent)

(redn. of thiazine dyes by bovine pulmonary arterial endothelial cells in culture)

61-73-4, Methylene blue

RL: RCT (Reactant); RACT (Reactant or reagent) (redn. of thiazine dyes by bovine pulmonary arterial endothelial cells in culture)

61-73-4 HCAPLUS RN

Phenothiazin-5-ium, 3,7-bis(dimethylamino)-, chloride (9CI) (CA INDEX CN NAME)

Cl-

167498-53-5P RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

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Page 112
LEE 09/828075
```

(redn. of thiazine dyes by bovine pulmonary arterial endothelial cells in culture)

167498-53-5 HCAPLUS RN

Phenothiazin-5-ium, 7-(dimethylamino)-2-methyl-3-[(2-methyl-1-oxo-2propenyl)amino]-, chloride, polymer with 2-propenamide (9CI) (CA INDEX NAME)

1 CM

CRN 79-06-1 CMF C3 H5 N O

CM 2

CRN 61-73-4 CMF C16 H18 N3 S . Cl

C1-

L85 ANSWER 28 OF 46 HCAPLUS COPYRIGHT 2002

1995:320946 HCAPLUS AN

DN

Spectroelectrochemical characterization of ultra-thin films formed by electropolymerization of phenothiazine derivatives on transparent gold electrodes

Schlereth, Daniela D.; Schuhmann, Wolfgang; Schmidt, Hanns-Ludwig AU .

Lehrstuhl fuer Allgemeine Chemie und Biochemie, Technische Universitaet Muenchen, Voettingerstrasse 40, Freising-Weihenstephan, D-85350, Germany

Journal of Electroanalytical Chemistry (1995), 381(1-2), 63-70 SO CODEN: JECHES; ISSN: 0368-1874/

Elsevier PΒ

DT Journal

English LA

72-2 (Electrochemistry) CC

Section cross-reference(s)/: 35, 36 Transparent gold electrodes covered with an ultra-thin polyphenothiazine film layer were obtained/by electropolymn. of the monomeric parent AΒ compds., Methylene Blue / Methylene Green and Azur A, under alk. pH conditions. The electrochem. behavior of the polymers immobilized on the electrode surface was characterized by UV-visible difference spectroelectrochem. % study of the dependence of the oxidn. state of the surface redox species with the applied potential of any of the different

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

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polymer layers immobilized on the electrode surface showed one rather
    irreversible redox process, at potentials shifted towards 100-120 mV more
    pos. values than those obtained for the monomer in soln. The
    spectroelectrochem. curves, obtained from the absorbance changes obsd. at
    different applied potentials, showed a hysteresis in the oxidn. process,
    as well as the loss of some electroactive material from the electrode
    surface during the redn. process for all polymers studied. For
    poly-Methylene Blue, a shift of the redox potential towards more neg.
    values coupled with an increase in the irreversibility of the redox
    process at increasingly higher pH values was obsd. The simultaneously
    obtained UV-visible difference spectra of the different films resembled
    those of the solubilized monomers. However, the spectra of the polymers
     showed a shift of the main visible max. towards 50-80 nm shorter
    wavelengths, depending on the monomer studied.
     electropolymn phenothiazine deriv transparent gold electrode;
ST
     polyphenothiazine redox potential UV visible spectra
     Ultraviolet and visible spectra
IT
        (of phenothiazine derivs. and their polymers)
     Polymerization
        (electrochem., of phenothiazine derivs. on gold in alk. solns.)
ΙT
     Redox reaction
ΙT
        (electrochem., of polyphenothiazine derivs.)
     Electric potential
IT
        (redox, of polyphenothiazine derivs.)
     7631-99-4, Sodium nitrate, uses 7647-01-0, Hydrochloric acid, uses
     7758-11-4, Dipotassium phosphate 10043-35-3, Boric acid h3bo3, uses
IT
     RL: NUU (Other use, unclassified); PRP (Properties); USES (Uses)
        (electrochem. polymn. of phenothiazine derivs. on gold in soln. contg.)
     143629-25-8P, Poly(Azur A) 150645-86-6P, Poly(methylene
ΙT
     blue) 161201-31-6P, Poly(methylene green)
     RL: PNU (Preparation, unclassified); PRP (Properties); RCT (Reactant);
     PREP (Preparation); RACT (Reactant or reagent)
         (electrochem. prepn. and cyclic voltammetry on gold electrode and
         UV-visible spectra of)
     7440-57-5, Gold, uses
 TI
      RL: DEV (Device component use); PRP (Properties); USES (Uses)
         (spectroelectrochem. characterization of ultra-thin films formed by
         electropolymn. of phenothiazine derivs. on transparent gold electrodes)
                                                   2679-01-8, Methylene
                               531-53-3, Azur A
      61-73-4, Methylene Blue
 ΙT
      RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent)
         (spectroelectrochem. characterization of ultra-thin films formed by
         electropolymn. of phenothiazine derivs. on transparent gold electrodes)
      143629-25-8P, Poly(Azur A) 150645-86-6P, Poly(methylene
      blue) 161201-31-6P, Poly(methylene green)
      RL: PNU (Preparation, unclassified); PRP (Properties); RCT (Reactant);
      PREP (Preparation); RACT (Reactant or reagent)
         (electrochem. prepn. and cyclic voltammetry on gold electrode and
         UV-visible spectra of)
      143629-25-8 HCAPLUS
      Phenothiazin-5-ium, 3-amino-7-(dimethylamino)-, chloride, homopolymer
 RN
 CN
      (9CI) (CA INDEX NAME)
      CM
      CRN 531-53-3
```

CMF C14 H14 N3 S . C1

Page 114

C1-

150645-86-6 HCAPLUS RN

Phenothiazin-5-ium, 3,7-bis(dimethylamino)-, chloride, homopolymer (9CI) CN(CA INDEX NAME)

1 CM

CRN 61-73-4

CMF C16 H18 N3 S . Cl

● Cl-

161201-31-6 HCAPLUS RN

Phenothiazin-5-ium, 3,7-bis(dimethylamino)-4-nitro-, chloride, homopolymer CN (9CI) (CA INDEX NAME)

CM 1

CRN 2679-01-8

CMF C16 H17 N4 O2 S . C1

● C1-

ΙT

61-73-4, Methylene Blue RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent) (spectroelectrochem. characterization of ultra-thin films formed by electropolymn. of phenothiazine derivs. on transparent gold electrodes)

Page 115

61-73-4 HCAPLUS RN

Phenothiazin-5-ium, 3,7-bis(dimethylamino)-, chloride (9CI) (CA INDEX CN

$$Me_2N$$
 S^+ NMe_2

C1-

```
L85 ANSWER 29 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1995:80676 HCAPLUS
ΑN
     122:174077
DN
     Micrometer patterning of organic materials by selective chemical vapor
ΤI
     deposition
     Sekiguchi, A.; Masuhara, H.
ΑU
     ERATO, Res. Dev. Corp. Japan, Kyoto, 606, Japan,
CS
     Microchem. Proc. JRDC-KUL Jt. Int. Symp. (1994), Meeting Date 1993,
SO
     147-58. Editor(s): Masuhara, Hiroshi. Publisher: North-Holland,
     Amsterdam, Neth.
     CODEN: 60NNAC
     Conference
DT
     English
LA
     74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
     Micrometer-sized patterns of copper phthalocyanines were
AR
     fabricated from 1,2,4,5-tetracyanobenzene by selective chem. vapor
     deposition. The deposition was achieved to produce phthalocyanine
     thin films only on copper micropatterns prepd. on silicon wafers and
     sapphire plates. By thermal annealing in vacuum, fabricated films were
     converted to polymer of copper phthalocyanine. Chem. vapor deposition of 1,2-dicyanobenzene of copper micropatterns under controlled
     conditions resulted in directional growth of copper phthalocyanine
      whiskers and debrises. By examg /effects of deposition rate and
      geometrical structure of micropatterns, the deposition mechanism and an
      important role of micrometer reaction vol. in chem. vapor deposition are
      discussed.
      chem vapor deposition micrometer patterning lithog; copper
 ST
     phthalocyanine micropattern chem vapor deposition; cyanobenzene
      chem vapor deposition copper micropattern
 ΙT
      Lithography
      Vapor deposition processes
         (fabrication of micrometer-sized patterns of copper
         phthalocyanines by selective chem. vapor deposition of
         tetracyanobenzene on copper patterns)
                                     712-74-3, 1,2,4,5-Tetracyanobenzene
      91-15-6, 1,2-Dicyanobenzene
 ΙT
      7440-50-8, Copper, reactions
      RL: PEP (Physical, engineering or chemical process); RCT (Reactant); PROC
      (Process); RACT (Reactant or reagent)
         (fabrication of micrometer-sized patterns of copper
         phthalocyanines by selective chem. vapor deposition of
          tetracyanobenzene on copper patterns)
      147-14-8P, Copper phthalocyanine 26893-93-6P, Copper
 IT
```

LEE 09/828075 Page 116

phthalocyanine polymer RL: PEP (Physical, engineering or chemical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (fabrication of micrometer-sized patterns of copper phthalocyanines by selective chem. vapor deposition of tetracyanobenzene on copper patterns) 26893-93-6P, Copper phthalocyanine polymer ΙT RL: PEP (Physical, engineering or chemical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (fabrication of micrometer-sized patterns of copper phthalocyanines by selective chem. vapor deposition of tetracyanobenzene on copper patterns) 26893-93-6 HCAPLUS Copper, [29H,31H-phthalocyaninato(2-)-.kappa.N29,.kappa.N30,.kappa.N31,.ka RN CN ppa.N32]-, homopolymer (9CI) (CA INDEX NAME) CM 1 CRN 147-14-8 CMF C32 H16 Cu N8 CCI CCS

PAGE 1-A



```
L85 ANSWER 30 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1995:16786 HCAPLUS
AN
     122:146846
     Photosubstitution in thin polymeric films containing Ru(II) polypyridyl
     complexes
     Leasure, Robert M.; Ou, Wei; Moss, John A.; Linton, Richard W.; Meyer,
ΑIJ
     Dep. Chem., Univ. North Carol. Chapel Hill, Chapel Hill, NC, 27599-3290,
CS
     Proceedings - Electrochemical Society (1994), 94-2, 222-34
SO
     CODEN: PESODO; ISSN: 0161-6374
DΤ
     Journal
     English
     74-1 (Radiation Chemistry, Photochemistry, and Photographic and Other
LA
CC
     Reprographic Processes)
     Thin films of poly[Ru(vbpy)2(py)2]2+ (vbpy/is 4-vinyl-4'-methyl-2,2'-
     bipyridine; py is pyridine) have been deposited on
     electrodes by reductive electropolymn. Upon photolysis in the presence of chloride ion, ligand loss photochem. occurs in the films resulting in the
     sequential formation of poly[Ru(vbpy)2(py)Cl]+ and poly[Ru(vbpy)2Cl2].
     Contact lithog. was used to form laterally resolved, bicomponent
     films with image resoln. below 50 .mu.m. Small spot XPS was used to
     confirm that the photolyzed and non-photolyzed regions of the film were
     chem. distinct. The redox and spectral properties of the
      ruthenium-polypyridyl complexes Mithin these films change as chloride ion
      is photosubstituted for the pyridine ligands. This provides a
     basis for the fabrication of bicomponent, electrochromic film assemblies.
     ruthenium polypyridyl complex polymer photolysis photoimaging; ligand
 ST
      photosubstitution ruthenium polypyridyl complex polymer
      Electrochromism
 ΙT
         (based on photosubstitution of pyridine ligand in
         electropolymd. films of Ru(II) polypyridyl complexes by chloride)
      Photolysis
         (of electropolymd. Pu(II) polypyridyl complexes in presence of chloride
 TT
         in solns. and in solid state)
      Redox reaction
 IT
         (electrochem., of bis (pyridine) bis (
         vinylmethylbipyridine)ruthenium(2+))
      Polymerization
 IT
          (electrochem.,/reductive, of bis(pyridine)bis(
         vinylmethylbipyridine) ruthenium(2+))
      Optical imaging devices
 IT
          (electrochromic, based on photosubstitution of pyridine
         ligand in electropolymd. films of Ru(II) polypyridyl complexes by
         chloride)
      Lithography
 IT
          (photo-, imaging based photosubstitution of pyridine ligand
          in electropolymd. films of Ru(II) polypyridyl complexes by chloride)
      Substitution reaction, coordinative
```

Page 118 LEE 09/828075

> (photochem., of pyridine ligand in electropolymd. films of Ru(II) polypyridyl complexes by chloride)

161272-31-7P IT

RL: PEP (Physical, engineering or chemical process); RCT (Reactant); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); RACT (Reactant or reagent); USES (Uses)

(lithog. imaging based on ligand photosubstitution in films

of) 16887-00-6, Chloride, reactions RL: PEP (Physical, engineering or chemical process); RCT (Reactant); PROC ΙT (Process); RACT (Reactant or reagent)

(photosubstitution of pyridine ligand in electropolymd. films of Ru(II) polypyridyl complexes by)

153824-72-7, (4-Vinyl-4'-methyl-2,2'-bipyridine)bis(IT pyridine) ruthenium (2+) RL: RCT (Reactant); RACT (Reactant or reagent)

(reductive electropolymn. of)

RL: PEP (Physical, engineering or chemical process); RCT (Reactant); 161272-31-7P IT SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); RACT (Reactant or reagent); USES (Uses)

(lithog. imaging based on ligand photosubstitution in films of)

161272-31-7 HCAPLUS RN

Ruthenium(2+), bis(4-ethenyl-4'-methyl-2,2'-bipyridine-CN .kappa.N1,.kappa.N1')bis(pyridine)-, bis[hexafluorophosphate(1-)], homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 153824-72-7 C36 H34 N6 Ru CMF CCI CCS

CM

CRN 16919-18-9

```
LEE 09/828075
                       Page 119 ·
     CMF F6 P
     CCI CCS
L85 ANSWER 31 OF 46 HCAPLUS COPYRIGHT 2002 ACS
      1994:486261 HCAPLUS
AN
DN
      121:86261
      Azeotrope-like compositions of 1,1-dichloro-1-fluoroethane, C6-alkane, and
ΤI
      optionally alkanol and/or nitromethane
      Swan, Ellen Louise; Logsdon, Peter Brian; Basu, Rajat Subhra
 IN
      Allied-Signal Inc., USA
 PΑ
      PCT Int. Appl., 37 pp.
 SO
      CODEN: PIXXD2
      Patent
 DТ
      English
 LA
      ICM C23G005-028
 IC
      ICS C11D007-50
      45-5 (Industrial Organic Chemicals, Leather, Fats, and Waxes)
      Section cross-reference(s): 37
 FAN.CNT 1
                                              APPLICATION NO. DATE
                        KIND DATE
      PATENT NO.
       _____
                        ____
                                               WO 1992-US9623
                                                                 19921106
                        A1
                               19930513
          W: AU, BB, BG, BR, CA, CS, FI, HU, JP, KP, KR, LK, MG, MN, MW, NO,
      WO 9309271
 PΤ
               PL, RO, RU, SD
           RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, SE
230696 A1 19930607 AU 1992-30696 19921106
      AU 9230696
                               199111/08
 PRAI US 1991-790743
                               19920205
      US 1992-830817
                               1992/1106
      WO 1992-US9623
      Ozone layer-compatible tit/e compns. contg. C6-alkanes: dimethylbutane,
      methylpentane and isohexame, are useful in degreasing, cold cleaning, and
 AB
       solvent cleaning, e.g., dry cleaning or printed circuit boards
       defluxing. For example, /a title compn. contg. 1,1-dichloro-1-fluoroethane
       98.96 and 2-methylpentare 1.04 wt.% has b.p. 32.16.degree. (760 mm).
       chlorofluoroethane alkane azeotrope; methylpentane dichlorofluoroethane
       azeotrope; degreasing dichlorofluoroethane azeotrope; ozone layer compatibility dichlorofluoroethane azeotrope
       Blowing agents
  ΙT
           (dichlorofluoroethane-alkane mixts., azeotrope-like, ozone
          layer-compatible)
       Alcohols, uses
  IT
       RL: USES (Uses)
           (mixts. with \text{di} \not khlorofluoroethane and methylpentane, azeotrope-like,
           ozone layer-compatible, for cleaning)
       Polyisocyanurates
  ΙT
       Urethane polymers, preparation
  KATHLEEN FULLER EIC/1700/LAW LIBRARY 308-4290
```

```
Page 120
LEE 09/828075
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (prepn. of, blowing agents for, dichloroethane-alkane azeotrope-like
        mixts. as)
     Air pollution
        (prevention of ozone layer depletion in, dichlorofluoroethane-alkane
ΙT
        azeotrope-like mixts. for)
     Alkanes, uses
TT
     RL: USES (Uses)
        (C6, mixts. with dichlorofluoroethane, azeotrope-like, ozone
        layer-compatible, for cleaning)
     Degreasing
IT
        (agents, dichlorofluoroethane-alkane mixts., azeotrope-like, ozone
        layer-compatible)
     Urethane polymers, preparation
IT
     RL: PREP (Preparation)
        (polyester-polyisocyanurate, cellular, prepn. of, blowing agents for,
        dichloroethane-alkane azeotrope-like mixts. as)
     Polyisocyanurates
ΙT
     RL: PREP (Preparation)
         (polyester-polyurethane-, cellular, prepn. of, blowing agents for,
        dichloroethane-alkane azeotrope-like mixts. as)
     Polyesters, preparation
ΙT
      RL: PREP (Preparation)
         (polyisocyanurate-polyurethane-, cellular, prepn. of, blowing agents
         for, dichloroethane-alkane azeotrope-like mixts. as)
     Electric circuits
ΙT
         (printed, boards, cold cleaning and defluxing of,
         dichlorofluoroethane-alkane azeotrope-like mixts. for)
                                               153657-96-6
                                  153657-95-5
                    153657-94-4
      153657-93-3
 TΤ
                    153657-99-9
      153657-98-8
      RL: USES (Uses)
         (azeotrope-like, ozone layer-compatible, for cleaning)
      68270-82-6P 153658-66-3P
 IT
      RL: PREP (Preparation)
         (cellular, prepn. of, blowing agents for, dichloroethane-alkane
         azeotropes as)
      10028-15-6, Ozone, miscellaneous
 ΙT
      RL: MSC (Miscellaneous)
         (layer, depletion prevention of, dichlorofluoroethane-alkane azeotropes
         for)
      153658-66-3P
 TΤ
      RL: PREP (Preparation)
         (cellular, prepn. of, blowing agents for, dichloroethane-alkane
         azeotropes as)
      153658-66-3 HCAPLUS
      Isocyanic acid, polymethylenepolyphenylene ester, polymer with
 RN
 CN
      10-[(1-methyl-4-piperidinyl)methyl]-10H-phenothiazine (9CI) (CA INDEX
      NAME)
           1
      CM
      CRN 9016-87-9
      CMF Unspecified
      CCI PMS, MAN
  *** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
       CM
       CRN 3772-71-2
```

CMF C19 H22 N2 S

```
Me
      CH<sub>2</sub>
L85 ANSWER 32 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1993:582947 HCAPLUS
AN
     119:182947
     Pigment compositions and pigment dispersions
TТ
     Ide, Jusaku; Kikuchi, Jiro; Maejima/, Masuhiko
IN
     Toyo Ink Mfg Co, Japan
PΑ
     Jpn. Kokai Tokkyo Koho, 9 pp.
SO
     CODEN: JKXXAF
DT
     Patent
     Japanese
LA
     ICM C09B067-46
IC
     ICS C09B067-20; C09C003-08; £09C003-10
     42-6 (Coatings, Inks, and Related Products)
     Section cross-reference(s): /41
FAN.CNT 1
                                            APPLICATION NO. DATE
                       KIND DATE
     PATENT NO.
                                            _____
                                                             ____
                                                             19911024
                                            JP 1991-305413
                            19Ø30514
     The high-concn. title compns. with good fluidity, useful for coatings and
                        A2
PΙ
     inks, comprise 100 parts pigments and 0.1-30 parts pigment dispersants
     which are salts of (poly) oxyalkylene alkyl ether sulfosuccinate ester Na
     salts and pigment derips. P[X(CH2)kNR1R2]m [P = residue of .gtoreq.1 org.
     dye selected from azo/ phthalocyanine, quinacridone,
     anthraquinone, dioxa ine, anthrapyrimidine, anthanthrone, indanthrone,
      flavanthrone, pyranthrone, perylene, perinone, thioindigo,
      diketopyrrolopyrrol , isoindolinone, insoindoline, and quinophthalone; X =
      direct bond, divalent group; R1-2 = H, (un) substituted alkyl, aryl or R1R2
      = (N- \text{ or } O-\text{contg.})/\text{heterocycle}; k, m = 1-4]. Dispersions comprising the
      pigment compns. and vehicles are also claimed. Thus, a pigment compn. was
      prepd. from 95 parts C.I. Pigment Blue 15:3 and 5 parts salt prepd. from
      100 parts CuPcSO/2NH(CH2)3NEt2 (CuPc = residue of Cu phthalocyanine
      ) and 80 parts Me(CH2)10CO(C2H4O)4COCH(SO3Na)CH2CO2Na. An offset printing
      ink prepd. from 15 parts of the compn. and 35 parts rosin-modified
      phenolic resin/varnish showed good flowability.
      pigment dispersion coating ink fluidity; dispersant pigment coating ink
      Pigments
 ΙT
         (for coat/ngs and inks, dispersants for)
      Dispersing #gents
 IT
         (for pigments, salts of alkylamino-modified pigments and
         polyoxyalkylene alkyl ether sulfosuccinates as)
```

Page 122 09/828075 LEE Inks IT (gravure, pigment compns. for, dispersants in) IT Inks (lithog., pigment compns. for, dispersants in) Phenolic resins, compounds ΙT RL: USES (Uses) (rosin-modified, vehicles, in pigment compns. for coatings and inks) 81-77-6, C.I. Pigment Blue 60 147-14-8, C.I. Pigment Blue 15:1 ΙT 1047-16-1, C.I. Pigment Violet 19 1328-53-6, C.I. Pigment Green 7 4051-63-2, C.I. Pigment Red 177 5281-04-9, C.I. Pigment Red 57:1 6358-30-1, C.I. Pigment Violet 23 5521-31-3, C.I. Pigment Red 179 68134-22-5, C.I. Pigment Yellow 154 6358-85-6, C.I. Pigment Yellow 12 84632-65-5, C.I. Pigment Red 254 RL: USES (Uses) (pigment compns., for coatings and inks, with good fluidity, dispersants for) 150431-69-9P **150521-50-9P** 150419-07-1P 150419-04-8P TΤ 150575-59-0P 150575-60-3P 150575-58-9P 150521-51-0P 150575-63-6P 150575-69-2P 150575-62-5P 150575-61-4P 150575-70-5P RL: PREP (Preparation) (prepn. of, dispersants, for pigments, in coatings and inks) 150521-50-9P 150521-51-0P 150575-69-2P ΙT 150575-70-5P RL: PREP (Preparation) (prepn. of, dispersants, for pigments, in coatings and inks) 150521-50-9 HCAPLUS Copper, [2-[[3-(dibutylamino)propyl]amino]-N-(29H,31H-phthalocyanin-C-CN ylmethyl)acetamidato(2-)-N29,N30,N31,N32]-, compd. with .alpha.-(3-carboxy-1-oxosulfopropyl)-.omega.-phenoxypoly(oxy-1,2ethanediyl) (9CI) (CA INDEX NAME) CM 1 150463-29-9 CRN CMF C46 H45 Cu N11 O CCI CCS, IDS

Page 123 LEE 09/828075

> 2 CM

CRN 150419-06-0

CMF (C2 H4 O)n C10 H10 O7 S

CCI IDS, PMS

3 CM

CRN 9004-78-8

CMF (C2 H4 O)n C6 H6 O

CCI PMS

$$HO = \begin{bmatrix} CH_2 - CH_2 - O \end{bmatrix}_n$$

CM4

CRN 5138-18-1 CMF C4 H6 O7 S

$$\begin{array}{c} \mathrm{SO_3H} \\ | \\ \mathrm{HO_2C-CH-CH_2-CO_2H} \end{array}$$

150521-51-0 HCAPLUS RN

Copper, [N,N-dipropyl-29H,31H-phthalocyanine-C-methanaminato(2-)-N29,N30,N31,N32]-, polymer with .alpha.-(3-carboxy-1-oxosulfopropyl)-.omega.-[(1-oxooctadecyl)oxy]poly(oxy-1,2-ethanediyl) (9CI) (CA INDEX CN NAME)

1 CM

CRN 150463-30-2 CMF C39 H31 Cu N9

CCI CCS, IDS

Page 124

 $(n-Pr)_2N-CH_2-D1$

CM 2

CRN 123543-88-4

CMF (C2 H4 O)n C22 H40 O8 S

CCI IDS, PMS

CM 3

CRN 9004-99-3

CMF (C2 H4 O)n C18 H36 O2

CCI PMS

Me-
$$(CH_2)_{16}$$
 - C - O - CH_2 - CH_2 - O - O

CM 4

CRN 5138-18-1 CMF C4 H6 O7 S

$$^{\rm SO_3H}_{\rm HO_2C-CH-CH_2-CO_2H}$$

RN 150575-69-2 HCAPLUS
CN Copper, [N-[3-(diethylamino)propyl]-29H,31H-phthalocyanine-Csulfonamidato(2-)-N29,N30,N31,N32]-, compd. with .alpha.-(3-carboxy-1oxosulfopropyl)-.omega.-[(1-oxododecyl)oxy]poly(oxy-1,2-ethanediyl) (9CI)
(CA INDEX NAME)

CM 1

LEE 09/828075 Page 125

CRN 93971-95-0 CMF C39 H32 Cu N10 O2 S CCI CCS, IDS

CM2

CRN 119845-20-4

CMF (C2 H4 O)n C16 H28 O8 S CCI IDS, PMS

CM 3

CRN 9004-81-3

CMF (C2 H4 O)n C12 H24 O2 CCI PMS

Me-
$$(CH_2)_{10}$$
 - C - CH_2 - CH

CM

CRN 5138-18-1 CMF C4 H6 O7 S

Page 126

 $\begin{array}{c} \text{SO}_3\text{H} \\ | \\ \text{HO}_2\text{C}-\text{CH}-\text{CH}_2-\text{CO}_2\text{H} \end{array}$

CM 1

CRN 150463-31-3 CMF C39 H24 C18 Cu N10 O2 S CCI CCS, IDS

PAGE 1-A

8 (D1-C1)

PAGE 2-A

CM 2

CRN 119845-20-4 CMF (C2 H4 O)n C16 H28 O8 S CCI IDS, PMS

CM 3

CRN 9004-81-3

```
LEE 09/828075
```

Page 127

CMF (C2 H4 O)n C12 H24 O2 CCI PMS

Me-
$$(CH_2)_{10}$$
 - C - CH_2 - CH

CM 4

CRN 5138-18-1 CMF C4 H6 O7 S

```
L85 ANSWER 33 OF 46 HCAPLUS COPYRIGHT 2002/ACS
     1991:91982 HCAPLUS
ΑN
     114:91982
DN
     Photocurable composition
ΤI
     Hamilton, John; Dickinson, Peter
IN
     Sericol Group Ltd., UK
·PΑ
     Eur. Pat. Appl., 11 pp.
SO
     CODEN: EPXXDW
DT
     Patent
     English
LA
     ICM G03F007-038
ICS G03F007-033; G03F007-105
IC
     74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
     Section cross-reference(s): 35
FAN.CNT 1
                                             APPLICATION NO. DATE
                        KIND DATE
     PATENT NO.
                                             _____
                       ____
                                                               19891211
                            19900620
                                             EP 1989-312926
                        A1
ΡI
     EP 373862
         R: AT, BF, CH, DE, ES, FR, GB, IT, LI, NL, SE
                                             GB 1988-29359
                                                               19881216
                             19900704
     GB 2226564
                        A1
                              19930317
                        B2
     GB 2226564
                                             US 1989-448966
                                                               19891212
                              19911029
                        Α
      US 5061603
                                             JP 1989-324211
                                                               19891215
      JP 022768∅6
                        A2
                              19901113
PRAI GB 1988-19359
OS CASREACT 114:91982
                              19881216
 GI
                            x<sup>-</sup>
 R<sup>1</sup>ON
                                Ι
/KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290
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LEE 09/828075 Page 128 Photopolymerizable grafted poly(vinyl alc.) prepd. using I [R1 = alkyl, AB aralkyl, H; R2 = alkyl, amino, cyano, etc.; R3 = Cl, H, a fused ring group; X- = an anion] and a photocurable compn. prepd. therefrom for producing screen printing stencils are claimed. The above compn. shows high sensitivity with low level of photocrosslinkable groups grafted to the poly(vinyl alc.). Thus, N-methoxy-4-(4-formylstyryl) pyridinium 4-methylbenzenesulfonate was prepd. and used to graft poly(vinyl alc.). The product was used in a photocurable compn. photopolymerizable compn graft polyvinyl alc; styrylpyridinium STpolyvinyl alc graft Printing plates ΙT (screen, photocurable compns. contg. methoxy(formylstyryl) pyridinium-grafted poly(vinyl alc.) for prepn. of) 105521-74-2, Gohsenol GH 20 IT RL: USES (Uses) (photopolymerizable compn. contg.) 27371-27-3P 1003-67-4P, 4 Methylpyridine-N-oxide Τጥ RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation) (prepn. and reaction of, for photopolymerizable compn. prepn.) 131854-13-2P ΙT RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. and use of, in photoimaging compn.) 131854-12-1P TΨ RL: PREP (Preparation) (prepn. of and poly(vinyl alc.) grafting with, for photopolymerizable compn.) IT 131854-13-2P RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. and use of, in photoimaging compn.) 131854-13-2 HCAPLUS Pyridinium, 4-[2-(4-formylphenyl)ethenyl]-1-methoxy-, salt with RN 4-methylbenzenesulfonic acid (1:1), polymer with ethenol, graft (9CI) (CA INDEX NAME) CM 1 CRN 557-75-5 CMF C2 H4 O H2C== CH- OH

CM 2

CRN 131854-12-1 CMF C15 H14 N O2 . C7 H7 O3 S

CM 3

CRN 131854-11-0 CMF C15 H14 N O2

$$\mathsf{OHC} \qquad \mathsf{CH} = \mathsf{CH} \qquad \mathsf{N}^+$$
 OMe

CM 4

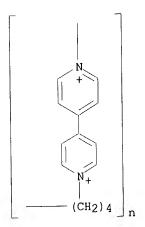
CRN 16722-51-3 CMF C7 H7 O3 S

```
L85 ANSWER 34 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1991:62910 HCAPLUS
AN
     114:62910
DN
     Photoreduction of alkyl polyviologens and their low molecular analogs in
ΤI
     poly(vinyl alcohol) (PVA) matrix
     Liang, Z. X.; Li, W.; Chen, Y. L.
ΑU
     Inst. Polym. Sci., Zhongshan Univ., Guangzhou, Peop. Rep. China
CS
     J. Macromol. Sci., Chem. (1990), A27(6), 699-709
SO
     CODEN: JMCHBD; ISSN: 0022-233X
DT
     Journal
     English
LA
     35-8 (Chemistry of Synthetic High Polymers)
CC
     The photoredn. behavior of alkyl viologen compds. in a PVA matrix was investigated. The initial photoredn. of alkyl viologens and related
     polyviologens followed second-order reaction kinetics, and the resp. rate
     consts., which vary only slightly with alkyl chain length, were detd.
     While the polymer effect was not remarkable in a PVA matrix, the assocn.
     behavior of radical cations was similar to that found in a 2-propanol aq.
     soln.
     viologen photoredn polyvinyl alc; /kinetics photoredn polyviologen
ST
IT
     Ionene polymers
     RL: RCT (Reactant)
         (bipyridine-dibromoalkane, photoredn. of, kinetics of, poly(vinyl alc.)
        matrix effect on)
ΙT
     Kinetics of reduction
         (photochem., of viologens,/in poly(vinyl alc.) matrix)
                              34075-15-5 36437-30-6 37540-89-9
                  32449-18-6
IT
     6198-51-2
                                52/243-87-5 53721-12-3
                   47660-20-8
     37541-09-6
      54391-26-3 60723-01-5 63812/59-9 69860-09-9
                   83613-45-0 105/009-21-0 105009-22-1
      69860-10-2
                                                 106917-11-7
                    106917-09-3
                                  106917-10-6
     106725-29-5
                                  106917-14-0
                                                106917-15-1
                    106917-13-9
      106917-12-8
      RL: RCT (Reactant)
         (photoredn. of, kinetiks of, poly(vinyl alc.) matrix effect on)
      9002-89-5, Poly(vinyl alcohol)
 ΙT
      RL: PRP (Properties)
         (viologen photoredn. in matrix of, kinetics of)
      37540-89-9 37541-09-6 53721-12-3
      60723-01-5 63812-59-9 69860-09-9
      69860-10-2 105009-21-0 105009-22-1
      106725-29-5
      RL: RCT (Reactant)
         (photoredn. of, kinetics of, poly(vinyl alc.) matrix effect on)
```

Page 130

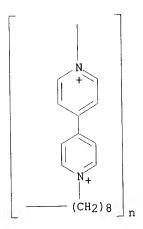
37540-89-9 HCAPLUS RN

Poly([4,4'-bipyridinium]-1,1'-diyl-1,4-butanediyl dibromide) (9CI) (CA CN INDEX NAME)



●2 Br-

37541-09-6 HCAPLUS RNPoly([4,4'-bipyridinium]-1,1'-diyl-1,8-octanediyl dibromide) (9CI) (CA CN INDEX NAME)



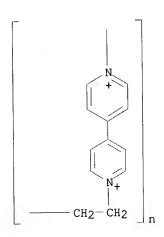
●2 Br-

53721-12-3 HCAPLUS RN 4,4'-Bipyridinium, 1,1'-diethyl-, dibromide (9CI) (CA INDEX NAME) CN

Page 131

●2 Br-

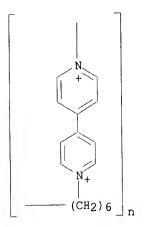
RN 60723-01-5 HCAPLUS CN Poly([4,4'-bipyridinium]-1,1'-diyl-1,2-ethanediyl dibromide) (9CI) (CA INDEX NAME)



●2 Br-

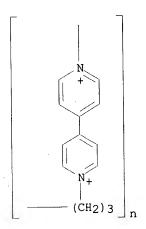
RN 63812-59-9 HCAPLUS CN Poly([4,4'-bipyridinium]-1,1'-diyl-1,6-hexanediyl dibromide) (9CI) (CA INDEX NAME)

Page 132



●2 Br-

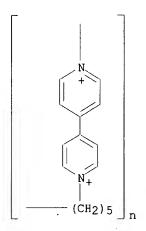
RN 69860-09-9 HCAPLUS CN Poly([4,4'-bipyridinium]-1,1'-diyl-1,3-propanediyl dibromide) (9CI) (CA INDEX NAME)



●2 Br-

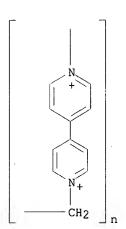
RN 69860-10-2 HCAPLUS CN Poly([4,4'-bipyridinium]-1,1'-diyl-1,5-pentanediyl dibromide) (9CI) (CA INDEX NAME)

Page 133



●2 Br-

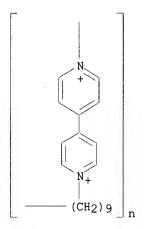
RN 105009-21-0 HCAPLUS
CN Poly([4,4'-bipyridinium]-1,1'-diylmethylene dibromide) (9CI) (CA INDEX NAME)



●2 Br-

RN 105009-22-1 HCAPLUS CN Poly([4,4'-bipyridinium]-1,1'-diyl-1,9-nonanediyl dibromide) (9CI) (CA INDEX NAME)

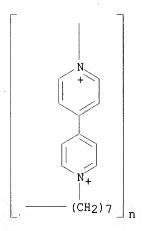
Page 134



●2 Br~

RN 106725-29-5 HCAPLUS CN

Poly([4,4'-bipyridinium]-1,1'-diyl-1,7-heptanediyl dibromide) (9CI) (CA INDEX NAME)



●2 Br-

L85 ANSWER 35 OF 46 HCAPLUS COPYRIGHT 2002 ACS

ΑN 1989:31418 HCAPLUS

DN 110:31418

ΤI Printing plate with intermediate layer containing

photosensitive composition Pawlowski, Georg; Lehmann, Peter Hoechst A.-G., Fed. Rep. Ger. IN

PA

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LEE 09/828075
                      Page 135
SO
     Eur. Pat. Appl., 21 pp.
     CODEN: EPXXDW
DT
     Patent
LA
     German
IC
     ICM G03F007-02
     74-4 (Radiation Chemistry, Photochemistry, and Photographic and Other
CC
     Reprographic Processes)
                                           APPLICATION NO.
     PATENT NO.
                      KIND DATE
                                                             DATE
     -----
                            -----
                      ----
PΙ
     EP 272550
                       A2
                            19880629
                                            EP 1987-118279
                                                             19871210
                            19890607
     EP 272550
                       A3
     EP 272550
                      В1
                            19920812
         R: AT, BE, CH, DE, ES, FR, GB, IT, LI, ML, SE
                                           DE 1986-3644160
     DE 3644160
                      A1
                           19880714
                                                            19861223
                                           FI 1987-5621
     FI 8705621
                       Α
                            19880624
                                                             19871221
                       A2
                            19880715
                                            JP 1/987-321630
     JP 63172152
                                                             19871221
     ZA 8709597
                       Α
                            19880727
                                            ZA/1987-9597
                                                             19871222
     BR 8707009
                      Α
                            19880802
                                           BR/ 1987-7009
                                                             19871222
     AU 8783010
                      A1
                            19880623
                                           AU 1987-83010
                                                             19871223
     AU 604815
                       B2
                            19910103
     US 4956261
                            19900911
                                            US 1989-403007
                                                             19890906
                       Α
PRAI DE 1986-3644160
                            19861223
     US 1987-135311
                            19871221
     A photosensitive material contg. a/polycondensation product diazonium
     salt, a swellable polymer, a photopolymn. initiator, and a polymerizable
     compd. having .gtoreq.1 unsatd. groups and m.p. .gtoreq.100.degree.,
     comprises an intermediate layer contg. a polycondensation product
     diazonium salt and H2O-insol., qrg. solvent-sol., and aq. alkali
     soln.-sol. swellable binder polymer. Thus, an anodized Al plate (was)
     coated with a compn. contg. maleic anhydride-vinyl acetate-vinyl butyral
     copolymer, 4,4 - bismethoxymethyldiphenyl ether-3-methoxydiphenylamine-4-
     diazonium sulfate copolymer, H3PO4, Colorant, phenylazodiphenylamine, ethyleneglycolmonomethyl ether and solvent. The above layer was coated
     with a photosensitive layer. The above material could be used to produce
     420,000 offset prints without any visible damage.
ST
     printing offset plate intermediate layer; diazonium
     salt polycondensation product; swellable polymer binder photosensitive
     compn
IT
     Vinyl acetal polymers
     RL: USES (Uses)
        (butyrals, photosensitive compn. contg., for printing
        plate)
IT
     Lithographic plates/
        (offset, swellable polymer and polymer diazonium salt in intermediate
        layer for)
ΙT
     78-63-7
              85-44-9D, Phthalic anhydride, reaction product with vinyl
     butyral polymers 108-05-4D, Vinyl acetate, reaction product with vinyl
                     108-30-5D, Succinic anhydride, reaction product with
     acetal polymers
     vinyl butyral polymers 108-31-6D, 2,5-Furandione, reaction product with
     vinyl acetal polymers 111-36-4D, Butyl isocyanate, reaction product with
     vinyl butyral polymers 111-46-6, uses and miscellaneous 123-86-4,
                     584-84-9 624-83-9D, Methyl isocyanate, reaction product
     Butyl acetate
     with vinyl butyral polymers
                                  699-98-9D, Pyridine
     -2,3-dicarboxylic acid anhydride, reaction product with vinyl butyral
     polymers 818-61-1 3524-68-3, Pentaerythritol triacrylate
     5-Hydroxypentanal, reaction product with vinyl butyral polymers
     4480-83-5D, 1,4-Dioxane-2,6-dione, reaction product with vinyl butyral
     polymers 4986-89-4, Pentaerythritol tetraacrylate 29570-58-9,
     Dipentaerythritol hexaacrylate 30674-80-7 71154-39-7 117992-04-8
```

Page 136

RL: USES (Uses)

(photosensitive compn. contg., for printing plate)

IT 71510-01-5P

RL: SPN (Synthetic preparation); PREP (Preparation)

(prepn. of)

IT **71510-01-5P**

RL: SPN (Synthetic preparation); PREP (Preparation)

(prepn. of)

RN 71510-01-5 HCAPLUS

CN Benzenediazonium, 2-methoxy-4-(phenylamino)-, sulfate (2:1), polymer with

1,1'-oxybis[4-(methoxymethyl)benzene] (9CI) (CA INDEX NAME)

CM 1

CRN 2509-26-4 CMF C16 H18 O3

CM 2

CRN 29377-89-7

CMF C13 H12 N3 O . 1/2 O4 S

CM 3

CRN 32445-12-8 CMF C13 H12 N3 O

CM 4

CRN 14808-79-8 CMF O4 S

L85 ANSWER 36 OF 46 HCAPLUS COPYRIGHT 2002 ACS

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

09/828075 Page 137 ΑN 1988:414813 HCAPLUS DN 109:14813 ΤI Photosensitive resin compositions containing photocrosslinkable vinyl alcohol polymers for screen printing IN Oka, Toshio PA Daicel Chemical Industries, Ltd., Japan Jpn. Kokai Tokkyo Koho, 8 pp. SO CODEN: JKXXAF DT Patent LA Japanese TC ICM G03C001-68 C08F002-48; C08F008-28; C08F299-00; C08K005-28; C08L029-14; G03C001-71 CC 74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes) FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE PΙ JP 62247354 Α2 19871028 JP 1986-87058 19860417 GΙ

$$\begin{array}{c} -\text{O} \\ \text{CH} \left\{ \left(\text{CH}_2 \right)_n \text{O} \right\}_m \\ -\text{O} \\ \text{CH} \left\{ \left(\text{CH}_2 \right)_n \text{O} \right\}_m \\ -\text{O} \\ \text{CH} \left\{ \left(\text{CH}_2 \right)_n \text{O} \right\}_m \\ \end{array}$$

$$\begin{array}{c} \text{CH} = \text{CHCOB} \\ \text{II} \\ -\text{O} \\ \text{COCH} = \text{CHB} \\ \end{array}$$

$$-0$$

$$R^{2}$$

$$X^{-}$$

$$X^{-}$$

$$X^{-}$$

$$X^{-}$$

'n1

IV

CH[(CH2)nO]m

The compns. contain (a) film-forming polymer, (b) photocrosslinkable poly(vinyl alc.) having alkylene oxide deriv. groups I, II, and/or III (A = IV, V; B = arom. or heterocyclic group having polar group; R1 = H, alkyl, or aralkyl optionally having OH, CONH2, ether bond, or unsatd. bond; R2 = H, alkyl; X- = anion; m = 0, 1; n = 1-6), and (c) H2O-sol. azide or quinoneazide. The compns. have good shelf-life and provide high resoln. latent images. Thus, an ad. soln. of a photopolymerizable poly(vinyl alc.) 4,4'-diazidostilbene-2,2'-disulfonic acid and a poly(vinyl acetate) emulsion were mixed, coated on a polyester mesh and dried to obtain a 3-.mu. layer. Exposure and development of the layer with H2O gave a pattern with 100-.mu. linewidth resoln.

V

ST printing plate screen photosensitive compn; azide photosensitive compn screen printing; quinonediazide photosensitive compn screen printing; vinyl acetal polymer printing

```
LEE 09/828075
                      Page 138
     platemaking
ΙT
     Photoimaging compositions and processes
        (contg. vinyl acetal polymer and azide compd.)
ΙT
     Vinyl acetal polymers
     RL: USES (Uses)
        (stilbazolinium and stilquinolinium group-contg.,
        photocrosslinkable, for screen printing plates)
ΙT
     Printing plates
        (screen, azide-contg. photosensitive compn. for)
ΙT
     15874-22-3
     RL: USES (Uses)
        (photosensitive layer contg., for screen printing
        platemaking)
IT
                   78521-12-7P 107845-60-3P
     74401-04-0P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation)
        (prepn. and reaction of, screen printing plate
        photosensitive polymer from)
IT
     9002-89-5D, stilbazolium or stilquinolinium group-contg.
     RL: USES (Uses)
        (screen printing photosensitive plate using)
ΙT
     107845-60-3P
     RL: RCT (Reactant); SPN (Synthetic preparation); PREP
     (Preparation)
        (prepn. and reaction of, screen printing plate
        photosensitive polymer from)
RN
     107845-60-3 HCAPLUS
     Pyridinium, 4-[2-[4-(2,2-dimethoxyethoxy)phenyl]ethenyl]-1-methyl-, methyl
CN
     sulfate, polymer with ethenol (9CI) (CA INDEX NAME)
     CM
     CRN
         557-75-5
     CMF C2 H4 O
H_2C = CH - OH
     CM
          2
         74401-18-6
     CRN
          C18 H22 N O3 . C H3 O4 S
               3
          CM
          CRN
               74401-13-1
          CMF C18 H22 N O3
                        CH=== CH-
    OMe
MeO-CH-CH2-
```

CM

LEE 09/828075 Page 139

> CRN 21228-90-0 CMF C H3 O4 S

Me-0-S03-

L85 ANSWER 37 OF 46 HCAPLUS COPYRIGHT 2002 ACS

AN 1987:587489 HCAPLUS

DN 107:187489

ΤI Light-sensitive composition and presensitized lithographic plate

IN Suzuki, Norihito; Goto, Sei; Maeda, Yoshihiro; Shimizu, Shigeki

Konishiroku Photo Industry Co., Ltd., Japan; Mitsubishi Chemical Industries Co., Ltd.

SO Jpn. Kokai Tokkyo Koho, 12 pp.

CODEN: JKXXAF

DTPatent

Japanese LA

IC

ICM G03C001-00 ICS G03C001-00; G03C001-71; G03C005-24; G03F007-02

74-6 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

FAN.CNT 1

111110	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡT	JP 62111249	A2	19870522	JP 1985-251743	19851109
	JP 06042071	B4	19940601	01 1903 291743	15051105
	CA 1311071	A1	19921201	CA 1986-522455	19861107
	US 4912013	Α	19900327	US 1988-268632	19881108
PRAI	JP 1985-251743		19851109		
	US 1986-926637		19861103		

GΙ

$$(X)^{m}$$

$$N = N$$

$$(X)^{m}$$

$$(X)^{m}$$

$$(X)^{m}$$

$$(X)^{m}$$

The title light-sensitive compn., comprising a light-sensitive diazo resin and lipophilic polymer, contains an org. solvent-sol. metal complex salt dye. A metal in the metal complex salt is selected from Cu, Co, Ni, and Cr. The metal complex salt may be represented by the formula, I [X, Y = H, C1-10 alkyl, C1-4 alkoxy, C2-5 alkoxy carbonyl, C2-5 acyl, aminocarbonyl, C2-5 alkylaminocarbonyl, C1-3 alkylsulfonyl, aminosulfonyl, C2-5 acylamino, nitro, cyano, halo; m = 1-4; n = 1-6; X = Y or X .noteq. Y when X .gtoreq. 2 and Y .gtoreq. 2; Q+ = cation; Q+ is preferably RR'NH2 (R, R' = C1-20 alkyl, hydroxyalkyl); M = Co, Ni, Cr]. The metal complex salt may be a modified Cu phthalocyanine complex. The presensitized lithog. plates prepd. by using the above photosensitive compns. are also claimed. The use of the org. solvent-sol. metal complex dyes improves mixing of the dye with photosensitive compn., and also dyes show greater resistance toward leaching out by developers, plate cleaners, or inks.

prate cleaners, of links.

ST presensitized plate metal complex dye; phthalocyanine copper presensitized lithog plate; phenylazonaphthol metal complex presensitized plate

Lithographic plates
(presensitized, light-sensitive compn. contg. diazo resin and hydrophobic polymer and metal complex dye for)

IT 39290-77-2 110880-69-8 1/0880-70-1 RL: USES (Uses)

TΤ

(light-sensitive compn. contg., for presensitized lithog.

plate, with improved fading resistance)

9070-36-4P 59592-92-6P, Acrylonitrile-2-hydroxyethyl methacrylate-methacrylic acid-methyl methacrylate copolymer 77833-95-5P, Acrylonitrile-ethyl acrylate-N-(4-hydroxyphenyl)methacrylamide-methacrylic acid copolymer 96536-79-7P

RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. and use of, for presensitized lithog. plates)

Page 141

IT 9070-36-4P

RL: SPN (Synthetic preparation); PREP (Preparation) (prepn. and use of, for presensitized lithog. plates)

RN 9070-36-4 HCAPLUS

Benzenediazonium, 4-(phenylamino)-, sulfate (2:1), polymer with formaldehyde (9CI) (CA INDEX NAME) CN

CM 1

CRN 50-00-0 CMF C H2 O

 $H_2C = O$

2 CM

CRN 150-33-4

CMF C12 H10 N3 . 1/2 O4 S

CM 3

CRN 16072-57-4 CMF C12 H10 N3

CM

CRN 14808-79-8 CMF 04 S

L85 ANSWER 38 OF 46 HCAPLUS COPYRIGHT 2002 ACS

1987:85201 HCAPLUS AN

DN 106:85201

Chemical reduction behavior of alkyl viologens and polyviologens

Li, Manfu; Liang, Zhaoxi; Li, Wen

CS

Inst. Polym., Zhongshan Univ., Canton, Peop. Rep. China Zhongshan Daxue Xuebao, Ziran Kexueban (1986), (1), 94-9 SO CODEN: CHTHAJ; ISSN: 0529-6579

DTJournal

Chinese LA

LEE 09/828075 Page 142 35-8 (Chemistry of Synthetic High Polym∉rs) CC Section cross-reference(s): 27 The relationship between structure and chem. redn. behavior of alkyl viologens and polyviologens in the presence of Na2S2O4 was studied using electrospectroscopy. The redn. of alkyl viologens was dependent on the alkyl chain length. The behavior of polyviologens was quite different from their resp. alkyl viologens, only the 1st step of redn. could be obsd. The intramol. assocn. of produced polyradical cations occurred even in dil. aq. soln. ST viologen structure chem redn; pólyviologen structure chem redn; structure redn relationship viologen; sodium hdyrosulfite redn viologen Reduction (of alkyl viologens and pólyviologens, structure in relation to) IT Chains, chemical (structure of, of polyviologens, redn. behavior in relation to) IT 7681-38-1, Sodium hydrosulfate RL: RCT (Reactant) (redn. by, of alkyl viologens and polyviologens) 32449-18-6 IT 36437-30-6 47660-20-8 3240-78-6 6159-05-3 /6198-51-2 52243-87-5 **53721-12-3** / 54391-26-3 **106899-75-6 106899-76** / **106899-77-8** 83613-45-0 106899-78-9 106899-79-0 106911-71-1 106911-72-2 106911-7\$-3 106911-74-4 10691/-09-3 106917-10-6 106917-11-7 106917-08-2 106917-12-8 106917-13-9 10691/7-14-0 106917-15-1 RL: RCT (Reactant) (redn. of) 53721-12-3 106899-75-6 106899-76-7 IT 106899-77-8 106899-78-9 106899-79-0 106911-71-1 106911-72-2 106911-73-3 106911-74-4 RL: RCT (Reactant) (redn. of) 53721-12-3 HCAPLUS RN CN 4,4'-Bipyridinium, 1,1'-diethyl-, dibromide (9CI) (CA INDEX NAME) Br-106899-75-6 HCAPLUS CN 4,4'-Bipyridinium, 1,1'-dimethyl-, dibromide, homopolymer (9CI) (CA INDEX NAME) CM 1

CRN 3240-78-6

CMF C12 H14 N2 . 2 Br

Page 143

●2 Br-

RN 106899-76-7 HCAPLUS

CN 4,4'-Bipyridinium, 1,1'-diethyl-, dibromide, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 53721-12-3

CMF C14 H18 N2 . 2 Br

●2 Br-

RN 106899-77-8 HCAPLUS

CN 4,4'-Bipyridinium, 1,1'-dipropyl-, dibromide, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 52243-87-5

CMF C16 H22 N2 . 2 Br

Page 144

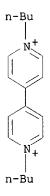
●2 Br-

RN 106899-78-9 HCAPLUS CN 4,4'-Bipyridinium, 1,1'-dibutyl-, dibromide, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 32449-18-6

CMF C18 H26 N2 . 2 Br



●2 Br-

RN 106899-79-0 HCAPLUS CN 4,4'-Bipyridinium, 1,1'-dipentyl-, dibromide, homopolymer (9CI) (CA INDEX NAME)

CM :

CRN 54391-26-3

CMF C20 H30 N2 . 2 Br

Page 145

●2 Br-

RN 106911-71-1 HCAPLUS
CN 4,4'-Bipyridinium, 1,1'-dihexyl-, dibromide, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 6198-51-2 CMF C22 H34 N2 . 2 Br

●2 Br-

RN 106911-72-2 HCAPLUS
CN 4,4'-Bipyridinium, 1,1'-dioctyl-, dibromide, homopolymer (9CI) (CA INDEX NAME)

CM :

CRN 36437-30-6 CMF C26 H42 N2 . 2 Br

Page 146

●2 Br-

RN 106911-73-3 HCAPLUS CN 4,4'-Bipyridinium, 1,1'-dinonyl-, dibromide, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 47660-20-8 CMF C28 H46 N2 . 2 Br

●2 Br~

RN 106911-74-4 HCAPLUS
CN 4,4'-Bipyridinium, 1,1'-didecyl-, dibromide, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 34075-15-5 CMF C30 H50 N2 . 2 Br

●2 Br-

```
L85 ANSWER 39 OF 46 HCAPLUS COPYRIGHT 2002 ACS
    1986:139359 HCAPLUS
AN
DN
    104:139359
    Electrochromic display devices
    Chevron Research Co. , USA
    Jpn. Kokai Tokkyo Koho, 16 pp.
    CODEN: JKXXAF
DT
    Patent
    Japanese
LA
IC
     ICM C09K009-00
     ICS G02F001-17; G09F009-00
     74-9 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 38
FAN.CNT 1
    PATENT NO.
                     KIND DATE
                                           APPLICATION NO.
                                                            DATE
PΙ
     JP 60137985
                      A2
                            19850722
                                           JP 1984-243147
                                                            19841117
     EP 145337
                      В1
                           19900808
                                           EP 1984-307966
                                                            19841116
        R: DE, FR, GB, IT, NL
PRAI US 1983-554153
                           19831121
    The claimed electrochromic display devices contain display electrodes
     coated with an electrochromic polymer having arom. heterocyclic ring type
     structural repeating units contq. a Group Va or VIa element and if
     necessary linkage units obtained from a radically polymerizable monomer.
     The linkage units should form A conjugated .pi.-electron system with the
     heterocyclic rings. The disp/ay devices also contain an electrolyte soln.
     in an org. solvent, and a counter electrode. Polymers having quinoline
     rings, oxazole rings, thiazole rings, phenoxazine rings, etc. are esp.
     useful.
    electrochromic display heterocycle polymer; quinoline polymer
     electrochromic display
IT
    Optical imaging devices
        (electrochromic, het∉rocyclic polymers for display electrodes of)
IT.
     25868-25-1
                  26023-43-8
                               26023-46-1
                                            26023-47-2
                                                         26183-62-0
     26894-31-5
                  26967-12-4
                               28576-59-2
                                            32038-01-0
                                                         39014-53-4
     51032-49-6
                  59827-48-2
                               59827-56-4
                                            60871-72-9
                                                         69794-31-6
                               85227-28-7
                                            85227-29-8 91777-82-1
     76996-74-2
                  85227-27-6
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LEE 09/828075
                      Page 148
     92093-07-7
                  92093-08-8
                               96208-50-3
                                           96679-82-2
                                                       101150-26-9
                   101150-63-4
                                 101150-64-5
     101150-62-3
                                               101150-65-6
                                                             101150-66-7
     101150-67-8
                   101150-68-9 101150-70-3
                                               101150-79-2
                                                             101150-80-5
     RL: USES (Uses)
        (electrochromic display devices contg. display electrode coated with)
ΙT
     429-42-5
     RL: USES (Uses)
        (electrolyte soln. contg., for electrochromic display devices)
TΤ
     67-68-5, uses and miscellaneous
                                       68-12-2, uses and miscellaneous
                                                            109-99-9, uses and
     75-05-8, uses and miscellaneous
                                       107-12-0
                                                109-74-0
     miscellaneous
                    110-71-4
                               140-29-4
                                           37248-85-4
     RL: USES (Uses)
        (electrolyte solvent, for electrochromic display device with polymer
        display electrode)
     70257-87-3P 101150-60-1P
ΙT
                               101150-61-2P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (prepn. of)
IT
     101150-60-1P
     RL: SPN (Synthetic preparation); PREP (Preparation)
        (prepn. of)
RN
     101150-60-1 HCAPLUS
     Poly[(10-methyl-10H-phenothiazine-3,7-diyl)-1,3,4-oxadiazole-2,5-diyl]
CN
     (9CI) (CA INDEX NAME)
```

```
L85 ANSWER 40 OF 46 HCAPLUS COPYRIGHT 2002 ACS
    1985:532463 HCAPLUS
AN
DN
    103:132463
ΤI
    Electrochromic materials
PA
    Mitsubishi Electric Corp., Japan
SO
    Jpn. Kokai Tokkyo Koho, 8 pp.
    CODEN: JKXXAF
DT
    Patent
LA
    Japanese
IC
     ICM C09K009-00
     74-9 (Radiation Chemistry, Photochemistry, and Photographic and Other
     Reprographic Processes)
     Section cross-reference(s): 73, 76
FAN.CNT 1
    PATENT NO.
                                          APPLICATION NO. DATE
                     KIND DATE
ΡI
     JP 60115681
                      A2
                            19850622
                                          JP 1983-225288 19831128
    JP 63065239
                      B4
                            19881215
GI
```

Ι

$$CH_2 = CH$$
 R_1
 R_2
 R_2

Electrochromic polymers contg. structural repeating units from phenothiazine derivs. I (R = H, alkyl, aryl, acyl; R1, R2 = H, halo, alkyl, alkoxy, NO2, CN, NH2, NMe2, CO2H, SO2NH2, SO3H) are claimed. The electrochronic polymers may be a homopolymer of I or copolymer of I with CH2:CR3R4 (R3 = H, alkyl, halo; R4 = H, alkyl, aryl, halo, CO2H, ester group, CONH2, CN). Thus, an electrochronic display cell was prepd. by using poly(3-vinyl-10-methylphenothiazine) and a LiClO4 soln. (in acetonitrile). The display device showed clear redish-purple display at 0.8 V and showed good memory effect. The 3-vinyl-10-methylphenothiazine was prepd. by methylation of phenothiazine by MeI in the presence of a Na metal-dimethyl sulfoxide reaction product, followed by reaction with N-methylformanilide, and subsequent reaction with [Ph3P+Me]Br- in the presence of n-BuLi.

ST electrochronic vinylphenothiazine polymer; phenothiazine vinyl polymer electrochromic; display device electrochromic

IT Electrochromic materials

(vinylphenothiazine deriv. polymers as)

IT Optical imaging devices

(electrochromic, contq. vinylphenothiazine deriv. polymers)

IT 98312-04-0

RL: PRP (Properties)

(electrochromic property of)

IT 35641-59-9

RL: USES (Uses)

(electrochromic \$ubstance contg. methylvinylphenothiazine polymer and)

IT 92-84-2

RL: RCT (Reactant)

(methylation of)

IT 1207-72-3P

RL: SPN (Synthetic preparation); PREP (Preparation)

(prepn. and formation of)

IT 52853-38-0P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)

(prepn. and formylation of)

IT 21520-21-8P 98312-05-1P

RL: RCT (Reactant); PREP (Preparation)

(prepn. and polymn. of)

IT 4997-36-8P 98349-79-2P

RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation)

(prepn. and reaction of)

IT 25657-63-0P 85709-38-2P 98312-06-2P

RL: PREP (Preparation)

(prepn. of, for electrochromic display devices)

IT 25657-63-0P 85709-38-2P 98312-06-2P

RL: PREP (Preparation)

(prepn. of, for electrochromic display devices)

RN 25657-63-0 HCAPLUS

CN 10H-Phenothiazine, 3-ethenyl-10-methyl-, homopolymer (9CI) (CA INDEX NAME)

CM 1

Page 150

CRN 21520-21-8 CMF C15 H13 N S

RN 85709-38-2 HCAPLUS

CN 2-Propenoic acid, 2-methyl-, methyl ester, polymer with 3-ethenyl-10-methyl-10H-phenothiazine (9CI) (CA INDEX NAME)

CM 1

CRN 21520-21-8 CMF C15 H13 N S

CM 2

CRN 80-62-6 CMF C5 H8 O2

$$\begin{array}{ccc} ^{\text{H}_2\text{C}} & \text{O} \\ & \parallel & \parallel \\ \text{Me-C-C-OMe} \end{array}$$

RN 98312-06-2 HCAPLUS

CN 10H-Phenothiazine, 3-bromo-7-ethenyl-10-methyl-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 98312-05-1 CMF C15 H12 Br N S

ANSWER 41 OF 46 HCAPLUS COPYRIGHT 2002 ACS

1981:570018 HCAPLUS

95:170018 DN

Role of solvation and desolvation in polymer "catalysis". III. Influence of high pressures on the alkaline fading reactions of triphenylmethane dyes catalyzed by macroions

Maruno, Tohru; Okubo, Tsuneo; Ise, Norio ΑU

Dep. Polymer Chem., Kyoto Univ., Kyoto, Japan CS

Ber. Bunsenges. Phys. Chem. (1981), 85(8), 667-71 SO CODEN: BBPCAX; ISSN: 0005-9021

DT Journal

LA English

35-6 (Synthetic High Polymers) CC

Section cross-reference(s): 22, 40, 65

Alk. fading reactions of ethyl violet /[2390-59-2] and crystal violet [548-62-9] were studied in the presencé of polyelectrolytes under high pressure, and the significant role of/the hydrophobic hydration effect in polyelectrolyte catalysis was demonstrated in terms of the vol. of activation (.DELTA.V.noteq.). Polyelectrolytes used were poly(4-vinylpyridine) quaternized with n-cetyl bromide (5 mol%) and benzyl chloride (95 mol%) (C16BzPVP), 3-6/ionene polymer (Polybrene [28728-55-4]), Na polyethylenesulfonate [25053-27-4], and Na polystyrenesulfonate (I) [25704-1/8-1]. These four samples, resp., enhanced, slightly retarded, retarded and strongly impeded the reactions. The influence of a cationic surfactant, cetyltrimethylammonium bromide [57-09-0], on the rates and the activation parameters was similar to that of C16BzPVP. These exptl. findings indicate that the hydrophobic macrocations such as C16BzPVP facilitated the formation of the "ice-berg" structure of water around the activated complex so that the complex and the macrocations are incorporated in the ice-berg structure causing a decrease in .DELTA.V.noteq.. On the other hand, the hydrophobic macroanions (e.g. I) enhanced the ice-berg formation around the reactants with resulting increase in .DELTA.V.noteq.. Correspondingly, addn. of C16BzPVP decreased the enthalpy and entropy of activation (.DELTA.H.noteq. and .DELTA.S.noteq.) whereas I increased both .DELTA.H.noteq. and .DELTA.S.noteq.. These results can be reasonably accepted if we assume that the contribution of the electrostatic hydration term is much less important than that from the hydrophobic hydration.

polyelectrolyte catalysis /dye fading; triphenylmethane dye alk fading; pressure alk fading dye; hydration alk fading dye; mechanism polyelectrolyte catalysis dye fading; hydroxylation triphenylmethane dye catalysis; solvation alk fading dye; macroion catalysis dye fading

Ionene polymers IT RL: USES (Uses)

(alk. fading in presence of, of triphenylmethane dyes under high pressure, kinetics/and mechanism of)

IT Electrolytes

(anionic, alk. fading in presence of, of triphenylmethane dyes under high pressure, kinetics and mechanism of)

Page 152

IT Activation volume

(of alk. fading of triphenylmethane dyes, effect of polyelectrolytes on)

IT Thermodynamics

(of alk. fading of triphenylmethane dyes, in presence of polyelectrolytes under high pressure, hydration in relation to)

IT Hydration, chemical

(of reactants and activated complexes, in alk. fading of triphenylmethane dyes, effect of polyelectrolytes on)

IT Kinetics of hydroxylation

(of triphenylmethane dyes in alk. soln., in presence of polyelectrolytes and high pressure)

IT Hydroxylation

(of triphenylmethane dyes, in alk. soln., kinetics and mechanism of polyelectrolyte-catalyzed)

IT Dyes

(triphenylmethane, alk. fading of, in presence of polyelectrolytes under high pressure, kinetics and mechanism of)

IT Fading

(alk., of triphenylmethane dyes in presence of polyelectrolytes and high pressure, kinetics and mechanism of)

IT Polyelectrolytes

(cationic, alk. fading in presence of, of triphenylmethane dyes under high pressure, kinetics and mechanism of)

TT 57-09-0 100-44-7D, quaternization product with cetyl bromide of poly(4-vinylpyridine) 112-82-3D, quaternization product with benzyl chloride of poly(4-vinylpyridine) 25053-27-4 25232-41-1D, quaternized with benzyl chloride and cetyl bromide 25704-18-1 28728-55-4 RL: USES (Uses)

(alk. fading in presence of, of triphenylmethane dyes under high pressure, kinetics and mechanism of)

IT 548-62-9 **2390-59-2**

RL: RCT (Reactant)

(alk. fading of, in presence of polyelectrolytes under high pressure, kinetics and mechanism of)

IT 14280-30-9, properties

RL: RCT (Reactant)

(reaction of, with triphenylmethane dyes under high pressure, fading in, kinetics and mechanism of polyelectrolyte-catalyzed)

IT 28728-55-4

RL: USES (Uses)

(alk. fading in presence of, of triphenylmethane dyes under high pressure, kinetics and mechanism of)

RN 28728-55-4 HCAPLUS

CN Poly[(dimethyliminio)-1,3-propanediyl(dimethyliminio)-1,6-hexanediyl
dibromide] (9CI) (CA INDEX NAME)

Page 153

ΙT 2390-59-2

RL: RCT (Reactant)

(alk. fading of, in presence of polyelectrolytes under high pressure, kinetics and mechanism of)

RN 2390-59-2 HCAPLUS

Ethanaminium, N-[4-[bis[4-(diethylamino)phenyl]methylene]-2,5-CN cyclohexadien-1-ylidene]-N-ethyl-, chloride (9CI) (CA INDEX NAME)

Cl

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L85 ANSWER 42 OF 46 HCAPLUS COPYRIGHT 2002 ACS
AN
    1980:164689 HCAPLUS
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DN 92:164689

ΤI Photosensitive polymers

ΙN Yamaguchi, Hiroyoshi; Iwaki, Akio; Kita, Moriyasu; Sasazawa, Tatsuya

Konishiroku Photo Industry Co., Ltd., Japan PA

Brit. UK Pat. Appl., 14 pp. SO

CODEN: BAXXDU

DT Patent

LA English

IC C08F008-30; G03C001-71

36-3 (Plastics Manufacture and Processing) Section cross-reference(s): 74, 76

FAN.CNT 1

T 7 7 L 4 .	0111 1		/	/
	PATENT NO.	KIND	DATE	APPLICATION NO. / DATE
ΡI	GB 2018779	Α	1 9791024	GB 1979-12930/ 19790412
	GB 2018779	B2	19820922	\ /
	JP 54135525	A2 /	19791020	JP 1978-42940 19780412
	DE 2915154	A1	19791025	DE 1979-2915154 19790412
	US 4442196	K	19840410	US 1980-207087 19801114
PRAI	JP 1978-42940		19780412	
	US 1979-29350		19790411	

Photocurable polymeric esters [CH2CRR1]n [R = ZZ1O2CC(CN):CHCH:CHC6H4N3-p; AΒ R1 = H, halogen, or alkyl; Z = divalent org. group; Z1 = optionally substituted phenylene or naphthylene], useful in the prodn. of printing plates and printed circuits, are manufd. by/treating a hydroxy functional polymers with p-azidoci/namylidene-.alpha.-cyanoacetic chloride (I) in the presence of a base. Thus, 20.4 g poly(p-hydroxystyrene) [24979-70-2] in 200 mL dry pyridine and 140 mL Me2CO at 50.degree. was treated by portionwise Af 9.7 g I. The mixt. was maintained 5 h at 50.degree. before

```
pouring into 2 L iced H2O contg. 60 mL concd. HCl to ppt. the esterified
     polymer (II) [73361-56-5] contg. 25% I-esterified OH groups. II (10 g)
     was dissolved in 200 mL Et cellosolve and applied to a sand-blasted Al
     plate by a rotary applicator and dried. The coated plate gave a clear
     colored image when exposed 3 min 1 m from a 3 kW Hg lamp, with the
     photosensitivity of the coated plate being better than similar plates
     coated with poly(vinyl cinnamate), poly(vinyl .alpha.-cyanocinnamate), or
     poly(vinyl p-azidobenzoate).
ST
     printing plate photoresist polymer; photocuring azido
     polymer ester; hydroxy polymer azidocinnamylidenecyanoacetic ester;
     azidocinnamylidenecyanoacetic polymer ester photocuring; resist
     photocuring azido polyester; elec circuit photoresist polymer
IT
     Printing plates
        (photocurable polymeric azidocinnamylidenecyanoacetic esters for)
TΤ
     Resists
        (photo-, polymeric azidocinnamylidenecyanoacetic esters for)
     Coating materials
ΙT
        (photocurable, polymeric azidocinnamylidenecyanoacetic esters for)
ΤT
     Electric circuits
        (printed, photocurable polymeric azidocinnamylidenecyanoacetic esters
        for)
TT
     920-46-7
     RL: RCT (Reactant)
        (amidation by, of aminonaphthol)
TΨ
     760-93-0
     RL: RCT (Reactant)
        (amidation by, of hydroxyaniline)
ΙT
     591-27-5
     RL: RCT (Reactant)
        (amidation of, by methacrylic anhydride)
     83-55-6
     RL: RCT (Reactant)
        (amidation of, by methacryloyl chloride)
                   24979-74-6P
                                56592-53-1P
TΤ
     24979-70-2P
                                              57167-08-5P
                                                              73310-43-7P
     73310-44-8P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP (Preparation)
        (manuf. and esterification of, by azidocinnamylidenecyanoacetic acid
        chloride)
TΤ
     73361-52-1P 73361-53-2P 73361-54-3P
     73361-55-4P 73361-56-5P 73361-57-6P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP
     (Preparation)
        (manuf. and photochem. crosslinking of)
     27931-11-9P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (manuf. and polymn. of)
TT.
     14473-49-5P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (manuf. and polymn. of, with Me methacrylate)
     73361-52-1P 73361-53-2P 73361-54-3P
     73361-55-4P 73361-56-5P 73361-57-6P
     RL: IMF (Industrial manufacture); RCT (Reactant); PREP
     (Preparation)
        (manuf. and photochem. crosslinking of)
RN
     73361-52-1 HCAPLUS
CN
     2-Propenoic acid, 2-methyl-, methyl ester, polymer with
     N-(3-hydroxyphenyl)-2-methyl-2-propenamide, 5-(4-azidophenyl)-2-cyano-2,4-
     pentadienoate (ester) (9CI) (CA INDEX NAME)
     CM
```

Page 155

CRN 42460-58-2 CMF C12 H8 N4 O2

$$CH = CH - CH = C - CO_2H$$

CM 2

CRN 56592-53-1

CMF (C10 H11 N O2 . C5 H8 O2) \times

CCI PMS

CM 3

CRN 14473-49-5 CMF C10 H11 N O2

CM 4

CRN 80-62-6 CMF C5 H8 O2

RN 73361-53-2 HCAPLUS

CN 2-Propenamide, N-(4-hydroxyphenyl)-2-methyl-, homopolymer, 5-(4-azidophenyl)-2-cyano-2,4-pentadienoate (ester) (9CI) (CA INDEX NAME)

CM 1

CRN 42460-58-2 CMF C12 H8 N4 O2

Page 156

$$CH = CH - CH = C - CO_2H$$

CM 2

CRN 57167-08-5

CMF (C10 H11 N O2)x

CCI PMS

CM 3

CRN 19243-95-9 CMF C10 H11 N O2

RN 73361-54-3 HCAPLUS

CN 2-Propenamide, N-(4-hydroxyphenyl)-2-methyl-, polymer with ethenylbenzene, 5-(4-azidophenyl)-2-cyano-2,4-pentadienoate (ester) (9CI) (CA INDEX NAME)

CM 1

CRN 42460-58-2 CMF C12 H8 N4 O2

$$CH = CH - CH = C - CO_2H$$

CM 2

CRN 73310-44-8

CMF (C10 H11 N O2 . C8 H8)x

CCI PMS

CM 3

CRN 19243-95-9

Page 157

CMF C10 H11 N O2

CM 4

CRN 100-42-5 CMF C8 H8

 ${\tt H_2C} = {\tt CH-Ph}$

RN 73361-55-4 HCAPLUS

CN 2-Propenamide, N-(5-hydroxy-1-naphthalenyl)-2-methyl-, homopolymer, 5-(4-azidophenyl)-2-cyano-2,4-pentadienoate (ester) (9CI) (CA INDEX NAME)

CM 1

CRN 42460-58-2 CMF C12 H8 N4 O2

$$CH = CH - CH = C - CO_2H$$

CM 2

CRN 73310-43-7

CMF (C14 H13 N O2)x

CCI PMS

CM 3

CRN 27931-11-9 CMF C14 H13 N O2

Page 158

RN 73361-56-5 HCAPLUS

CN Phenol, 4-ethenyl-, homopolymer, 5-(4-azidophenyl)-2-cyano-2,4-pentadienoate (9CI) (CA INDEX NAME)

CM 1

CRN 42460-58-2 CMF C12 H8 N4 O2

$$CH = CH - CH = C - CO_2H$$

CM 2

CRN 24979-70-2 CMF (C8 H8 O)x CCI PMS

CM 3

CRN 2628-17-3 CMF C8 H8 O

RN 73361-57-6 HCAPLUS

CN Phenol, 4-ethenyl-, polymer with ethenylbenzene, 5-(4-azidophenyl)-2-cyano-2,4-pentadienoate (9CI) (CA INDEX NAME)

CM 1

CRN 42460-58-2 CMF C12 H8 N4 O2

Page 159

$$CH = CH - CH = C - CO_2H$$

CM 2

CRN 24979-74-6

CMF (C8 H8 O . C8 H8) \times

CCI PMS

CM 3

CRN 2628-17-3 CMF C8 H8 O

CM 4

CRN 100-42-5 CMF C8 H8

 $H_2C = CH - Ph$

L85 ANSWER 43 OF 46 HCAPLUS COPYRIGHT 2002 ACS

AN 1980:119694 HCAPLUS

DN 92:119694

TI Photoconducting polymer material and its use

IN Watarai, Syu; Sawada, Kenichi; Saida, Takashi

PA Fuji Photo Film Co., Ltd., Japan

SO Ger. Offen., 23 pp.

CODEN: GWXXBX

DT Patent

LA German

IC C08G075-00; G03C005-04

CC 74-3 (Radiation Chemistry, Photochemistry, and Photographic Processes)

FAN.CNT 1

1777.	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
ΡI	DE 2916383	A1	19791108	DE 1979-2916383	19790423
	JP 54141896	A2	19791105	JP 1978-49459	19780426
	GB 2019860	A	19791107	GB 1979-14409	19790425
	GB 2019860	B2	19820428		
	US 4229510	A	19801021	US 1979-33630	19790426

Page 160

PRAI JP 1978-49459

19780426

Polymers superior in photocond. to-poly(vinylcarbazole), softening at 30-100.degree., and usable as 0.05-20.mu. particles in 0.5-20% dispersions in an insulating binder in the form of 2-50.mu. electrophotog. layers of high mech. strength or as photoelectrophoretic pigments, are obtained by treating an N-C1-6 alkylphenothiazine with an excess of HCHO in an org. solvent at 40-100.degree. and at a reactant concn. of 5-50% with an acid as catalyst. The polymers contain 2-20 recurring phenothiazine-CH2-units. Thus, N-ethylphenothiazine 20.2, (CH2O)n 3, and H2SO4 0.5 g were stirred in dioxane 150 mL at 90.degree for 5 h. The reaction product was pptd. in EtOH and repptd. by MeOH from its THF soln. An Al plate was coated with a 10.mu. film of polymer of N-butylphenothiazine prepd. in the above manner (applied as 10% PhMe soln.), charged to +270 V by a 6 kV corona discharge, exposed through a pos. transparency with a 100-W W lamp at 30 cm for 0.5 s, and the latent image developed to a pos. copy by a polystyrene-C toner with 0.5 mm glass beads as carrier.

ST photoconductor polymer electrophotog

IT Photoconductors

Photography, electro-, photoconductors

(alkylphenothiazine-førmaldehyde polymers as)

IT 73025-91-9 73025-92-0 / 73025-94-2

RL: USES (Uses)

(electrophotog. photoconductor, properties of)

IT 73025-95-3P 73025-96-4/P

RL: SPN (Synthetic preparation); PREP (Preparation)

(prepn. of)

IT 73025-95-3P 73025-96-4P

RL: SPN (Synthetic preparation); PREP (Preparation)

(prepn. of)

RN 73025-95-3 HCAPLUS

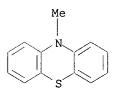
CN Formaldehyde, polymer with 10-methyl-10H-phenothiazine (9CI) (CA INDEX

NAME)

CM 1

CRN 1207-72-3

CMF C13 H11 N S



CM 2

CRN 50-00-0 CMF C H2 O

H2C==O

RN 73025-96-4 HCAPLUS

CN Formaldehyde, polymer with 10-propyl-10H-phenothiazine (9CI) (CA INDEX NAME)

```
LEE 09/828075 Page 161
CM 1
```

15375-48-1

C15 H15 N S

CRN

CMF

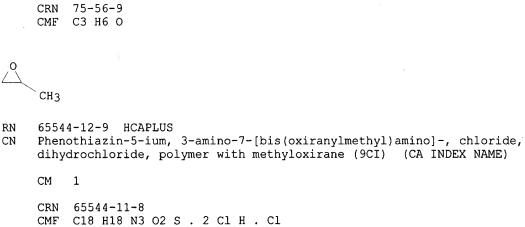
CM 2

CRN 50-00-0 CMF C H2 O

 $H_2C = 0$

```
T.85
   ANSWER 44 OF 46 HCAPLUS COPYRIGHT 2002 ACS
     1979:475084 HCAPLUS
AN
DN
     91:75084
     Glycidyl group-containing dye polymers
ΤI
     Shigehara, Kiyotaka; Tsuchida, Eishun
ΤN
                                                                   (Suppositions)
PΑ
     Japan
SO
     Jpn. Kokai Tokkyo Koho, 13 pp.
     CODEN: JKXXAF
DT
     Patent
     Japanese
LΑ
IC
     C08G065-08
     35-3 (Synthetic High Polymers)
CC
FAN.CNT 1
                                           APPLICATION NO.
                                                            DATE
     PATENT NO.
                     KIND
                           DATE
                                           _____
                      ____
                                                            19780911
                            19790417
                                           JP 1978-111526
                       A2
PΙ
     JP 54048897
                            19850228
     JP 60008010
                       B4
     Coloring materials having glycidyl groups are polymd. to give polymers
AB
     having coloring groups. Thus, a mixt. of 0.392 g 7-glycidylamino-3-imino-
     3H-phenothiazine-HCl, 100 mL Me2SO, and 1 mL of 10% BF3 in Et2O, was
     stirred in a sealed tube at 60.degree. for 6 h to give 0.102 g polymer
     [65544-58-3] having reduced viscosity 0.12 d L/g (30.degree., 0.1 g/17 mL
     Me2SO).
     glycidyl group dye polymer; glycidylthionine polymer
ST
     65544-10-7P 65544-12-9P 65544-14-1P
ΙT
                  65544-17-4P
                                 65544-18-5P 65544-20-9P
     65544-15-2P
                               65544-25-4P 65544-51-6P
     65544-22-1P 65544-23-2P
     65544-53-8P 65544-54-9P 65544-56-1P
     65544-57-2P 65544-58-3P
                               65587-55-5P
     71092-19-8P 71092-20-1P
     RL: PREP (Preparation)
        (prepn. of colored)
                                    482-89-3 548-62-9
                        135-59-1
IT
     61-73-4 81-88-9
     573-58-0
```

```
LEE 09/828075
                      Page 162
     RL: RCT (Reactant)
        (reaction of, with epichlorohydrin)
IT
     106-89-8, reactions
     RL: RCT (Reactant)
        (reaction of, with iminophenothiazinamine derivs.)
IT
     65544-10-7P 65544-12-9P 65544-14-1P
     65544-15-2P 65544-20-9P 65544-22-1P
     65544-23-2P 65544-53-8P 65544-54-9P
     65544-56-1P 65544-57-2P 65544-58-3P
     71092-19-8P 71092-20-1P
     RL: PREP (Preparation)
        (prepn. of colored)
RN
     65544-10-7 HCAPLUS
     Phenothiazin-5-ium, 3-amino-7-[(oxiranylmethyl)amino]-, chloride,
CN
     dihydrochloride, polymer with methyloxirane (9CI) (CA INDEX NAME)
     CM
          1
     CRN 65544-09-4
     CMF C15 H14 N3 O S . 2 Cl H . Cl
                       ин-сн2
                 Cl-
              ●2 HC1
     CM
          2
     CRN
          75-56-9
         СЗ Н6 О
     CMF
```



Page 163

$$H_2N$$
 S_+
 N
 CH_2
 CH_2

● cl-

●2 HCl

CM 2

CRN 75-56-9 CMF C3 H6 O

RN 65544-14-1 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[dimethyl(oxiranylmethyl)ammonio], dichloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-13-0

CMF C19 H23 N3 O S . 2 Cl

$$\begin{array}{c|c} N & \text{Me} \\ \hline \\ Me_2N & N \\ \hline \\ Me \end{array}$$

●2 C1-

RN 65544-15-2 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[dimethyl(oxiranylmethyl)ammonio]-, dichloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

Page 164

CRN 65544-13-0

CMF C19 H23 N3 O S . 2 Cl

$$\begin{array}{c|c} & & & \text{Me} \\ & & & \\ & & & \\ \text{Me} & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & &$$

●2 C1-

CM 2

CRN 75-56-9

CMF C3 H6 O



RN 65544-20-9 HCAPLUS

CN Oxiranemethanaminium, N-[4-[[4-(dimethylamino)phenyl][4-(dimethyliminio)-2,5-cyclohexadien-1-ylidene]methyl]phenyl]-N,N-dimethyl-, dichloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

CRN 65544-18-5

CMF C28 H35 N3 O . 2 C1

●2 C1-

CM 2

CRN 75-56-9

KATHLEEN FULLER EIC 1700/LAW LIBRARY 308-4290

Page 165

CMF C3 H6 O

RN 65544-22-1 HCAPLUS

CN Oxiranemethanaminium, N-[9-(2-carboxyphenyl)-6-(diethylamino)xanthylium-3-yl]-N,N-diethyl-, dichloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-21-0

CMF C31 H36 N2 O4 . 2 C1

$$\begin{array}{c|c} & & & & \\ & & & & \\ & & & & \\ Et_2N & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & \\ & \\ & \\ & \\ & \\ \end{array}$$

●2 C1_

RN 65544-23-2 HCAPLUS

CN Oxiranemethanaminium, N-[9-(2-carboxyphenyl)-6-(diethylamino)xanthylium-3-yl]-N,N-diethyl-, dichloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

CRN 65544-21-0

CMF C31 H36 N2 O4 . 2 Cl

$$\begin{array}{c|c} & & & \\ & \text{HO}_2\text{C} \\ & & \\ \text{Et}_2\text{N} \\ & & \\ \text{O}_+ \\ & & \\ \text{Et} \\ \end{array}$$

Page 166

CM 2

CRN 75-56-9 CMF C3 H6 O

СНЗ

RN 65544-53-8 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-(oxiranylmethoxy)propyl]methylamino]-, chloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-52-7

CMF C21 H26 N3 O3 S . Cl

● Cl-

RN 65544-54-9 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-(oxiranylmethoxy)propyl]methylamino]-, chloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM :

CRN 65544-52-7

CMF C21 H26 N3 O3 S . C1

● c1-

CM 2

Page 167

CRN 75-56-9 CMF C3 H6 O

СНЗ

RN 65544-56-1 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-[2-(oxiranylmethoxy)ethoxy]propyl]methylamino]-, chloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-55-0

CMF C23 H30 N3 O4 S . Cl

● Cl-

RN 65544-57-2 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-[2-(oxiranylmethoxy)ethoxy]propyl]methylamino]-, chloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

CRN 65544-55-0

CMF C23 H30 N3 O4 S . Cl

• c1-

CM 2

CRN 75-56-9 CMF C3 H6 O

Page 168

RN 65544-58-3 HCAPLUS

CN Phenothiazin-5-ium, 3-amino-7-[(oxiranylmethyl)amino]-, chloride, dihydrochloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-09-4

CMF C15 H14 N3 O S . 2 Cl H . Cl

● Cl-

●2 HC1

RN 71092-19-8 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-[4-(oxiranylmethoxy)-1,4-dioxobutoxy]propyl]methylamino]-, chloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65620-28-2

CMF C25 H30 N3 O6 S . C1

PAGE 1-A

Page 169

PAGE 1-B

0

RN 71092-20-1 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-[4-(oxiranylmethoxy)-1,4-dioxobutoxy]propyl]methylamino]-, chloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

CRN 65620-28-2

CMF C25 H30 N3 O6 S . C1

PAGE 1-A

• c1-

PAGE 1-B

0

CM 2

CRN 75-56-9 CMF C3 H6 O

СНЗ

IT 61-73-4 548-62-9

RL: RCT (Reactant)

(reaction of, with epichlorohydrin)

RN 61-73-4 HCAPLUS

CN Phenothiazin-5-ium, 3,7-bis(dimethylamino)-, chloride (9CI) (CA INDEX NAME)

Page 170

• c1-

RN 548-62-9 HCAPLUS

CN Methanaminium, N-[4-[bis[4-(dimethylamino)phenyl]methylene]-2,5-cyclohexadien-1-ylidene]-N-methyl-, chloride (9CI) (CA INDEX NAME)

● Cl-

GI

```
L85 ANSWER 45 OF 46 HCAPLUS COPYRIGHT 2002 ACS
AN
     1978:426022 HCAPLUS
DN
    89:26022
ΤI
    Glycidyl group-containing monomeric and polymeric dyes
IN
     Shigehara, Kiyotaka; Tsuchida, Hidetoshi
PA
     Japan
     Japan. Kokai, 15 pp.
SO
     CODEN: JKXXAF
DT
     Patent
LA
     Japanese
IC
     C09B057-00
CC
     40-6 (Dyes, Fluorescent Whitening Agents, and Photosensitizers)
FAN.CNT 1
     PATENT NO.
                     KIND
                                          APPLICATION NO.
                                                           DATE
                           DATE
                                           _____
PΙ
                      A2
                            19771012
                                          JP 1976-36986
                                                           19760403
     JP 52121038
     JP 60018701
                      В4
                           19850511
```

```
AΒ
     Amino group-contg. dyes were treated with epichlorohydrin (I) [106-89-8]
     or other glycidyl compds., and the resulting glycidyl group-contg. dyes
     were homopolymd. or copolymd. with propylene oxide. For example, I and
     thionine (II) [581-64-6] in DMF were heated at 40.degree. for 5 h in the
     dark and treated with HCl to give 47.3% violet black III [65544-09-4]
     which was homopolymd. in the presence of BF3.Et20 to give polymer with
     better lightfastness than II.
ST
     glycidyl dye polymer
IT
    Epoxy group
        (dyes contg.)
IT
     Quaternary ammonium compounds, uses and miscellaneous
     RL: MSC (Miscellaneous)
        (dyes, mono- and polymeric)
TΤ
     Crosslinking agents
        (for polymeric dyes)
IT
     Polymerization
        (of glycidyl group-contg. dyes, in the presence of boron trifluoride
        etherate)
IT
     Dyes
        (mono- and polymeric, glycidyl derivs.)
IT
     108-77-0
               111-50-2
                          629-03-8
                                      7710-20-5
                                                  36182-48-6
     RL: USES (Uses)
        (crosslinking agents for reaction products from polyethylenimine and
        glycidy group-contg. dyes)
IT
     65544-09-4P
                   65544-11-8P
                                 65544-13-0P
                                                65544-16-3P
                                                              65544-18-5P
     65544-21-0P
                   65544-24-3P
                                 65544.-52-7P
                                                65544-55-0P
                                                              65620-28-2P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (manuf. and polymn. of)
IT
     9002-98-6DP, reaction products with glycidyl group-contg. dyes
     65544-10-7P 65544-12-9P 65544-14-1P
     65544-15-2P
                   65544-17-4P 65544-19-6P
     65544-20-9P 65544-22-1P 65544-23-2P
     65544-25-4P
                   65544-51-6P 65544-53-8P 65544-54-9P
     65544-56-1P 65544-57-2P 65544-58-3P
     65587-55-5P 65684-17-5P 65684-18-6P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (prepn. of)
IT
     2224-15-9
     RL: RCT (Reactant)
        (reaction of, with Azure B)
                21739-14-0
TT
     2238-07-5
     RL: RCT (Reactant)
        (reaction of, with Azure B)
IT
     106-89-8, reactions
     RL: RCT (Reactant)
        (reaction of, with dyes)
IT
              81-88-9
                       482-89-3 548-62-9
     61-73-4
     581-64-6
     RL: RCT (Reactant)
```

```
LEE 09/828075
                      Page 172
        (reaction of, with epichlorohydrin)
     531-55-5
IT
     RL: RCT (Reactant)
        (reaction of, with glycidyl compds.)
     65544-10-7P 65544-12-9P 65544-14-1P
     65544-15-2P 65544-19-6P 65544-20-9P
     65544-22-1P 65544-23-2P 65544-53-8P
     65544-54-9P 65544-56-1P 65544-57-2P
     65544-58-3P 65684-17-5P 65684-18-6P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (prepn. of)
RN
     65544-10-7 HCAPLUS
CN
     Phenothiazin-5-ium, 3-amino-7-[(oxiranylmethyl)amino]-, chloride,
     dihydrochloride, polymer with methyloxirane (9CI) (CA INDEX NAME)
     CM
     CRN 65544-09-4
     CMF C15 H14 N3 O S . 2 Cl H . Cl
                      ин-сн2
               ● Cl -
             ●2 HCl
     CM
          2
     CRN 75-56-9
     CMF C3 H6 O
     CH3
RN
     65544-12-9 HCAPLUS
     Phenothiazin-5-ium, 3-amino-7-[bis(oxiranylmethyl)amino]-, chloride,
CN
     dihydrochloride, polymer with methyloxirane (9CI) (CA INDEX NAME)
     CM
```

CMF C18 H18 N3 O2 S . 2 Cl H . Cl

CRN 65544-11-8

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$$H_2N$$
 S_+
 CH_2
 CH_2

● C1⁻

●2 HC1

CM 2

CRN 75-56-9 CMF C3 H6 O

СНЗ

RN 65544-14-1 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[dimethyl(oxiranylmethyl)ammonio]-, dichloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-13-0

CMF C19 H23 N3 O S . 2 Cl

$$\begin{array}{c|c} & \text{Me} & \text{Me} \\ & \text{N} & \text{CH}_2 \\ & \text{Me} \end{array}$$

●2 Cl⁻

RN 65544-15-2 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[dimethyl(oxiranylmethyl)ammonio]-, dichloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

Page 174

CRN 65544-13-0 CMF C19 H23 N3 O S . 2 Cl

$$\begin{array}{c|c} & & & \text{Me} \\ & & & \\ \text{Me} & & \\ \text{Me} & & \\ \end{array}$$

●2 Cl⁻

CM 2

CRN 75-56-9 CMF C3 H6 O



RN 65544-19-6 HCAPLUS

CN Oxiranemethanaminium, N-[4-[[4-(dimethylamino)phenyl]][4-(dimethyliminio)-2,5-cyclohexadien-1-ylidene]methyl]phenyl]-N,N-dimethyl-, dichloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-18-5 CMF C28 H35 N3 O . 2 Cl

●2 C1-

RN 65544-20-9 HCAPLUS

CN Oxiranemethanaminium, N-[4-[[4-(dimethylamino)phenyl][4-(dimethyliminio)-2,5-cyclohexadien-1-ylidene]methyl]phenyl]-N,N-dimethyl-, dichloride,

Page 175

polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

CRN 65544-18-5

CMF C28 H35 N3 O . 2 Cl

●2 C1-

CM 2

CRN 75-56-9

CMF C3 H6 O



RN 65544-22-1 HCAPLUS

CN Oxiranemethanaminium, N-[9-(2-carboxyphenyl)-6-(diethylamino)xanthylium-3-yl]-N,N-diethyl-, dichloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-21-0

CMF C31 H36 N2 O4 . 2 Cl

Page 176

$$\begin{array}{c|c} & & & \\ & \text{HO}_2\text{C} \\ & & & \\ \text{Et}_2\text{N} \\ & & & \\ & & & \\ \text{Et} \\ & & \\ \end{array}$$

●2 C1-

RN 65544-23-2 HCAPLUS

CN Oxiranemethanaminium, N-[9-(2-carboxyphenyl)-6-(diethylamino)xanthylium-3-yl]-N,N-diethyl-, dichloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

CRN 65544-21-0 CMF C31 H36 N2 O4 . 2 C1

$$\begin{array}{c|c} & & & & \\ & \text{HO}_2\text{C} & & & \\ & & & \text{Et} & \\ & & & \text{CH}_2 & \\ & & & \text{Et} & \\ \end{array}$$

●2 Cl⁻

CM 2

CRN 75-56-9 CMF C3 H6 O

RN 65544-53-8 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-(oxiranylmethoxy)propyl]methylamino]-, chloride, homopolymer (9CI) (CF INDEX NAME)

Page 177

CM 1

CRN 65544-52-7

CMF C21 H26 N3 O3 S . C1

● cl-

RN 65544-54-9 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-(oxiranylmethoxy)propyl]methylamino]-, chloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

CRN 65544-52-7

CMF C21 H26 N3 O3 S . Cl

• c1-

CM 2

CRN 75-56-9 CMF C3 H6 O

СН3

RN 65544-56-1 HCAPLUS

CM 1

CRN 65544-55-0

Page 178

CMF C23 H30 N3 O4 S . C1

● C1-

RN 65544-57-2 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-[2-(oxiranylmethoxy]propyl]methylamino]-, chloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM 1

CRN 65544-55-0

CMF C23 H30 N3 O4 S . Cl

• c1-

CM 2

CRN 75-56-9 CMF C3 H6 O

СН3

CN

RN 65544-58-3 HCAPLUS

Phenothiazin-5-ium, 3-amino-7-[(oxiranylmethyl)amino]-, chloride, dihydrochloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-09-4

CMF C15 H14 N3 O S . 2 Cl H . Cl

Page 179

c1 =

●2 HCl

RN 65684-17-5 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-[[6-(oxiranylmethoxy)-1,6-dioxohexyl]oxy]propyl]methylamino]-, chloride, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65684-16-4 CMF C27 H34 N3 O6 S . Cl

PAGE 1-A

• c1-

PAGE 1-B



RN 65684-18-6 HCAPLUS

CN Phenothiazin-5-ium, 3-(dimethylamino)-7-[[2-hydroxy-3-[[6-(oxiranylmethoxy)-1,6-dioxohexyl]oxy]propyl]methylamino]-, chloride, polymer with methyloxirane (9CI) (CA INDEX NAME)

CM :

CRN 65684-16-4 CMF C27 H34 N3 O6 S . Cl

Page 180

PAGE 1-A

● Cl-

PAGE 1-B

CM 2

CRN 75-56-9 CMF C3 H6 O

СНЗ

IT 61-73-4 548-62-9

RL: RCT (Reactant)

(reaction of, with epichlorohydrin)

RN 61-73-4 HCAPLUS

CN Phenothiazin-5-ium, 3,7-bis(dimethylamino)-, chloride (9CI) (CA INDEX NAME)

Me₂N NMe₂

● C1-

RN 548-62-9 HCAPLUS

Methanaminium, N-[4-[bis[4-(dimethylamino)phenyl]methylene]-2,5cyclohexadien-1-ylidene]-N-methyl-, chloride (9CI) (CA INDEX NAME)

L85 ANSWER 46 OF 46 HCAPLUS COPYRIGHT 2002 ACS AN 1978:91046 HCAPLUS 88:91046 DN ΤI Active halogen-containing dye derivatives IN Shiqehara, Kiyotaka; Tsuchida, Hidetoshi PΑ Japan Japan. Kokai, 14 pp. SO CODEN: JKXXAF DT Patent LA Japanese IC C09B057-00 CC 40-6 (Dyes, Fluorescent Whitening Agents, and Photosensitizers) FAN.CNT 1 PATENT NO. KIND DATE APPLICATION NO. DATE ΡI JP 52121037 A2 19771012 JP 1976-36985 19760402 JP 55033776 В4 19800902 GI Cl HBr Br (CH2) 6NH II AB

Amino group-contg. dyes were treated with 1,6-dibromohexane (I) [629-03-8] or adipoyl chloride [111-50-2], and the resulting compds. contg. active halogen were treated with amino-group contg. polymers to give polymeric dyes For example, thionine [581-64-6] in DMF was treated with I to give 19.6% violet black II [65544-48-1] which was treated with poly(4-vinylpyridine) to give bluish violet polymeric dye [65544-50-5].

ST 'active halogen dye polymer; polymeric dye

IT Crosslinking agents

(dibromohexane, for polymeric dyes)

Ouaternary ammohium compounds, uses and miscellaneous RL: MSC (Miscellaneous) (dyes)

IT Dyes

Page 182

```
(mono- and polymeric, active halogen-contg.)
     9002-98-6DP, reaction products with active halogen-contg. dyes
TΨ
     26809-43-8DP, hydrolyzed, reaction products with active halogen-contg.
                                     65544-48-1DP, reaction products
            65544-45-8P 65544-47-0P
     with polyethylenimine and hydrolyzed poly(vinylphthalimide)
     65544-50-5P 65561-98-0DP, reaction products with
     polyethylenimine
                       65561-98-0P
                                     65561-99-1P
                                                   65562-00-7DP, reaction
     products with polyethylenimine and hydrolyzed poly(vinylphthalimide)
     65562-00-7P 65562-01-8DP, reaction products with polyethylenimine and
     hydrolyzed poly(vinylphthalimide)
                                                     65562-02-9DP, reaction
                                       65562-01-8P
     products with polyethylenimine and hydrolyzed poly(vinylphthalimide)
                 65562-03-0DP, reaction products with polyethylenimine and
     65562-02-9P
                                         65562-03-0P
                                                     65562-04-1DP, reaction
     hydrolyzed poly(vinylphthalimide)
     products with polyethylenimine and hydrolyzed poly(vinylphthalimide)
     65562-04-1P 65562-05-2DP, reaction products with polyethylenimine and
     hydrolyzed poly(vinylphthalimide)
                                        65562-05-2P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (prepn. of)
IT
     111-50-2
     RL: RCT (Reactant)
        (reaction of, with Azure B and thionine)
IT
     RL: RCT (Reactant)
        (reaction of, with adipoyl chloride)
ΙT
     61-73-4
               92-62-6
     RL: RCT (Reactant)
        (reaction of, with dibromohexane)
     50-71-5 581-64-6
TΥ
     RL: RCT (Reactant)
        (reaction of, with dibromohexane and adipoyl chloride)
ΤT
     629-03-8
     RL: RCT (Reactant)
        (reaction of, with thionine and methylene blue and alloxan)
ΙT
     65544-47-0P 65544-50-5P
     RL: IMF (Industrial manufacture); PREP (Preparation)
        (prepn. of)
     65544-47-0 HCAPLUS
RN
     Phenothiazin-5-ium, 3-[(6-bromohexyl)dimethylammonio]-7-(dimethylamino)-,
CN
     bromide chloride, compd. with 4-ethenylpyridine homopolymer (9CI)
     INDEX NAME)
     CM
          1
     CRN 65544-45-8
     CMF C22 H30 Br N3 S . Br . Cl
```

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$$Me_{2N}$$
 S_{+}
 Me_{N}^{+}
 $CH_{2})_{6}-Br$
 Me_{Me}

• Br-

● c1-

CM 2

CRN 25232-41-1 CMF (C7 H7 N)x

CCI PMS

CM 3

CRN 100-43-6 CMF C7 H7 N

RN 65544-50-5 HCAPLUS

Phenothiazin-5-ium, 3-amino-7-[(6-bromohexyl)amino]-, chloride, monohydrobromide, compd. with 4-ethenylpyridine homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 65544-48-1

CMF C18 H21 Br N3 S . Br H . Cl

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• HBr

● cl-

CM 2

CRN 25232-41-1 CMF (C7 H7 N)x CCI PMS

> CM 3

> > CRN 100-43-6 CMF C7 H7 N

ΙT 61-73-4

RL: RCT (Reactant)

(reaction of, with dibromohexane) 61-73-4 HCAPLUS

RN

Phenothiazin-5-ium, 3,7-bis(dimethylamino)-, chloride (9CI) (CA INDEX CNNAME)

● C1-